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Response surface methodology for optimization of turmeric essential oil-loaded nanoemulgel

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ABSTRACT: Turmeric essential oil has antioxidant and anti-inflammatory activities in a topical dosage form. Nanoemulgels are one of them. Using emulsifiers and gelling agents in the formula affects the nanoemulgels (Negs) characteristics and stability. This study focuses on the systematic development, optimization, and characterization of Turmeric Essential Oil-loaded Nanoemulgels (TEO-loaded Negs), having promising topical applications. The TEO-loaded Negs were prepared by the high-speed homogenization method. The formula design used Central Composite Design from Response Surface Methodology (CCD-RSM) to obtain the optimal TEO-loaded Negs formula. The optimization variables of the TEO-loaded Negs formula include the concentrations of Span 80-Tween 80 (X1) and Carbopol 980 (X2) with the response of particle size (PS) (Y1), polydispersity index (PDI) (Y2), zeta potential (ZP) (Y3), pH (Y4), spreadability (Y5) and adhesion (Y6). The actual TEO-loaded Negs responses were compared with the CCD-RSM predictions. In addition, other physical evaluations such as homogeneity observations, phase separation tests, viscosity, and flow properties were carried out. The optimal TEO-loaded Negs developed were made with 8.68% Span 80-Tween 80 and 1.18% Carbopol 980. The evaluation results showed the optimal TEO-loaded Negs nano-metric size (182.3 ± 5.5 nm) with low PDI (0.242 ± 0.003), good ZP (-57.23 ± 2.91 mV), pH (4.51 ± 0.02), spreadability (6.0 ± 0.2 cm), and adhesion (6.45 ± 0.19). TEO-loaded Negs have a good appearance and do not undergo phase separation at extreme temperature storage. Thus, the developed TEO-loaded Negs can be a potential delivery system and a promising suitable approach for topical preparations.

KEYWORDS: Central composite design; nanoemulgels; response surface methodology; turmeric essential oil; topical delivery.

1. INTRODUCTION

Most water-based liquid or semisolid systems have limitations for delivering lipophilic drugs (1). Nanoemulsions are an established alternative for delivering lipophilic drugs by increasing topical absorption (2). The main advantage of topically administered nanoemulsions is the ability to increase penetration and permeation of active ingredients through the skin without adding chemical enhancers and non-friendly solvents to the formula, which can cause skin irritation, especially with long-term usage (2,3). However, this system has problems with low viscosity due to poor spreadability and retention in the skin (4). A combination of nanoemulsion and hydrogel was made to improve the characteristics, called nanoemulgels (Negs) (2,3,5).

Negs consist of two systems; an emulsion containing nano-sized globules and a hydrogel system. The emulsion acts as a drug delivery platform and is stabilized by an emulsifier in the form of surfactants and cosurfactants (2). Surfactants reduce the interfacial tension of immiscible liquids and change the entropy of the dispersion, thereby stabilizing a thermodynamically unstable system; cosurfactants are combined with surfactants in the emulsification process by disrupting the surface layer (6). The emulsifier plays a role in the emulsification process to increase stability when the product is stored for a long time. On the other side, gels are made from polymers, a gelling agent, that expand after absorbing a liquid (7). Gelling agents increase the viscosity of the formula and can react with surfactants to change the viscosity (8). In addition, topical Negs can improve patient compliance due to their non-greasy, non-irritating properties and better drug release (9).

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32 Turmeric essential oil (TEO) is extracted from the turmeric rhizome of *Curcuma Longa* L.
33 (*Zingiberaceae*) by steam distillation (10). Chemical constituents with the most significant proportion were
34 oxygenated monoterpenes, and sesquiterpenes include α -turmerone, β -turmerone (11-
35 13). The pharmacological activities of TEO have been reported in the form of antioxidants, anti-
36 inflammatory, antinociceptive, antidermatophytic, antifungal, and antibacterial activities (10,14-17). Like
37 other essential oil, TEO has limited use due to its volatility, instability under certain conditions, lipophilicity,
38 and low aqueous solubility (18,19). Many recent studies are oriented toward solving these limitations, so that
39 efficacy of the essential oil lasts longer and increases. Previously, TEO has been developed for cream as a
40 conventional drug delivery system; patch and nanoemulsions as drug delivery systems (15,20-22). However,
41 the available methods for manufacturing Negs exhibit various limitations, which directly or indirectly affect
42 the quality of the Negs formulations. Currently, the principle of quality by design is adopted to ensure the
43 quality of drugs, their safety, and efficacy (23). The quality by design (QbD) trend is used to design,
44 optimize, and examine the relationship between certain factors and their associated responses to obtain the
45 most optimal formula (24). Central Composite Design (CCD) on Response Surface Methodology (RSM) can
46 be used to select optimal formulas and predict models that rely on statistical analysis (ANOVA) and exact
47 equations (25,26). In this aspect, we intended to develop a TEO-Negs formulation containing turmeric oil
48 that would be optimized using a complete 22 factorial design and determine the independent factors' precise
49 influence on the investigated dependent variables. The choice and procedure factors were Span 80-Tween 80
50 (X1) and Carbopol 980 (X2). The process and formulation factors on particle size (PS) (Y1), polydispersity
51 index (PDI) (Y2), zeta potential (ZP) (Y3), pH (Y4), spreadability (Y5), and adhesion (Y6) were investigated.
52 This work is the first step in developing an optimized turmeric oil preparation suitable for transdermal drug
53 delivery for topical application.

54 2. RESULTS AND DISCUSSION

55 2.1. Preparation and Optimization of TEO-loaded Negs

56 TEO-loaded Negs are produced using the high-energy method efficiently. Variations in the emulsifier
57 and gelling agent concentration in different compositions and formulations of Negs have been produced. A
58 total of 14 formulas were prepared and optimized using Design-Expert®, version 13 software. The
59 optimized Negs are selected based on the minimum of PS, PDI, maximum ZP, pH range, spreadability, and
60 adhesion values. The analysis results show that the optimal formula is obtained with the composition of
61 Span 80-Tween 80 of 8.68% and Carbopol 980 of 1.18% based on minimum PS, PDI, maximum ZP, the
62 desired range of pH, dispersion, and adhesion that are suitable for the application transdermally.

63 2.2. Central Composite Design (CCD)

64 TEO-loaded Negs were optimized based on CCD in the RSM. CCD was applied to determine the
65 optimum emulsifier concentration (X1) and gelling agent concentration (X2) as the main factors influencing
66 the dependent response. Prediction of the factorial axial design and the possible curvature in the response
67 can be obtained from the optimization process with an effective second-level design (27). In developing
68 TEO-loaded Negs, prediction of the main effect of the independent variable on the dependent variable is
69 essential. Two factors were selected as independent variables based on the literature survey, and six
70 responses were decided as the dependent variable with the most significant effect on Negs. The independent
71 variables with their levels and the observed response variables are shown in Table 1.

72 The six responses show different models in their application, depending on the most considerable R-
73 squared value and the smallest residual predictive sum of squares value. The selected model showed a non-
74 statistically significant lack of fit, and model validation was confirmed by the residual plot test of the
75 regression model indicated from supplementary information for all responses. The quadratic model
76 describes the effects of various factors, including individual factors, interactions, and quadratic effects on
77 responses. Table 2 shows the statistical analysis of the quadratic model for the response of the PDI, the 2FI
78 model for the ZP response, and linear models for the response of PS, pH, spreadability, and adhesion. This
79 table selects factors with p-values below the pre-defined threshold (here 0.05, with a 95% confidence level) as
80 influential factors.

81 Particle size parameters are often used to characterize nanoparticles. The mean TEO-loaded Negs
82 globule diameter (Y1) was adjusted from 160.8 nm (STD#10) to 457.0 nm (STD#2). As shown in Table 2,

83 statistical analysis revealed that the most critical factor affecting the PS of Negs globules was the emulsifier
84 concentration because the p-value was more significant than other factors (p-value <0.01). The positive
85 coefficient has a synergistic effect on the response. In contrast, the negative coefficient has an antagonistic
86 effect which concludes the inverse relationship of the independent variable with the response (27,28). In
87 addition, this factor has a more significant coefficient. It directly affects the PS, which means that increasing
88 the emulsifier concentration causes a decrease in the diameter of the globule PS in the TEO-loaded Negs.
89 High surfactant concentrations (above 5.5%) resulted in globules measuring below 200 nm and low
90 surfactant concentrations (below 5.5%) producing globules above 200 nm. That fact is in line with other
91 studies that increasing surfactant concentrations can reduce droplet PS in Negs (29,30). The surfactant
92 lowered the interfacial tension between the oil and water phases, reducing the free energy required to
93 disrupt or break the droplets and resulting in a smaller droplet diameter. It can also form a protective layer
94 around the droplets and prevent them from coalescing with others. However, the emulsifier must absorb
95 quickly enough around the droplet to form this protective layer (30,31).

96 In Table 1, the polydispersity index of TEO-loaded Negs (Y2) varied from 0.000 (STD#2 and #4) to
97 0.571 (STD#1, #5, #6, #8, #9, #11 to #14). From the analysis results, no single factor significantly affects the
98 PDI because all p-values are > 0.05. In addition, no factors directly affect the PDI of the Negs. The PDI
99 measures the distribution of molecular mass in a sample. The smaller the PDI value (close to 0), the more
100 stable the formula from Negs is because the greater the PDI value indicates the particles formed are not
101 uniform, so that the formula will flocculate quickly. An index value less than 0.05 is included in
102 monodisperse, while an index greater than 0.7 indicates that the sample has a broad particle size
103 distribution. A-0.2 and below are considered acceptable for nanoparticle preparations (32).

104 The zeta potential of TEO-loaded Negs (Y3) was in the range of 13.90 mV (STD#2) to 46.72 mV
105 (STD#10) (Table 1). According to statistical analysis on CCD, the emulsifier concentration factor significantly
106 affected ZP compared to other factors. ZP represents the electric charge between the shear plane of the final
107 outer layer and the bulk solution, which significantly affects the stability of the dispersion (33). This factor is
108 strongly influenced by the composition of the Negs and its electrical phenomena. TEO-loaded Negs, which
109 have positive zeta potential, show good interaction with negatively charged skin (34). ZP is the scientific
110 term for the electrokinetic potential in colloidal systems. The high electric charge on the nanoparticle surface
111 will prevent the aggregation of the nanoparticles because of the strong repulsion between the particles. ZP is
112 usually influenced by the physicochemical properties of the drug, polymer, carrier, presence of electrolytes,
113 and their adsorption (35). The ZP requirement is above ± 30 mV. The higher the ZP value, the slower the
114 aggregation formed to prevent separation (36).

115 The pH test is carried out to measure the level of acidity or alkalinity of Negs. The pH values (Y4)
116 were in the range of 4.57 (STD#13) to 6.39 (STD#14) (Table 1). In this response, the gelling agent
117 concentration factor has a significant effect compared to other factors. The pH requirement of a topical
118 preparation is the same as the skin pH. Preparations that are too acidic can irritate the skin and cause a
119 stinging sensation, while preparations that are too alkaline can cause dry and itchy skin. The results of the
120 pH test carried out on 14 formulas were eligible, where compatible with the skin (4.5-6.5) (37). The increase
121 influenced the decrease in the pH value obtained from this study in the concentration of Carbopol. That is
122 caused by the reaction between the carboxylate group in Carbopol with water so that more H_3O^+ (acid) is
123 formed and makes the preparation more acidic.

124 On the spreadability response (Y5), the results were obtained in the range of 4.90 (STD#11) to 6.25
125 (STD#10). In statistical analysis, the emulsifier concentration factor significantly affects the dispersion of
126 Negs. The spreadability is carried out to see the dispersion of Negs on the skin. Terms of good dispersion are
127 5-7 cm. If the dispersion is too small, it is relatively difficult to spread when applied to the skin, while the
128 dispersion tends to spread too quickly when applied so that it will cause an uncomfortable feeling when
129 used (38). Based on the results of the spreadability test, only one Negs did not meet the requirements,
130 namely F11, ie. 4.9 cm. The smaller the concentration of Carbopol 980, the greater the dispersion obtained.
131 The higher the concentration of Carbopol in preparation, the more the preparation's viscosity. Adhesion and
132 spreadability have the opposite results. The higher the viscosity of preparation, the higher the adhesive
133 power produced, while the smaller the dispersion power (39).

134 Finally, the adhesion value is from 4.22 (STD#14) to 7.08 (STD#13) in Table 1. Statistically, this
135 adhesion was significantly affected by the concentration of the gelling agent. They met the requirements
136 based on the adhesion test results carried out on 14 formulas. An adhesion test is carried out to see how long
137 a preparation can be attached to the skin. The stickiness requirement is more than 4 seconds. The longer a

138 preparation can be attached to the skin, the better, where it is expected that more active substances can be
139 absorbed due to the time the preparation is in contact with the skin (40). This result also shows that the
140 smaller the concentration of Carbopol 980, the lower the adhesive power obtained. In the end, the emulsifier
141 concentration factor significantly affected the response of PS, ZP, and spreadability of TEO-loaded Negs. At
142 the same time, the concentration factor of the gelling agent affects the pH and adhesion.

143 2.3. Optimized TEO-loaded Negs

144 The analysis was carried out using CCD-RSM to determine the optimal conditions for the
145 concentrations of Span 80-Tween 80 and Carbopol 980 for manufacturing Negs. The optimal value
146 determines each response, and the results are used to perform optimization. Variables and response
147 measurements will be used by the program to perform optimization. The effect of variables in this study
148 were the concentrations of Span 80-Tween 80 and Carbopol 980 that responded to PS, PDI, ZP, pH,
149 spreadability, and adhesion. The formula with the maximum desirability value is the optimal formula
150 generated from the optimization phase of the program (41).

151 The optimization value formed is indicated by the desirability value close to one. The desirability
152 value range is 0-1. Figure 1 describes the optimization results in the form of a 2D contour. *Contour* is a two-
153 dimensional response image that is presented using a predictive model for the response values of PS, PDI,
154 ZP, pH, spreadability, and adhesion. The contour graph shows the desirability value of 0.801, which is the
155 closest value to 1 compared to the other points. Figure 2 shows the projection in the form of a 3D surface; the
156 low area shows low desirability while the high area shows high desirability and is getting closer to 1. 25, ZP
157 56.30 mV, pH 4.5, adhesion 6.98 seconds, and spreadability 6.07 cm.

158 2.4. Evaluation of TEO-loaded Negs

159 Optimal TEO-loaded Negs evaluations were carried out, including organoleptic, homogeneity, freeze-
160 thaw, PS, PDI, ZP, pH, adhesion and spreadability, viscosity, and flow properties be seen in Tables 4 and 5.
161 In addition, Table 3 contains the actual and predicted values of the optimized Negs for each response-
162 dependent variable. The optimum formula results from the predictions of the RSM program using the
163 Central Composite Design (CCD) model are the concentrations of Span 80-Tween 80 of 8.68% and Carbopol
164 980 of 1.18%.

165 The particle size of the optimum formula is 182.3 nm; this result is almost close to the RSM
166 prediction's PS, 180.2 nm. The PDI value of the optimum formula obtained is 0.242; this result is close to the
167 PDI result of the RSM prediction, which is 0.250. The ZP of the optimum formula is 57.23 mV; this result is
168 close to the ZP of the RSM prediction, which is 56.30 mV. The pH value obtained from the optimum formula
169 is 4.51; this result is close to the pH value predicted by RSM, which is 4.5. The adhesion of the optimum
170 formula is 6.45 seconds; this result is close to the result of RSM prediction, which is 6.98 seconds. The
171 spreadability obtained from the optimum formula is 6.00 cm; this result is close to the dispersion from the
172 RSM prediction, which is 6.07 cm.

173 Based on Table 4, the organoleptic results of TEO-loaded Negs have a distinctive turmeric odor, are
174 white, and are semi-solid. The homogeneity test results of the optimum TEO-loaded Negs formula showed a
175 homogeneous preparation, as evidenced by the absence of coarse grains on the object glass. The results of
176 the Freeze-Thaw test on the optimum formula for six cycles showed promising results; namely, there was no
177 separation. It aims to see the separation of the water and oil phases due to the influence of extreme
178 temperatures (42). Viscosity testing of the optimal formula for TEO-loaded Negs was carried out using a
179 Brookfield digital RV DV-E spindle seven at a speed of 50 rpm. The viscosity of the TEO-loaded Negs
180 preparation was 32240 cP, indicating reasonably high viscosity. That is due to the neutralization of the gel
181 base with the addition of triethanolamine. Carbopol is dispersed in water to form an acidic colloidal solution
182 with low viscosity. Neutralizing the gel base will increase the viscosity of the preparation because the gelling
183 agent expands well. Judging from the dispersion and adhesion produced, although the TEO-loaded Negs
184 have a reasonably high viscosity, the spreadability follows the requirements. In contrast, the adhesion power
185 is relatively short despite meeting the requirements.

186 The results of the flow properties test showed that the optimal formula made was a thixotropic plastic
187 flow type. *Thixotropic* is a flow property expected in pharmaceutical preparations because it has high
188 consistency in the container but can be poured and dispersed easily. That can be seen from the graph results,

189 which show that the descending curve is to the left of the ascending curve. The plastic flow curve does not
190 pass through the point (0,0) but intersects the shearing stress axis (or will intersect if the straight section of
191 the curve is extrapolated to the axis) at a certain point called the yield value. The yield value is the amount of
192 force or shear stress that must be exceeded so that the preparation can flow. The larger the flocculated
193 substance in the preparation, the greater the yield value (43). From the curve obtained, the yield value is
194 224,276 dyne/cm³.

195 4. CONCLUSION

196 Based on the results of the RSM analysis, the optimum concentration of Span 80-Tween 80 as a
197 surfactant was 8.68%, and Carbopol 980 as a gelling agent was 1.18%. The resulting response is a PS of 182.3
198 nm, PDI 0.242, ZP 57.23 mV, pH 4.51, adhesion of 6.45 seconds, and spreadability of 6 cm. The viscosity of
199 the optimum formula is 32240 cP with thixotropic plastic flow properties. Thus, the developed TEO-Negs
200 can be a potential delivery system and a promising suitable approach for topical preparations.

201 5. MATERIALS AND METHODS

202 5.1. Materials

203 TEO (*Curcuma longa*) purchased from Darjeeling Sembrani Aroma (Indonesia), Sorbilene O E/P from
204 Lamberti (Italy), Span 80 from Croda (Singapura), Propylene glycol from Dow Chemical Pacific (Singapura),
205 Carbopol 980 NF from Lubrizol AM (Cleveland), Nipagin M from Clariant Produkte (Deutschland), Propyl
206 Paraben from Alpha Chemika (India), and Triethanolamine from Dow Chemical Pacific (Switzerland).

207 5.2. Methods

208 5.2.1. Preparation of TEO-loaded Negs

209 TEO-loaded Negs are made using a high-energy method, which uses a mechanical device to produce a
210 highly disruptive force to break up the oil and water phases to obtain nano-sized globules (2). The oil phase
211 (M1) was prepared by mixing Span 80 with 5% turmeric oil using a magnetic stirrer (WiseStir Wisd) at 1500
212 rpm for 20 min. A total of 0.18% methylparaben and 0.02% propylparaben dissolved in 15% propylene glycol
213 (M2). Then the distilled water was stirred with tween 80, and M2 was added gradually until homogeneous
214 at 1500 rpm for 20 min (M3). M1 was stirred with M3 until homogeneous at 1500 rpm for 40 min to form a
215 clear and transparent nanoemulsion, then let left for 24 hours. The gel base was prepared by mixing
216 Carbopol 980 NF with distilled water at 70°C and left for 24 hours. Then add gradually 1% TEA to form a gel
217 mass. Nanoemulsion was added slowly into the gel base while homogenized using a homogenizer (AEG) at
218 2000 rpm for 10 min.

219 5.2.2. Experimental design (CCD-RSM)

220 This study selects the CCD-RSM method to develop the TEO-loaded Negs formulation. A 2-factor CCD-RSM
221 at two levels (high and low) was used for preliminary screening on PS, PDI, ZP, pH, spreadability, and
222 adhesion. Based on previous experiments and literature reviews, high and low levels of variables were
223 determined. CCD of the statistical package Design-Expert® version 13 software (Stat-Ease Inc., Minneapolis,
224 MN) was used to assess the effect of the selected independent variable on the response variable to obtain the
225 optimal formula for TEO-loaded Negs. CCD planned 14 experiment runs under controlled circumstances
226 (Table 5). The independent variables were Span 80-Tween 80 concentration (X1) and Carbopol 980
227 concentration (X2). The observed response of the dependent variables was PS (Y1), PDI (Y2), ZP (Y3), pH
228 (Y4), spreadability (Y5), and adhesion (Y6).

229 5.2.3. Determination of PS, PDI, and ZP of Negs

230 PS and PDI of Negs were determined by dynamic light scattering (DLS) technique or photo correlation
231 spectroscopy using the Delsa Max Pro Particle Size Analyzer LS 100Q (Beckman Coulter, USA) at 25°C. For
232 analysis, 1 mL of sample was dispersed in 9 mL aqua pro injection. Into the cuvette, 1 mL of suspension and
233 5 mL of aqua pro injection were added as a diluent, and the results were read on the instrument. All

234 measurements were made at a scattering angle of 90°. ZP was determined by particle size analyzer through
235 mobility and conductivity measurements. The temperature and the mean electric field were applied at 25°C
236 and 16 V/cm, respectively (44). The mean of the three repeated measurements of each sample is reported as
237 the final result.

238 5.2.4. Determination of pH, spreadability, and adhesion

239 pH was measured at a temperature of 25°C using a pH meter that had been previously calibrated with buffer
240 solutions of pH 4 and 7. The calibration process was completed when the pH value indicated on the screen
241 matched the expected pH value and was stable. Afterward, the electrode was dipped in Negs and recorded
242 the value shown on the screen (45). Spreadability was measured by adding 0.5 g of Negs in the center of a
243 glass covered with another glass. Measurement of the diameter of the preparation distribution
244 longitudinally and transversely, and every minute of adding 50 g to a total weight of 150 g (46).
245 Adhesiveness was determined by placing 0.5 g of Negs on a slide, then covering it with another slide, and
246 being given a load of 1 kg for 3 minutes. The glass object was mounted on the test apparatus, and 80 g of the
247 load was released until both glass objects were released, and the time was recorded (46).

248 5.2.5. Visual observation and homogeneity test

249 Negs are placed in a glass object and directly observed for color, smell, and shape (47). A-0.1 g Negs were
250 spread over the slide, and homogeneity was observed. If there are no coarse grains, the test preparation is
251 declared homogeneous (48).

252 5.2.6. Viscosity and rheological flow

253 Viscosity and rheology were determined with a Brookfield RV DV-E Viscometer with appropriate spindle
254 and speed. Negs were put into a beaker glass until it reached 500 mL; the spindle was installed, and the
255 measured value was recorded as viscosity Negs. In this study, the spindle used is spindle no. 7. Flow
256 properties are determined by measuring the viscosity using a suitable spindle from low to high rotational
257 speed and vice versa (45).

258 5.2.7. Freeze-thaw test

259 Negs were stored at $4 \pm 2^\circ\text{C}$, then transferred to $40 \pm 2^\circ\text{C}$ for 48 hours (1 cycle), then repeated for 6 cycles.
260 Phase separation was observed in each cycle (45).

261 5.2.8. Statistical analysis

262 Statistical analysis was performed using the Design-Expert® version 13 software (Stat-Ease Inc.,
263 Minneapolis, MN). All measurements were repeated three times, and the analysis was performed at a sig. p
264 < 0.05 .

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Table 1. Evaluation results of independent variables and dependent variables with the CCD design.

Std	X ₁ (%)	X ₂ (%)	Y ₁ (nm)	Y ₂	Y ₃ (mV)	Y ₄	Y ₅ (cm)	Y ₆ (sec)
1	5.50	1.02	206.9±0.7	0.571±0.000	35.39±1.50	5.45	5.55±0.07	5.65
2	2.32	1.15	457.0±0.9	0.000±0.000	13.90±1.92	4.60	5.00±0.00	6.99
3	8.68	0.90	166.7±0.4	0.378±0.074	31.64±0.53	6.35	6.20±0.14	4.23
4	2.32	0.90	411.0±5.8	0.000±0.000	30.67±0.89	6.29	5.00±0.00	4.30
5	5.50	1.02	302.1±3.6	0.571±0.000	36.45±0.45	5.46	5.60±0.28	5.63
6	5.50	1.02	263.0±3.7	0.571±0.000	18.49±1.77	5.47	5.55±0.21	5.64
7	8.68	1.15	198.1±1.2	0.242±0.003	57.23±2.91	4.65	6.15±0.07	6.95
8	5.50	1.02	246.9±7.2	0.571±0.000	26.60±0.17	5.46	5.60±0.14	5.67
9	5.50	1.02	206.0±2.3	0.571±0.000	31.77±1.01	5.46	5.60±0.00	5.65
10	10.00	1.02	160.8±1.3	0.285±0.024	46.72±1.97	5.31	6.25±0.21	5.84
11	1.00	1.02	244.2±4.3	0.571±0.000	31.26±0.57	5.78	4.90±0.14	6.88
12	5.50	1.02	238.1±3.0	0.571±0.000	30.78±0.53	5.45	5.55±0.07	5.64
13	5.50	1.20	206.2±1.6	0.571±0.000	30.56±0.78	4.57	5.50±0.14	7.08
14	5.50	0.85	240.5±2.2	0.571±0.000	20.12±0.16	6.39	5.65±0.21	4.22

Y1: Particle size (PS), Y2: Polydispersity index (PDI), Y3: Zeta potensial (ZP), Y4: pH, Y5: Spreadability, Y6: Adhesion

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381

382 **Table 2.** Statistical analysis of PS (Y1), PDI (Y2), ZP (Y3), pH (Y4), spreadability (Y5), and adhesion (Y6) TEO-loaded
383 Negs on CCD.

Factors		Y1	Y2	Y3	Y4	Y5	Y6
A	Coefficient	-77.64	0.027	8.27	-0.069	0.53	-0.20
	p-value	0.0063**	0.7325	0.0080**	0.1430	<0.0001**	0.1406
B	Coefficient	3.61	-0.017	2.95	-0.75	-0.033	1.18
	p-value	0.8783	0.8289	0.2668	<0.0001**	0.1157	<0.0001**
AB	Coefficient		-0.034	10.59			
	p-value		0.7602	0.0136*			
A ²	Coefficient		-0.16				
	p-value		0.0818				
B ²	Coefficient		-0.086				
	p-value		0.3087				
Intercept	Coefficient	253.39	0.57	31.54	5.48	5.58	5.74
Degree of freedom		2	5	3	2	2	2
Sum of squares		48332.05	0.24	1065.33	4.48	2.28	11.49
Mean of squares		24166.02	0.047	355.11	2.24	1.14	5.74
F-value		5.69	1.02	7.06	145.08	386.43	46.35
p-value		0.0201	0.4645	0.0078	<0.0001	<0.0001	<0.0001
R-Squared		0.5084	0.3898	0.6794	0.9635	0.9860	0.8939

384 A: X1 (Span 80-Tween 80); B: X2 (Carbopol 980)

385 * p-value < 0.05

386 ** p-value < 0.01

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Table 3. Optimized TEO-loaded Negs actual and predicted values for each response.

Responses	Predicted values	Actual value ^a	Error ^b (%)
Y1 (nm)	180.2	182.3 ± 5.5	1.165
Y2	0.250	0.242 ± 0.003	-3.200
Y3 (mV)	56.30	57.23 ± 2.91	1.652
Y4	4.50	4.51 ± 0.02	0.222
Y5 (cm)	6.07	6.0 ± 0.2	-1.153
Y6 (sec)	6.98	6.45 ± 0.19	-7.593

a Data listed is the mean ± standart deviation, n = 3

b Error (%) = [(Actual value - Predicted value)/Predicted value] * 100%

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Table 4. Additional evaluation on TEO-loaded Negs with an optimal formula.

Evaluation	Result
Organoleptic	Color: White; Odor: typical turmeric
Homogeneity	Homogeneous
<i>Freeze-Thaw</i>	Cycles 0-6, no separation occurs
Viscosity	32240 cP

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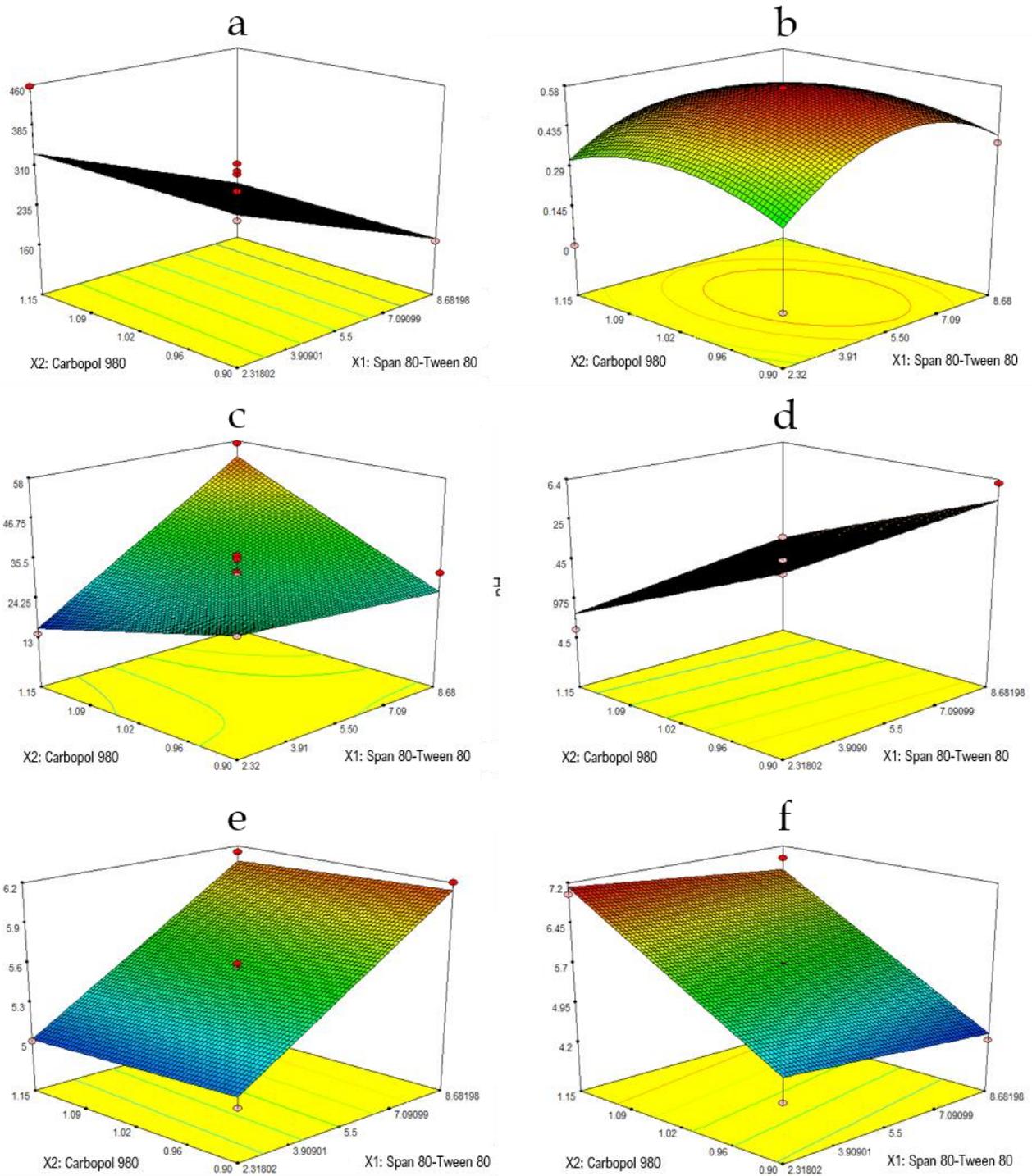
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Table 5. The independent and dependent variables with levels and limits in CCD for TEO-loaded Negs development

Variables	Code	Low level	High level	Units
Independent variables				
Span 80-Tween 80 concentration	X1	2.32	5.50	% w/w
Carbopol 980 concentration	X2	0.90	1.02	% w/w
Dependent variables				
Limits				
Particle size	Y1	Minimum		nm
Polydispersity index	Y2	Minimum		
Zeta potensial	Y3	Maximum		mV
pH	Y4	is in range		
Spreadability	Y5	is in range		cm
Adhesion	Y6	is in range		

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Figure 1. Effect of Span 80-Tween 80 and Carbopol 980 concentration on PS (a) PDI (b) ZP (c) pH (d) spreadability (e) adhesion (f).

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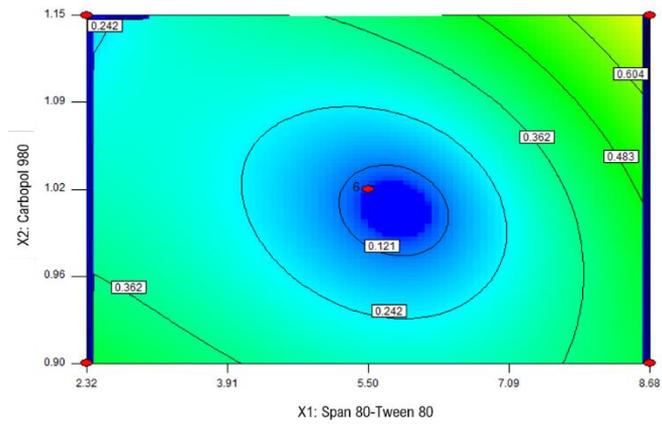


Figure 2. Contour plot desirability value of optimal formula.

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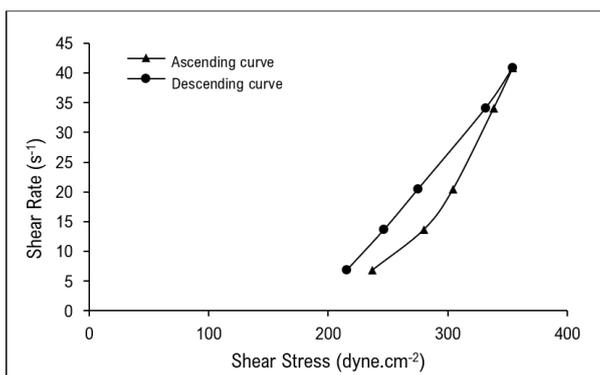


Figure 3. Optimal formula flow properties.

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**2. Bukti Konfirmasi Review dan Hasil Review
Pertama (5 November 2022)**

Journal of Research in Pharmacy : Revision request for your manuscript

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Suggestions

1. Reviewer Comments

Comments to the Author

A very nice and well-designed study about the the turmeric essential oil-loaded nanoemulgel. Authors have fully characterized the prepared formulations while their conclusions are supported by their results. However, the some comments must be taken into consideration for the improvement of the study. I'm sending my comments and recommendations.

1. Please carefully check the text for minor grammar and syntax errors.
2. Abbreviations should be used at the first occurrence of the word.
For example: Nanoemulgels (Negs)
3. Attention should be paid to punctuation marks and spaces In vitro and in vivo should be written in italics.
4. Why was the quantification of turmeric essential oil and drug content study of nanoemulgel not done?
5. Why did the authors choose turmeric essential oil? How the authors analyzed the effectiveness of the oil.
6. Result and discussion part should be supported by the literatures.
7. In methods: Explain what TEA means.
8. Adhesiveness is the work required to overcome the attractive forces between the surface of the sample and the probe surface. The unit of adhesiveness cannot be seconds. The authors should explain the adhesion study in more detail. The results of adhesion should be supported by the literature.

2. Reviewer Comments

The content of the manuscript and the topic seems in line with journals target audience but there are some serious drawbacks need to be addressed:

- The English of the manuscript needs to be revised by a native speaker or a professional editing service,
- The application of CCD method is not clear nor understandable in the manuscript. In particular, how many center points were used, which $\pm\alpha$ was preferred in the design etc?
- The most significant missing factor is validation of the CCD. After application of CCD-RSM, suggested optimal points are better presented in table3. Once the optimal points are clarified, three Negs formulation at this suggested optimal point are needed to be prepared experimentally along with all the characterization in order to show the correlation between predicted and experimentally found values.
- The purpose of the Negs formulation is not clear in the manuscript which is another bottleneck of the manuscript. The purpose of this formulation needs to be addressed and further characterized along with the purpose.

Manuscript Information

Manuscript ID: MPJ-10622**Title in English:** Response surface methodology for optimization of turmeric essential oil-loaded nanoemulgel**Small Title in English:** No information entered**Authors:** Nining Nining¹, Anisa Amalia¹, Fatimatuz Zahrok²**Institutions:** ¹Universitas Muhammadiyah Prof. DR. HAMKA, Pharmaceutical Technology, East Jakarta, Indonesia²Universitas Muhammadiyah Prof. DR. HAMKA, Pharmacy, East Jakarta, Indonesia**Keywords in English:** Central composite design; nanoemulgels; response surface methodology; turmeric essential oil; topical delivery.**Manuscript Type:** Research article**Processing Status:** Major Revision**Abstract in English**

Turmeric essential oil has antioxidant and anti-inflammatory activities in a topical dosage form. Nanoemulgels are one of them. Using emulsifiers and gelling agents in the formula affects the nanoemulgels (Negs) characteristics and stability. This study focuses on the systematic development, optimization, and characterization of Turmeric Essential Oil-loaded Nanoemulgels (TEO-loaded Negs), having promising topical applications. The TEO-loaded Negs were prepared by the high-speed homogenization method. The formula design used Central Composite Design from Response Surface Methodology (CCD-RSM) to obtain the optimal TEO-loaded Negs formula. The optimization variables of the TEO-loaded Negs formula include the concentrations of Span 80-Tween 80 (X1) and Carbopol 980 (X2) with the response of particle size (PS) (Y1), polydispersity index (PDI) (Y2), zeta potential (ZP) (Y3), pH (Y4), spreadability (Y5) and adhesion (Y6). The actual TEO-loaded Negs responses were compared with the CCD-RSM predictions. In addition, other physical evaluations such as homogeneity observations, phase separation tests, viscosity, and flow properties were carried out. The optimal TEO-loaded Negs developed were made with 8.68% Span 80-Tween 80 and 1.18% Carbopol 980. The evaluation results showed the optimal TEO-loaded Negs nano-metric size (182.3 ± 5.5 nm) with low PDI (0.242 ± 0.003), good ZP (-57.23 ± 2.91 mV), pH (4.51 ± 0.02), spreadability (6.0 ± 0.2 cm), and adhesion (6.45 ± 0.19). TEO-loaded Negs have a good appearance and do not undergo phase separation at extreme temperature storage. Thus, the developed TEO-loaded Negs can be a potential delivery system and a promising suitable approach for topical preparations.

Manuscript Files

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MPJ-10622-9-isi-jrp-checklist.pdf (../pdf-files/in/10622-MPJ-10622-9-isi-jrp-checklist.pdf)	3388 KB	Sep 03, 2022	Author Checklist Form	None
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Score Sheet

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necessary ?

Yes

Are the figures tables
and graphics clear ?

No

Is the introduction
part

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Are the experimental
procedures sound?

No

Is the results and
discussion part

sufficiently developed

Is conclusion
sufficient and
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results ?

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COMMISSION
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**3. Bukti Konfirmasi Review Submit Revisi
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Response surface methodology for optimization of turmeric essential oil-loaded nanoemulgel

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ABSTRACT: Turmeric Essential Oil (TEO) has an antioxidant and anti-inflammatory activity to be formulated in a topical dosage form. Nanoemulgels (Negs) development, based on varying concentrations of emulsifiers and gel formers, affects their characteristics and stability. This study focuses on optimizing TEO-loaded Negs based on physical and mechanical characterization, which have promising topical applications. Negs were created using the high-energy approach and optimized using Response Surface Methodology (RSM) and the Central-Composite Design (CCD) for the optimization of span-80/tween-80 (X_1) and Carbopol® 980. (X_2). Observed variable responses were particle size (PS) (Y_1), polydispersity index (PDI) (Y_2), zeta potential (ZP) (Y_3), pH (Y_4), spreadability (Y_5), and adhesion time (AT) (Y_6). Actual responses of Negs were compared with the CCD-RSM predictions to validate the model. In addition, other physical evaluations were observed, such as organoleptic observations, homogeneity, freeze-thaw tests, viscosity, and flow properties. Optimized TEO-loaded Negs were made with 8.68% span-80/tween-80 and 1.18% Carbopol® 980. The evaluation results showed the optimal TEO-loaded Negs on nano-metric size (182.3 ± 5.5 nm) with low PDI (0.242 ± 0.003), good ZP (-57.23 ± 2.91 mV), pH (4.51 ± 0.02), spreadability (6.0 ± 0.2 cm), and AT (6.45 ± 0.19 s). TEO-loaded Negs have an excellent appearance and did not run phase separation at extreme temperature storage with pseudoplastic thixotropy flow. Thus, the developed TEO-loaded Negs can be a potential delivery system and a promising suitable approach for topical preparations.

KEYWORDS: Central composite design; nanoemulgels; response surface methodology; turmeric essential oil; topical delivery.

1. INTRODUCTION

Turmeric is the dried rhizome of *Curcuma longa* L. (Zingiberaceae), which derives from Southeast Asia and is cultivated mainly in India, followed by Bangladesh, China, Thailand, Cambodia, Indonesia, Malaysia, and the Philippines (1). Steam distillation extracts turmeric essential oil (TEO) from the turmeric rhizome (2). Chemical constituents with the most significant proportion were oxygenated monoterpenes and sesquiterpenes, which include β -turmerone, α -turmerone, and ar-turmerone (3-5). The pharmacological activities of TEO have been reported in the form of antioxidants, anti-inflammatory, antinociceptive, antidermatophytic, antifungal, and antibacterial activities (2,6-9). These reducing power and radical scavenging abilities are associated with the high antioxidant potential of TEO (8,10). This pharmacological activity justifies its use in various applications, including cosmetics and phytomedicines (7,11,12). Like other essential oil, TEO has limited use due to its volatility, instability under certain conditions, lipophilicity, and low aqueous solubility (13,14). Many recent studies are oriented toward solving these limitations, so that efficacy of the essential oil lasts longer and increases.

Previously, TEO has been developed for cream as a conventional drug delivery system, patch, and nanoemulsions (7,11,15,16). Most water-based liquid or semisolid systems have limitations in delivering lipophilic drugs (17). Nanoemulsions are an established alternative for delivering lipophilic drugs by increasing topical absorption (18). The main advantage of topically administered nanoemulsions is the ability to increase penetration and permeation of drugs through the skin without adding non-physical enhancers and non-friendly solvents to the formulation, which can cause skin irritation, especially with long-term usage (18,19). However, this system has problems with low viscosity due to poor spreadability and skin

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33 retention (20). Nanoemulgels (Negs), a combination of nanoemulsion and hydrogel, were made to improve
34 the characteristics (18,19,21).

35 Negs consist of a hydrogel system and an emulsion with nano-sized globules. An emulsifier in the
36 form of surfactants and co-surfactants stabilizes the emulsion, which serves as a drug delivery platform (18).
37 Surfactants reduce the interfacial surface tension of immiscible liquids and change the entropy of the
38 dispersion, thereby stabilizing a thermodynamically unstable system; co-surfactants are combined with
39 surfactants in the emulsification process by disrupting the surface layer (22). The emulsifier plays a role in
40 the emulsification process to increase stability when the product is kept for an extended time. On the other
41 side, gels are made from polymers, a gel former, that expand after absorbing a liquid (23). Gel former
42 increase the viscosity of the formulation and can react with emulsifiers to change the thickness (24). In
43 addition, topical Negs can increase patient compliance because of their non-irritating, non-greasy
44 characteristics and improved drug release (25). However, the available methods for manufacturing Negs
45 exhibit various limitations, which directly or indirectly affect the quality of the Negs formulations.

46 Currently, the principle of quality by design is adopted to ensure the quality of drugs, their safety, and
47 efficacy (26). The quality by design (QbD) trend is used to develop, optimize, and investigate the interaction
48 between particular variables and their related responses to achieve the optimal formulation (27). Our study
49 used Central Composite Design (CCD) on Response Surface Methodology (RSM) to select optimal
50 formulations and predict models that rely on statistical analysis (ANOVA) and exact equations (28,29). In
51 this aspect, we intended to develop a TEO-loaded Negs that would be optimized using a complete 2²-
52 factorial design and determine the independent factors' precise influence on the investigated dependent
53 variables. The choice and procedure factors were span-80/tween-80 (X₁) and Carbopol® 980 (X₂). Response
54 variables investigated were particle size (PS) (Y₁), polydispersity index (PDI) (Y₂), zeta potential (ZP) (Y₃),
55 pH (Y₄), spreadability (Y₅), and adhesion time AT) (Y₆). This work is the first step in developing an
56 optimized TEO preparation suitable for transdermal drug delivery for topical application.

57 2. RESULTS AND DISCUSSION

58 2.1. Identification of TEO

59 TEO identification is carried out to ensure the quality of the active ingredients. The tests included
60 organoleptic, phytochemical identity in the form of phenolics and terpenoids, and antioxidant activity by the
61 DPPH method. The results of this study can be seen in **Figure 1**. Organoleptically, TEO is an orange liquid
62 with a distinctive turmeric aroma. Identification results in the phenolic test showed a black-green color
63 (**Figure 1A**) due to the formation of a phenolic-Fe³⁺ complex (30). In the terpenoid test, reaction results show
64 a purple color (**Figure 1B**) as a positive sign. Based on these results, it can be concluded that TEO contains
65 phenolic and terpenoid compounds. Several studies reported that TEO has powerful antioxidant potential
66 both *in-vitro* and *in-vivo* (2,8,31). Oxygenated monoterpenes and sesquiterpenes, such as α -turmerone,
67 α -turmerone, and β -turmerone, were included in the terpenoid compounds and reported to play a role in this
68 activity. In addition, contained phenol was also reported to perform as an excellent free radical scavenger
69 because its reduction potential was lower than oxygen's (8). In our study, the antioxidant activity of TEO
70 was presented as an IC₅₀ of 9.88 μ m/ml, which was included in the powerful antioxidant category because of
71 below 50 μ m/ml (32). It differed from other studies' results, which showed an IC₅₀ of 2274.02 μ m/ml and
72 above 1000 μ m/ml (2,10). Meanwhile, similar findings stated IC₅₀ values of 10.03 μ m/ml, 3.227 μ m/ml, and
73 14.5 μ m/ml (33–35). The difference value was due to the use of TEO from different origin sources, regional
74 conditions, and the extraction method used.

75 2.2. Preparation of TEO-loaded Negs

76 TEO-loaded Negs were produced using the high-energy method efficiently. A high-pressure
77 homogenizer provides sufficient energy to increase the interfacial area and generate nano-size globules (36).
78 An emulsion system with a nanodroplet size must have flow properties that allow it to pass through the
79 homogenizer (37). Flow properties are inversely correlated with the amount of emulsifier and gel former
80 added to the Negs formulation. In this study, the selected optimized formulation was the one that produces
81 Negs with the smallest size, as shown by the limitations in **Table 3**. Therefore, the range of regulated
82 emulsifiers and gel formers concentrations still provides flow properties that allow the flow to remain good
83 using a high-energy homogenizer. In general, the preparation of Negs was carried out in 3-stages, namely,

84 nanoemulsions preparation, gel preparation, and Negs preparation. Negs were successfully produced in 14-
85 formulation with variations of emulsifiers and gel former with compositions that can be seen in **Table 1**.

86 2.3. Experimental Design by CCD-RSM

87 2.3.1. Fitting the Model

88 TEO-loaded Negs were optimized based on CCD in the RSM. CCD was used to establish the optimal
89 concentrations of emulsifier (X_1) and gel former (X_2) as the key parameters influencing the dependent
90 response. Prediction of the factorial axial design and the possible curvature in the response could be
91 obtained from the optimization process with an effective second-level design (38). Predicting the significant
92 influence of the independent variable on the dependent variable is essential for generating TEO-loaded
93 Negs. Based on the literature survey, two factors were chosen as independent variables, and six responses
94 were chosen as the dependent variable with the most significant influence on Negs. The independent
95 variables with their levels and the observed response variables are presented in **Table 1**.

96 2.3.2. Analysis of Design

97 Statistical data analysis must be carried out to predict and recognize the model. **Table 2** shows the
98 statistical analysis of a quadratic model for PDI, a 2FI model for ZP, and linear models for PS, pH, AT, and
99 spreadability. This table identifies factors with p-values less than a predefined threshold (0.01 and 0.05, with
100 a 99% and 95% confidence level, respectively) as influential factors. Besides the significant p-value, a large F-
101 value minimizes error in the model and lack of fit, preferably non-significant to fit the model data (27,39).
102 Depending on the most significant R-squared value and the least residual predictive sum of squares value,
103 the six responses demonstrated distinct models in their application. The chosen model had a non-statistically
104 significant lack of fit, and model validation was confirmed by the residual plot test of the regression model,
105 which was supported by supplemental information for all responses. Compared to the 2FI and simple linear
106 models, the quadratic model represented the impacts of numerous variables, including individual factors,
107 interactions, and the quadratic influence on the response.

108 2.3.3. Effect of Independent Variables on Dependent Variables

109 PS parameters are often used to characterize nanoparticles. Negs globule diameter means (Y_1) was
110 adjusted from 160.8 nm (STD#10) to 457.0 nm (STD#2). Based on **Table 2**, the PS response indicates a
111 significantly linear model with an F-value of 5.69 (p-value 0.0201 < 0.05). The suggested linear model
112 equation can be seen as follows:

$$113 Y_1 = 253.39 - 77.64X_1 + 3.61X_2$$

114 The equation shows that the Negs PS is significantly (p-value < 0.01) affected by emulsifier
115 concentration (X_1). The positive coefficient has a synergistic effect on the response. In contrast, the negative
116 coefficient has an antagonistic effect which concludes the inverse relationship of the independent variable
117 with a response (38,40). In addition, this factor has a more significant coefficient. It directly affects the PS,
118 which means that increasing emulsifier concentration causes a decrease in the globule diameter PS in the
119 TEO-loaded Negs. High emulsifier (above 5.5%) resulted in globules measuring below 200 nm, and low
120 emulsifier (below 5.5%) produced globules above 200 nm. This fact is in line with other studies that
121 increasing emulsifiers could reduce droplet Negs PS (41,42). The emulsifier reduced the interfacial surface
122 tension between the water and oil phases, which decreased the free energy required to disrupt or break the
123 globules and resulted in a smaller droplet diameter. It can also produce a protective cover over the globules,
124 preventing them from coalescence. However, the emulsifier must absorb quickly enough around the droplet
125 to form this protective layer (42,43).

126 Meanwhile, Carbopol® 980 had no significant effect on PS. The same findings were obtained from the
127 globule size Negs from Carbopol® 934 and 940 as gel former (19). A response surface plot (**Figure 2A**) may
128 therefore be used to represent the combined influence of variables X_1 and X_2 , which shows that Y_1 changes
129 linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with
130 span-80/tween-80 (X_1) – not Carbopol® 980 (X_2) – was evidenced from the comparative plot of the response
131 surface. From this explanation, it can be concluded that the PS can be changed by selecting the right X_1 level.

In **Table 1**, Negs PDI (Y_2) varied from 0.000 (STD#2 and #4) to 0.571 (STD#1, #5, #6, #8, #9, #11 to #14). PDI measures the distribution of molecular mass in a sample. The smaller PDI (close to 0), the more stable the Negs formulation caused; the large PDI indicates particles formed are not uniform, and the formulation will flocculate quickly. An index value less than 0.05 is included in monodisperse, while an index greater than 0.7 indicates that the sample has a broad PS distribution. A-0.2 and below are considered acceptable for nanoparticle preparations (44). Based on **Table 2**, the PDI response shows a non-significant quadratic model with an F-value of 1.02 (p-value 0.4645 > 0.05). Measurements between variables and responses are not a precise cause of that. The suggested quadratic model equation can be seen as follows:

$$Y_2 = 0.57 + 0.027X_1 - 0.017X_2 - 0.034X_1X_2 - 0.16X_1^2 - 0.086X_2^2$$

The equation and p-value of each factor did not significantly affect PDI. The response surface plot (**Figure 2B**) depicts the combined influence of variables X_1 and X_2 , which shows that Y_2 changes with the sum of the two variables by model.

The Negs ZP (Y_3) was in the range of 13.90 mV (STD#2) to 46.72 mV (STD#10) (**Table 1**). ZP represents the electric charge between the shear plane of a final outer layer and bulk solution, which significantly affects dispersion stability (45). This factor is strongly influenced by the composition of the Negs and its electrical phenomena. TEO-loaded Negs, which have positive ZP, show good interaction with negatively charged skin (46). ZP is the scientific term for the electrokinetic potential in colloidal systems. The high electric charge on the nanoparticle surface will prevent nanoparticle aggregation because of the strong repulsion between particles. The ZP requirement for stability is above ± 30 mV (19). The higher ZP and the slower aggregation are formed to prevent separation (47). Based on **Table 2**, the ZP response indicates a significant 2FI model with an F-value model of 7.06 (p-value 0.0078 < 0.05). The linear model equation suggested by the software can be seen as follows:

$$Y_3 = 31.54 + 8.27X_1 + 2.95X_2 + 10.59X_1X_2$$

The equation shows that the Negs ZP was significantly affected by the emulsifier (X_1) and gel former-emulsifier interaction (X_1X_2). ZP was usually influenced by the physicochemical properties of the drug, polymer, carrier, electrolyte presence, and their adsorption (48). One study stated that adding Carbopol® only slightly increased the ZP Negs (19). The response surface plot (**Figure 2C**) depicts the combined influence of variables X_1 and X_2 , showing that Y_3 changes with the sum of the two variables. Nevertheless, the higher gradient in the response surface with span-80/tween-80 (X_1) - not Carbopol® 980 (X_2) - is the evidence from the comparative plot of the response surface. This description concludes that the ZP can be changed by selecting the proper X_1 level.

The pH test was carried out to measure Negs's acidity or alkalinity level. The pH values (Y_4) were in the range of 4.57 (STD#13) to 6.39 (STD#14) (**Table 1**). The pH requirement of Negs is the same as the skin pH. Too-acidic preparations can irritate the skin and cause a stinging sensation, while too-alkaline preparations can cause dry and itchy skin. The pH result test on 14 formulations was eligible and compatible with the skin (4.5-6.5) (49). Based on **Table 2**, a linear model was found to be significant in pH response with a model F-value of 145.08 (p-value < 0.0001). The linear model equation suggested by the software can be seen as follows:

$$Y_4 = 5.48 - 0.069X_1 - 0.75X_2$$

The equation shows that the pH was significantly (p-value < 0.0001) affected by gel former (X_2). Carbopol is a high molecular weight homopolymer and acrylic acid copolymer crosslinked with polyalkenyl polyethers (50). They are anionic and acidic (2.5-4.0 in 2% dispersion) when not neutralized with bases to achieve a specific viscosity (50,51). Therefore, adding Carbopol to a formulation with a fixed amount of base (triethanolamine) will significantly lower the pH of Negs. The response surface plot (**Figure 2D**) may then be used to depict the combined influence of variables X_1 and X_2 , which shows that Y_4 changes linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with Carbopol® 980 (X_2) - not span-80/tween-80 (X_1) - is the evidence from the comparative plot of the response surface. This description concludes that the choice of the X_2 level affects the pH Negs.

Spreadability was measured to ensure comfortable use on the skin because it spreads quickly (24). Terms of good dispersion are 5-7 cm. If the dispersion is too small, it is relatively difficult to spread when applied to the skin, while the dispersion tends to spread too quickly when applied, so it will cause an uncomfortable feeling when used (52). Based on the results, only one Neg did not meet the requirements, namely F11. **Table 1** shows the range of spreadability (Y_5) from 4.90 cm (STD#11) to 6.25 cm (STD#10). Based on **Table 2**, the spreadability response indicates a significant linear model with an F-value of 386.43 (p-value < 0.0001). The linear model equation suggested by the software can be seen as follows:

$$Y_5 = 5.58 + 0.53X_1 - 0.033X_2$$

The equation shows that the spreadability of Negs was significantly (p -value < 0.0001) positively affected by the emulsifier (X_1). That was also found in other literature studies (24). The higher the Carbopol-contained Negs, the more viscous Negs. AT and spreadability have the opposite results. The higher the Negs' viscosity, the higher the adhesive strength produced, while the smaller the dispersion power (53). The response surface plot (**Figure 2E**) depicts the combined influence of variables X_1 and X_2 , which shows that Y_5 changes linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with span-80/tween-80 (X_1) - not Carbopol® 980 (X_2) - is the evidence from the comparative plot of the response surface. This description concludes that the scatter can be changed by choosing the right X_1 level.

Topical dosage forms, such as Negs, adhere to the skin in two ways: they adhere directly to the rough surface to form a "mechanical interlock" and to the surface via interaction (54). Good adhesion to Negs supports a higher concentration gradient towards the skin and provides more drug penetration (18). Adhesive strength is directly related to the AT on the Negs as measured using the single-lap shear test method with slight modifications (55,56). The test was carried out by applying a shear load to the plates that flank the sample, which had been pre-loaded, and given a measured force; the time taken for the plates to separate was recorded as AT. In our study, the AT was from 4.22 s (STD#14) to 7.08 s (STD#13) in **Table 1**. They met the requirements based on the AT test results on a 14 formulation. An AT was carried out to see how long a Negs could be attached to the skin. The AT requirement is more than 4 seconds. The longer a Negs could be attached to the skin, it showed the better result, where it is expected that more active substances can be absorbed due to the time the Negs was in contact with the skin (57). Based on **Table 2**, the linear model was found to be significant in the AT response with an F-value model of 46.35 (p -value < 0.0001). The linear model equation suggested by the software can be seen as follows:

$$Y_6 = 5.74 - 0.20X_1 + 1.18X_2$$

The equation reveals that the AT was significantly (p -value < 0.0001) affected by the gel former (X_2) or, indirectly, the same adhesion strength. This finding is in agreement with the literature (58,59). A response surface plot (**Figure 2F**) may therefore be used to depict the combined influence of variables X_1 and X_2 , which shows that Y_6 changes linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with Carbopol® 980 (X_2) - not span-80/tween-80 (X_1) - is the evidence from the comparative plot of the response surface. From this explanation, it can be concluded that the AT can be changed by selecting the right X_2 level. Details of the ANOVA results for measured responses are also presented in **Table 2**. In the end, the emulsifier factor significantly affected the response of PS, ZP, and spreadability of TEO-loaded Negs. At the same time, gel former affects the AT and pH.

2.4. Optimized TEO-loaded Negs

The formulation was optimized with Design-Expert®, version 13 software. The optimized Negs were selected based on the minimum PS and PDI; maximum ZP; value in pH range, spreadability, and AT (**Table 3**). Variables composition for optimized Negs is span-80/tween-80 of 8.68% and Carbopol® 980 of 1.18% with a desirability value of 0.801. The formulation with the maximum desirability value is the optimal formulation generated from the optimization phase of the program (60) – the optimization value formed as indicated by the desirability value close to one.

The desirability value range is 0-1. **Figure 2** describes the optimization results in the form of a 2D contour. *Contour* is a two-dimensional response image that was presented using a predictive model for PS, PDI, ZP, pH, spreadability, and AT response values. The contour graph shows the desirability value of 0.801, which is the closest value to 1 compared to the other points. **Figure 3** shows the projection in the form of a 3D surface; the low area shows low desirability, while the high area shows high desirability and is getting closer to 1. At this stage, the software predicts the response values shown in **Table 4**. Three confirmation runs need to be performed to validate optimization (61). The optimization model and estimates are validated by the observed optimized Negs, which show an acceptable variation from the predicted values (**Table 4**). We tested the optimized Negs' physical properties for further investigation, such as organoleptic, homogeneity, freeze-thaw, viscosity, and flow properties.

2.5. Evaluation of TEO-loaded Negs

Nanoemulsion systems can cover oily drugs' bitter or unpleasant taste (62). The organoleptic results of TEO-loaded Negs have a less distinctive turmeric odor, which is white and semisolid (**Table 5**). That is due

239 to the drug entrainment of oil with the oil phase effectively preventing evaporation and masking its specific
240 odor (63). The homogeneity test results of the optimum TEO-loaded Negs formulation showed a
241 homogeneous preparation, as evidenced by the absence of coarse grains. This homogeneity was correlated
242 with the optimal formulation of PS and PDI of 182.3 ± 5.5 and 0.242 ± 0.003 , respectively. The low PDI
243 indicates uniformity or homogeneous dispersion of globules Negs (64). In addition, the small size of the
244 globule (± 200 nm) is not included in the coarse dispersion (41).

245 The thermodynamic stability test of the system was carried out using a freeze-thaw cycle to identify
246 the presence of metastable Negs in the optimal formulation. It aims to see the separation of the water and oil
247 phases due to the influence of extreme temperatures (65). The thermodynamic stability of any system is
248 determined by the change in free energy between the system and its surroundings (66). The test results on
249 the optimum formula for six cycles showed promising results; namely, there was no separation. This
250 stability was correlated with the ZP of the optimal formulation of 57.23 ± 2.91 mV. The surface charge's
251 magnitude was directly related to the stability of any Negs. It is evidenced by the high repulsive force
252 between the Negs globules preventing coalescence, which was characterized by the absence of phase
253 separation (67). Similar results were found in the Negs study containing thymoquinone, which had ZPs
254 between -26.7 and -30.6 mV (66).

255 The pH conditions indirectly affect the viscosity indicated by Negs because they influence the
256 swelling ability of Carbopol® 980. This excipient is a gel former and a thickener (52). It plays an essential role
257 in the viscosity of Negs. Carbopol is dispersed in water to form an acidic colloidal solution with a low
258 viscosity. Neutralizing with triethanolamine increases Negs' viscosity because a stable water-soluble gel was
259 formed (50). Viscosity was carried out with #7 spindle (Brookfield digital RV DV-E) at 50 rpm of $32240 \pm$
260 2257.7 cP, indicating significantly high viscosity on Negs with pH 4.5. The magnitude of the viscosity is
261 correlated with AT and spreadability. Viscosity is inversely proportional to spreadability (52). In contrast,
262 the AT is directly proportional to the viscosity. A high-viscosity system will form stronger interfacial
263 interactions and increase intermolecular interactions in the polymer network, increasing cohesion, adhesion
264 strength, and AT (54).

265 Determining the rheology of a semisolid preparation is essential for controlling the consistency
266 required to ensure the performance and formulation durability and to describe the mechanical (flow
267 properties) system (68). The rheological study was conducted in the shear rate range of 6.81 – 40.86 s⁻¹ at 25°C.
268 The consistency index equals the apparent viscosity at a shear rate of 1 s⁻¹. The consistency index measured
269 on TEO-loaded Negs was 155.67 cP and $n = 0.22$. The flow index measures the system's deviation from
270 Newtonian behavior ($n = 1$). A value of $n > 1$ indicates dilatation or shear thickening flow, and $n < 1$
271 indicates pseudoplastic or shear thinning. The flow index typically lowers the thicker the base. Negs
272 produce a 0.22 flow index, which implies pseudoplastic flow behavior. A colloidal network structure aligned
273 with the shear direction and decreases viscosity as the shear rate increases have led to this pseudo-plasticity.
274 The developed system will require a specific force to discharge (69). The results of the flow properties test
275 showed that the optimal formula made was a pseudoplastic thixotropic flow type (**Figure 4**). *Thixotropic* is a
276 flow property expected in pharmaceutical preparations because it has high consistency in the container but
277 can be poured and dispersed easily (70).

278 4. CONCLUSION

279 Based on the results of the CCD-RSM analysis, the optimum span-80/tween-80 as an emulsifier is
280 8.68%, and Carbopol® 980 as a gel former was 1.18%. The resulting response is a PS of 182.3 nm, PDI 0.242,
281 ZP 57.23 mV, pH 4.51, AT 6.45 seconds, and spreadability of 6 cm. Optimized formulation viscosity is 32240
282 cP with pseudoplastic thixotropic flow properties. Thus, the developed TEO-Negs can be a potential
283 delivery system and a promising suitable approach for topical preparations.

284 5. MATERIALS AND METHODS

285 5.1. Materials

286 TEO (*Curcuma longa*) was purchased from Darjeeling Sembrani Aroma (Indonesia), sorbilene O E/P from
287 Lamberti (Italy), span-80 from Croda (Singapore), propylene glycol from Dow Chemical Pacific (Singapore),
288 Carbopol® 980 NF from Lubrizol AM (Cleveland), nipagin M from Clariant Produkte (Deutschland),
289 propylparaben from Alpha Chemika (India), triethanolamine from Dow Chemical Pacific (Switzerland), and
290 1.1-diphenyl-2-picrylhydrazyl (DPPH) from Smart-Lab (Indonesia).

291 5.2. Methods

292 5.2.1. Identification of TEO

293 Organoleptic tests include observing form, color, and odor. Phenolic identification was carried out by
294 adding one drop of 5% FeCl₃ to a 1 ml sample. Dark green to black colors indicate phenolic compounds'
295 presence (71). Terpenoid testing was performed by adding Lieberman-Burchard reagent containing
296 anhydrous acetic acid and concentrated sulfuric acid (3:1) into a 1 ml sample. Brownish or violet ring form
297 indicates the presence of terpenoids (72). Spectrophotometry was used to determine antioxidant activity
298 with the DPPH method (5). The calibration curve for the DPPH concentration against absorbance was made
299 at a maximum wavelength of 516 nm. The absorbance was measured in a mixture of sample solution and
300 DPPH with a particular concentration after 30-min of incubation in a dark room. IC₅₀ was calculated from
301 the inhibition percentage and absorbance.

302 5.2.2. Preparation of TEO-loaded Negs

303 TEO-loaded Negs were produced using a high-energy method, which used a mechanical device to produce a
304 highly disruptive force to break up the water and oil phases to obtain nano-sized globules (18). The oil phase
305 (M1) was prepared by mixing span-80 with 5% turmeric oil using a magnetic stirrer (WiseStir Wisd) at 1,500
306 rpm for 20 min. A total of 0.18% methylparaben and 0.02% propylparaben dissolved in 15% propylene glycol
307 (M2). Then, the distilled water was stirred with tween-80, and M2 was added gradually until homogeneous
308 at 1,500 rpm for 20 min (M3). M1 was stirred with M3 until homogeneous at 1,500 rpm for 40 min to form a
309 clear and transparent nanoemulsion, then let left for 24 hours. The gel base was prepared by mixing
310 Carbopol® 980 NF with distilled water at 70°C and left for 24 hours. Then, gradually add 1% triethanolamine
311 to form a gel mass. Nanoemulsion was added slowly into the gel base while homogenized using a
312 homogenizer (AEG) at 2,000 rpm for 10 min.

313 5.2.3. Experimental Design

314 This study selects the CCD-RSM method to develop the TEO-loaded Negs formulation. For preliminary
315 screening on PS, PDI, ZP, pH, AT, and spreadability, a 2-factor CCD-RSM at two levels (high and low) was
316 used. Based on previous experiments and study literature, high and low variables were determined. The
317 CCD of the statistical package Design-Expert® version 13 software (Stat-Ease Inc., Minneapolis, MN) was
318 used to examine the influence of the specified independent variable on the response variable to obtain the
319 optimal formulation for TEO-loaded Negs. CCD planned 14 experiments were done under controlled
320 circumstances (Table 3). The independent variables were span-80/tween-80 (X₁) and Carbopol® 980
321 concentration (X₂). The observed response of the dependent variables was PS (Y₁), PDI (Y₂), ZP (Y₃), pH (Y₄),
322 spreadability (Y₅), and AT (Y₆).

323 5.2.4. Determination of PS, PDI, and ZP of Negs

324 PS and PDI of Negs were assessed using the Delsa Max Pro Particle Size Analyzer LS 100Q (Beckman
325 Coulter, USA) at 25°C utilizing the dynamic light scattering (DLS) method or photo correlation spectroscopy.
326 For analysis, 1 ml of samples was dispersed in 9 ml aqua pro injection. Into the cuvette, 1 ml of suspension
327 and 5 ml of aqua pro injection were added as a diluent, and the results were read on the instrument. All
328 measurements were made at a scattering angle of 90°. ZP was determined by particle size analyzer through
329 mobility and conductivity measurements. The temperature was set to 25°C, and the mean electric field was
330 set to 16 V/cm (73). The final result is the mean of each sample's three repeated measurements.

331 5.2.5. Determination of pH, Spreadability, and AT

332 pH was measured at a temperature of 25°C using a pH meter that had been previously calibrated with buffer
333 solutions of pH 4 and 7. The calibration step was completed when the pH value indicated on the screen
334 matched the correct pH value and was stable. Afterward, the electrode was dipped in Negs and recorded the
335 value shown on the screen (74). Spreadability was measured by adding 0.5 g of Negs in the center of a glass
336 covered with another glass. The preparation diameter was measured longitudinally and transversely; for

337 every minute added, 50 g was to a total weight of 150 g (75). Adhesion time determine by the single-lap
338 shear test method (54). A-0.5 g of Negs was placed on a slide, then covered with another slide, and given a
339 load of 1 kg for 3 min. The glass object was mounted on the test apparatus, and 80 g of the load was released
340 until both glass objects were released, and the time was recorded (75).

341 5.2.6. Organoleptic Observation and Homogeneity Test

342 Negs were placed in a glass object and directly observed for color, smell, and shape (76). A-0.1 g Negs were
343 spread over the slide, and homogeneity was observed. Test preparation is declared homogeneous if no
344 coarse grains exist (77).

345 5.2.7. Viscosity and Rheological Flow

346 Viscosity and rheology were determined with a Brookfield RV DV-E Viscometer with appropriate spindle
347 and speed. A-500 ml of Negs were put into a beaker glass; the spindle was installed, and the measured value
348 was recorded as viscosity Negs. In this study, the spindle used spindle no. 7. Flow properties were
349 determined by measuring the viscosity using a right spindle from low to high rotational speed and vice
350 versa (74). Flow index and consistency index are determined from the power law equation:

$$351 \tau = K r^n$$

352 where τ is the shear stress, K is the consistency index, r is the shear rate, and n is the flow index.

353 Taking logs on both sides,

$$354 \log \tau = \log K + n \log r$$

355 So, from the log, shear stress Vs. Log shear rate plot, the plot slope was used as the flow index and the
356 antilog of the Y-intercept as the consistency index (69).

357 5.2.8. Freeze-thaw Test

358 Negs were stored at 4 ± 2 °C, then transferred to 40 ± 2 °C for 48 hours (1-cycle), then repeated for 6-cycles.
359 Phase separation was observed in each cycle (74).

360 5.2.9. Statistical Analysis

361 The Design-Expert® version 13 software was used to conduct the statistical study (Stat-Ease Inc.,
362 Minneapolis, MN). Analysis was done at a sig – $p < 0.05$ and $p < 0.01$ after three times of each measurement.

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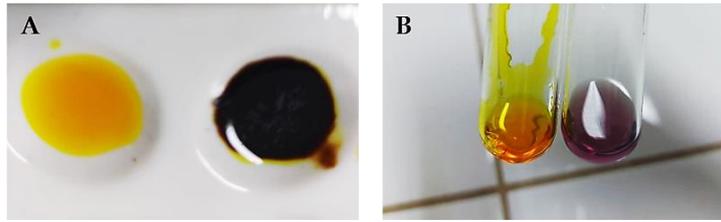
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541

542 **Figure 1.** The result of qualitative test observation is a phenolic test with a positive result marked in a blackish green
543 color (a) and a terpenoid test with a positive result marked with a purple color (b).

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546

547 **Table 1.** Evaluation results of independent variables and dependent variables with the CCD design for optimizing TEO-
548 loaded Negs.

Std	X ₁ (%)	X ₂ (%)	Y ₁ (nm)	Y ₂	Y ₃ (mV)	Y ₄	Y ₅ (cm)	Y ₆ (s)
1	5.50	1.02	206.9 ± 0.7	0.571 ± 0.000	35.39 ± 1.50	5.45	5.55	5.65
2	2.32	1.15	457.0 ± 0.9	0.000 ± 0.000	13.90 ± 1.92	4.60	5.00	6.99
3	8.68	0.90	166.7 ± 0.4	0.378 ± 0.074	31.64 ± 0.53	6.35	6.20	4.23
4	2.32	0.90	411.0 ± 5.8	0.000 ± 0.000	30.67 ± 0.89	6.29	5.00	4.30
5	5.50	1.02	302.1 ± 3.6	0.571 ± 0.000	36.45 ± 0.45	5.46	5.60	5.63
6	5.50	1.02	263.0 ± 3.7	0.571 ± 0.000	18.49 ± 1.77	5.47	5.55	5.64
7	8.68	1.15	198.1 ± 1.2	0.242 ± 0.003	57.23 ± 2.91	4.65	6.15	6.95
8	5.50	1.02	246.9 ± 7.2	0.571 ± 0.000	26.60 ± 0.17	5.46	5.60	5.67
9	5.50	1.02	206.0 ± 2.3	0.571 ± 0.000	31.77 ± 1.01	5.46	5.60	5.65
10	10.00	1.02	160.8 ± 1.3	0.285 ± 0.024	46.72 ± 1.97	5.31	6.25	5.84
11	1.00	1.02	244.2 ± 4.3	0.571 ± 0.000	31.26 ± 0.57	5.78	4.90	6.88
12	5.50	1.02	238.1 ± 3.0	0.571 ± 0.000	30.78 ± 0.53	5.45	5.55	5.64
13	5.50	1.20	206.2 ± 1.6	0.571 ± 0.000	30.56 ± 0.78	4.57	5.50	7.08
14	5.50	0.85	240.5 ± 2.2	0.571 ± 0.000	20.12 ± 0.16	6.39	5.65	4.22

Y₁: Particle size (PS), Y₂: Polydispersity index (PDI), Y₃: Zeta potensial (ZP), Y₄: pH, Y₅: Spreadability, Y₆: Adhesion time (AT)

549
550

551 **Table 2.** Statistical analysis of PS (Y₁), PDI (Y₂), ZP (Y₃), pH (Y₄), spreadability (Y₅), and AT (Y₆) TEO-loaded Negs on
552 CCD.

Factors		Y ₁	Y ₂	Y ₃	Y ₄	Y ₅	Y ₆
X ₁	Coefficient	-77.64	0.027	8.27	-0.069	0.53	-0.20
	p-value	0.0063**	0.7325	0.0080**	0.1430	<0.0001**	0.1406
X ₂	Coefficient	3.61	-0.017	2.95	-0.75	-0.033	1.18
	p-value	0.8783	0.8289	0.2668	<0.0001**	0.1157	<0.0001**
X ₁ X ₂	Coefficient		-0.034	10.59			
	p-value		0.7602	0.0136*			
X ₁ ²	Coefficient		-0.16				
	p-value		0.0818				
X ₂ ²	Coefficient		-0.086				
	p-value		0.3087				
Intercept	Coefficient	253.39	0.57	31.54	5.48	5.58	5.74
Degree of freedom		2	5	3	2	2	2
Sum of squares		48332.05	0.24	1065.33	4.48	2.28	11.49
Mean of squares		24166.02	0.047	355.11	2.24	1.14	5.74
F-value		5.69	1.02	7.06	145.08	386.43	46.35
p-value		0.0201	0.4645	0.0078	<0.0001	<0.0001	<0.0001
R-Squared		0.5084	0.3898	0.6794	0.9635	0.9860	0.8939

553 X₁: span-80/tween-80; X₂: Carbopol® 980

554 * p-value < 0.05

555 ** p-value < 0.01

556

557

558 **Table 3.** The independent and dependent variables with levels and limits in CCD for TEO-loaded Negs development.

Variables	Code	Start point (- α)	Low level (-1)	Central level	High level (+1)	Start point (+ α)	Units
Independent variables							
Span-80/tween-80	X ₁	0.85	2.32	1.02	5.50	1.20	% w/w
Carbopol® 980	X ₂	1	0.90	5.50	1.02	10	% w/w
Dependent variables							
Limits							
Particle size (PZ)	Y ₁	Minimum					nm
Polydispersity index (PDI)	Y ₂	Minimum					
Zeta potensial (ZP)	Y ₃	Maximum					mV
pH	Y ₄	is in range					
Spreadability	Y ₅	is in range					cm
Adhesion time (AT)	Y ₆	is in range					s

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Table 4. Optimized TEO-loaded Negs' actual and predicted values for each response.

Responses	Predicted values	Actual value ^a	Error ^b (%)
Y1 (nm)	180.2	182.3 ± 5.5	1.165
Y2	0.250	0.242 ± 0.003	-3.200
Y3 (mV)	56.30	57.23 ± 2.91	1.652
Y4	4.50	4.51 ± 0.02	0.222
Y5 (cm)	6.07	6.0 ± 0.2	-1.153
Y6 (s)	6.98	6.45 ± 0.19	-7.593

a Data listed is the mean ± standart deviation, n = 3

b Error (%) = [(Actual value - Predicted value)/Predicted value] * 100%

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Table 5. Additional evaluation on TEO-loaded Negs with an optimal formulation.

Evaluation	Result
Organoleptic	Color: White; Odor: typical turmeric
Homogeneity	Homogeneous
Freeze-Thaw	Cycles 0-6, no separation occurs
Viscosity ^a	32240 ± 2257,7 cP

568

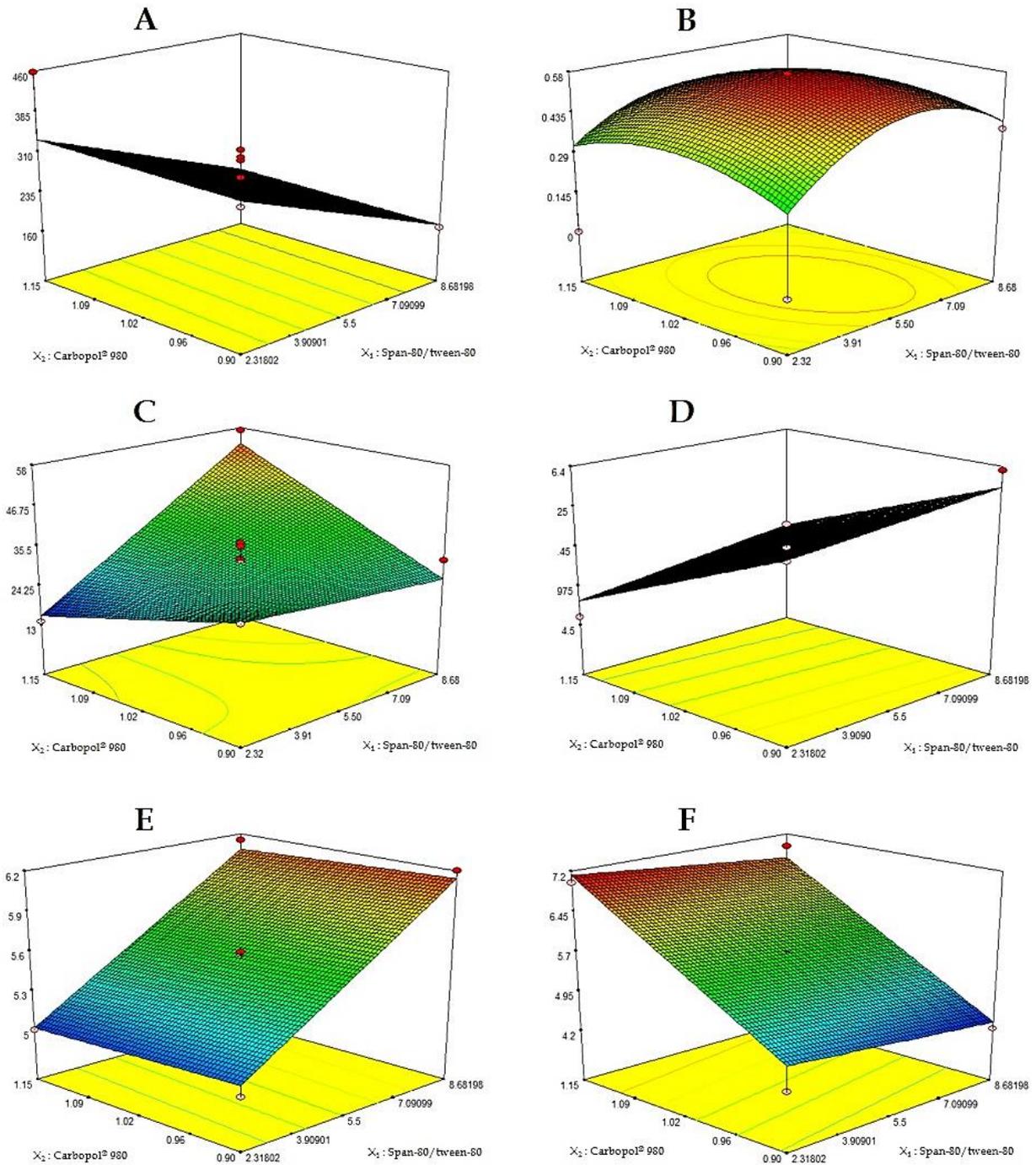
^a Data listed is the mean ± standart deviation, n = 3

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Figure 2. Effect of span-80/tween-80 and Carbopol[®] 980 concentration on PS (a) PDI (b) ZP (c) pH (d) spreadability (e) AT (f).

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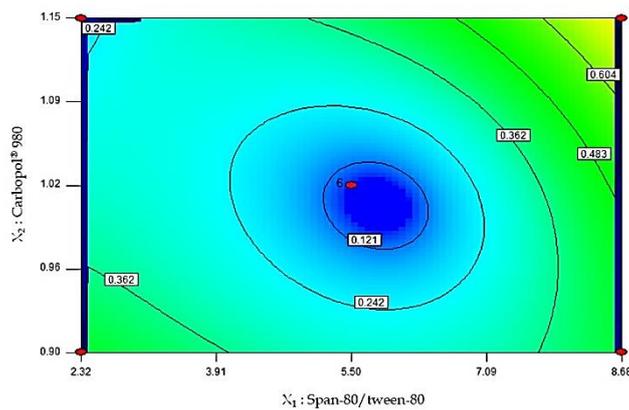


Figure 3. Contour plot desirability value of optimal formulation.

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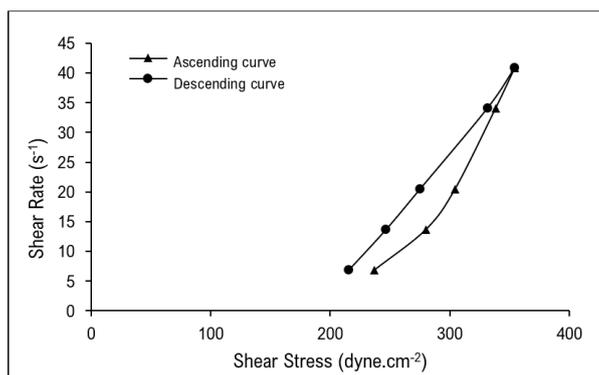


Figure 4. Optimal formulation flow properties.

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RESPONSE TO REVIEWERS

N. Nining
Universitas Muhammadiyah Prof. DR. HAMKA
East Jakarta
Indonesia
nining@uhamka.ac.id

November 23, 2022

Dear Reviewers:

We appreciate the invaluable review and suggestions from the reviewers to improve our research results. We will respond to several points related to that.

1. We apologize for our poor writing. Improvements have been made with the help of the Language editing service.
2. We have regenerated the abbreviations section, italics, and usage of appropriate terms and units.
3. This study was conducted based on the adequate TEO availability in our country, and several research reports have promising results on its pharmacology activity.
4. This report demonstrates the physical and mechanical characterization of the Negs, while quantitative analysis is a limitation due to limited research funding. We hope to continue this in further research.
5. Several details provide to clarify the CCD design.
6. Results and discussion enriched with sufficient literature.

We hope the revised and updated articles can be published in the Journal of Research in Pharmacy.

We appreciate your consideration.

Sincerely,
Nining apt., M.Si
Lecturer, Faculty of Pharmacy and Science
Universitas Muhammadiyah Prof. DR. HAMKA

**4. Bukti Konfirmasi Review dan Hasil Review
Kedua (27 Januari 2023)**

Journal of Research in Pharmacy : Revision request for your manuscript

1 message

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Fri, Jan 27, 2023 at 6:10 PM

To: nining@uhamka.ac.id

Journal of Research in Pharmacy

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The Editors have now assessed the reviewer response and have concluded that, in its present form, the manuscript is not yet ready for publication in the Journal. Below you will find the relevant review comments and editorial notes. Acceptance of the paper is contingent upon effectively revising the work by taking these comments into serious consideration, and by responding or rebutting them in detail. We ask you to submit your revision through the online system within 30 days. The web site is located at <https://www.jrespharm.com>.

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Suggestions

1. Reviewer Comments

The authors' response to my previous comments was presented as a separate file without referring to the comments and it is impossible to follow the updated manuscript file since the changes made are not traceable. Under these circumstances the manuscript is not acceptable.

2. Reviewer Comments

This revised paper is possibly suitable to be published in "Journal of Research in Pharmacy".

Manuscript Information

Manuscript ID: MPJ-10622.REV-1

Title in English: Response surface methodology for optimization of turmeric essential oil-loaded nanoemulgel

Small Title in English: No information entered

Authors: Nining Nining¹, Anisa Amalia¹, Fatimatuz Zahrok²

Institutions: ¹Universitas Muhammadiyah Prof. DR. HAMKA, Pharmaceutical Technology, East Jakarta, Indonesia

²Universitas Muhammadiyah Prof. DR. HAMKA, Pharmacy, East Jakarta, Indonesia

Keywords in English: Central composite design; nanoemulgels; response surface methodology; turmeric essential oil; topical delivery.

Manuscript Type: Research article

Processing Status: Major Revision

Abstract in English

Turmeric Essential Oil (TEO) has an antioxidant and anti-inflammatory activity to be formulated in a topical dosage form. Nanoemulgels (Negs) development, based on varying concentrations of emulsifiers and gel formers, affects their characteristics and stability. This study focuses on optimizing TEO-loaded Negs based on physical and mechanical characterization, which have promising topical applications. Negs were created using the high-energy approach and optimized using Response Surface Methodology (RSM) and the Central-Composite Design (CCD) for the optimization of span-80/tween-80 (X1) and Carbopol® 980. (X2). Observed variable responses were particle size (PS) (Y1), polydispersity index (PDI) (Y2), zeta potential (ZP) (Y3), pH (Y4), spreadability (Y5), and adhesion time (AT) (Y6). Actual responses of Negs were compared with the CCD-RSM predictions to validate the model. In addition, other physical evaluations were observed, such as organoleptic observations, homogeneity, freeze-thaw tests, viscosity, and flow properties. Optimized TEO-loaded Negs were made with 8.68% span-80/tween-80 and 1.18% Carbopol® 980. The evaluation results showed the optimal TEO-loaded Negs on nano-metric size (182.3 ± 5.5 nm) with low PDI (0.242 ± 0.003), good ZP (-57.23 ± 2.91 mV), pH (4.51 ± 0.02), spreadability (6.0 ± 0.2 cm), and AT (6.45 ± 0.19 s). TEO-loaded Negs have an excellent appearance and did not run phase separation at extreme temperature storage with pseudoplastic thixotropy flow. Thus, the developed TEO-loaded Negs can be a potential delivery system and a promising suitable approach for topical preparations.

Manuscript Files

File Name	File Size	Date Created	Category	Description
MPJ-10622-1-cover-letter-nining-et-al..pdf (../pdf-files/out/11427-MPJ-10622-1-cover-letter-nining-et-al..pdf)	31 KB	Sep 03, 2022	Cover letter	None
MPJ-10622-9-isi-jrp-checklist.pdf (../pdf-files/in/11427-MPJ-10622-9-isi-jrp-checklist.pdf)	3388 KB	Sep 03, 2022	Author Checklist Form	None
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MPJ-10622-4-manuscript-nining-et-al..pdf (../pdf-files/out/11427-MPJ-10622-4-manuscript-nining-et-al..pdf)	818 KB	Sep 03, 2022	Main Document	None
MPJ-10622-7-figure-1..jpeg (../pdf-files/in/11427-MPJ-10622-7-figure-1..jpeg)	513 KB	Sep 03, 2022	Figure	None
MPJ-10622-9-figure-2..jpeg (../pdf-files/in/11427-MPJ-10622-9-figure-2..jpeg)	190 KB	Sep 03, 2022	Figure	None
MPJ-11427-3-manuscript-nining-et-al.-rev01-proofread.rev-1.pdf (../pdf-files/out/11427-MPJ-11427-3-manuscript-nining-et-al.-rev01-proofread.rev-1.pdf)	2042 KB	Nov 23, 2022	Main Document	None
MPJ-11427-8-figure-1..rev-1.jpg (../pdf-files/in/11427-MPJ-11427-8-figure-1..rev-1.jpg)	26 KB	Nov 23, 2022	Figure	None
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MPJ-11427-3-response-to-reviewers.rev-1.pdf (../pdf-files/out/11427-MPJ-11427-3-response-to-reviewers.rev-1.pdf)	32 KB	Nov 23, 2022	Response to Reviewers	None
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Are the figures tables and graphics clear ?

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about the approval of
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presented ?

Not applicable

**5. Bukti Konfirmasi Review Submit Revisi Kedua,
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Journal of Research in Pharmacy : Confirmation for revised manuscript

1 message

Journal of Research in Pharmacy <editor@marmarapharmj.com>
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Mon, Feb 13, 2023 at 1:46 PM

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This is to acknowledge receipt of your revised manuscript entitled Response surface methodology for optimization of turmeric essential oil-loaded nanoemulgel. You can follow the stage of your manuscript in the review process through the author center. Thank you.

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Response surface methodology for optimization of turmeric essential oil-loaded nanoemulgel

Nining NINING^{1*} , Anisa AMALIA¹ , Fatimatuz ZAHROK² 

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² Department of Pharmacy, Faculty of Pharmacy and Science, Universitas Muhammadiyah Prof. DR. Hamka, Jakarta, Indonesia.

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ABSTRACT: Turmeric Essential Oil (TEO) has an antioxidant and anti-inflammatory activity to be formulated in a topical dosage form. Nanoemulgels (Negs) development, based on varying concentrations of emulsifiers and gel formers, affects their characteristics and stability. This study focuses on optimizing TEO-loaded Negs based on physical and mechanical characterization, which have promising topical applications. Negs were created using the high-energy approach and optimized using Response Surface Methodology (RSM) and the Central-Composite Design (CCD) for the optimization of span-80/tween-80 (X_1) and Carbopol® 980. (X_2). Observed variable responses were particle size (PS) (Y_1), polydispersity index (PDI) (Y_2), zeta potential (ZP) (Y_3), pH (Y_4), spreadability (Y_5), and adhesion time (AT) (Y_6). Actual responses of Negs were compared with the CCD-RSM predictions to validate the model. In addition, other physical evaluations were observed, such as organoleptic observations, homogeneity, freeze-thaw tests, viscosity, and flow properties. Optimized TEO-loaded Negs were made with 8.68% span-80/tween-80 and 1.18% Carbopol® 980. The evaluation results showed the optimal TEO-loaded Negs on nano-metric size (182.3 ± 5.5 nm) with low PDI (0.242 ± 0.003), good ZP (-57.23 ± 2.91 mV), pH (4.51 ± 0.02), spreadability (6.0 ± 0.2 cm), and AT (6.45 ± 0.19 s). TEO-loaded Negs have an excellent appearance and did not run phase separation at extreme temperature storage with pseudoplastic thixotropy flow. Thus, the developed TEO-loaded Negs can be a potential delivery system and a promising suitable approach for topical preparations.

KEYWORDS: Central composite design; nanoemulgels; response surface methodology; turmeric essential oil; topical delivery.

1. INTRODUCTION

Turmeric is the dried rhizome of *Curcuma longa* L. (Zingiberaceae), which derives from Southeast Asia and is cultivated mainly in India, followed by Bangladesh, China, Thailand, Cambodia, Indonesia, Malaysia, and the Philippines (1). Steam distillation extracts turmeric essential oil (TEO) from the turmeric rhizome (2). Chemical constituents with the most significant proportion were oxygenated monoterpenes and sesquiterpenes, which include β -turmerone, α -turmerone, and ar-turmerone (3–5). The pharmacological activities of TEO have been reported in the form of antioxidants, anti-inflammatory, antinociceptive, antidermatophytic, antifungal, and antibacterial activities (2,6–9). These reducing power and radical scavenging abilities are associated with the high antioxidant potential of TEO (8,10). This pharmacological activity justifies its use in various applications, including cosmetics and phytomedicines (7,11,12). Like other essential oil, TEO has limited use due to its volatility, instability under certain conditions, lipophilicity, and low aqueous solubility (13,14). Many recent studies are oriented toward solving these limitations, so that efficacy of the essential oil lasts longer and increases.

Previously, TEO has been developed for cream as a conventional drug delivery system, patch, and nanoemulsions (7,11,15,16). Most water-based liquid or semisolid systems have limitations in delivering lipophilic drugs (17). Nanoemulsions are an established alternative for delivering lipophilic drugs by increasing topical absorption (18). The main advantage of topically administered nanoemulsions is the ability to increase penetration and permeation of drugs through the skin without adding non-physical enhancers and non-friendly solvents to the formulation, which can cause skin irritation, especially with long-term usage (18,19). However, this system has problems with low viscosity due to poor spreadability and skin

How to cite this article: Surname N, Surname N. Title of the manuscript. J Res Pharm. 2019; 23(6): 1-XX.

33 retention (20). **Nanoemulgels (Negs)**, a combination of nanoemulsion and hydrogel, were made to improve
34 the characteristics (18,19,21).

35 Negs consist of a hydrogel system and an emulsion with nano-sized globules. An emulsifier in the
36 form of surfactants and co-surfactants stabilizes the emulsion, which serves as a drug delivery platform (18).
37 Surfactants reduce the interfacial surface tension of immiscible liquids and change the entropy of the
38 dispersion, thereby stabilizing a thermodynamically unstable system; co-surfactants are combined with
39 surfactants in the emulsification process by disrupting the surface layer (22). The emulsifier plays a role in
40 the emulsification process to increase stability when the product is kept for an extended time. On the other
41 side, gels are made from polymers, a gel former, that expand after absorbing a liquid (23). Gel former
42 increase the viscosity of the formulation and can react with emulsifiers to change the thickness (24). In
43 addition, topical Negs can increase patient compliance because of their non-irritating, non-greasy
44 characteristics and improved drug release (25). However, the available methods for manufacturing Negs
45 exhibit various limitations, which directly or indirectly affect the quality of the Negs formulations.

46 Currently, the principle of quality by design is adopted to ensure the quality of drugs, their safety, and
47 efficacy (26). The quality by design (QbD) trend is used to develop, optimize, and investigate the interaction
48 between particular variables and their related responses to achieve the optimal formulation (27). Our study
49 used Central Composite Design (CCD) on Response Surface Methodology (RSM) to select optimal
50 formulations and predict models that rely on statistical analysis (ANOVA) and exact equations (28,29). In
51 this aspect, we intended to develop a TEO-loaded Negs that would be optimized using a complete 2²-
52 factorial design and determine the independent factors' precise influence on the investigated dependent
53 variables. The choice and procedure factors were span-80/tween-80 (X₁) and Carbopol® 980 (X₂). Response
54 variables investigated were particle size (PS) (Y1), polydispersity index (PDI) (Y2), zeta potential (ZP) (Y3),
55 pH (Y4), spreadability (Y5), and adhesion time AT) (Y6). This work is the first step in developing an
56 optimized TEO preparation suitable for transdermal drug delivery for topical application.

57 2. RESULTS AND DISCUSSION

58 2.1. Identification of TEO

59 TEO identification is carried out to ensure the quality of the active ingredients. The tests included
60 organoleptic, phytochemical identity in the form of phenolics and terpenoids, and antioxidant activity by the
61 DPPH method. The results of this study can be seen in **Figure 1**. Organoleptically, TEO is an orange liquid
62 with a distinctive turmeric aroma. Identification results in the phenolic test showed a black-green color
63 (**Figure 1A**) due to the formation of a phenolic-Fe³⁺ complex (30). In the terpenoid test, reaction results show
64 a purple color (**Figure 1B**) as a positive sign. Based on these results, it can be concluded that TEO contains
65 phenolic and terpenoid compounds. Several studies reported that TEO has powerful antioxidant potential
66 both *in-vitro* and *in-vivo* (2,8,31). Oxygenated monoterpenes and sesquiterpenes, such as α -turmerone,
67 α -turmerone, and β -turmerone, were included in the terpenoid compounds and reported to play a role in this
68 activity. In addition, contained phenol was also reported to perform as an excellent free radical scavenger
69 because its reduction potential was lower than oxygen's (8). In our study, the antioxidant activity of TEO
70 was presented as an IC₅₀ of 9.88 μ m/ml, which was included in the powerful antioxidant category because of
71 below 50 μ m/ml (32). It differed from other studies' results, which showed an IC₅₀ of 2274.02 μ m/ml and
72 above 1000 μ m/ml (2,10). Meanwhile, similar findings stated IC₅₀ values of 10.03 μ m/ml, 3.227 μ m/ml, and
73 14.5 μ m/ml (33–35). The difference value was due to the use of TEO from different origin sources, regional
74 conditions, and the extraction method used.

75 2.2. Preparation of TEO-loaded Negs

76 TEO-loaded Negs were produced using the high-energy method efficiently. A high-pressure
77 homogenizer provides sufficient energy to increase the interfacial area and generate nano-size globules (36).
78 An emulsion system with a nanodroplet size must have flow properties that allow it to pass through the
79 homogenizer (37). Flow properties are inversely correlated with the amount of emulsifier and gel former
80 added to the Negs formulation. In this study, the selected optimized formulation was the one that produces
81 Negs with the smallest size, as shown by the limitations in **Table 3**. Therefore, the range of regulated
82 emulsifiers and gel formers concentrations still provides flow properties that allow the flow to remain good
83 using a high-energy homogenizer. In general, the preparation of Negs was carried out in 3-stages, namely,

84 nanoemulsions preparation, gel preparation, and Negs preparation. Negs were successfully produced in 14-
85 formulation with variations of emulsifiers and gel former with compositions that can be seen in **Table 1**.

86 2.3. Experimental Design by CCD-RSM

87 2.3.1. Fitting the Model

88 TEO-loaded Negs were optimized based on CCD in the RSM. CCD was used to establish the optimal
89 concentrations of emulsifier (X_1) and gel former (X_2) as the key parameters influencing the dependent
90 response. **Prediction of the factorial axial design and the possible curvature in the response could be**
91 **obtained from the optimization process with an effective second-level design (38).** Predicting the significant
92 influence of the independent variable on the dependent variable is essential for generating TEO-loaded
93 Negs. Based on the literature survey, two factors were chosen as independent variables, and six responses
94 were chosen as the dependent variable with the most significant influence on Negs. The independent
95 variables with their levels and the observed response variables are presented in **Table 1**.

96 2.3.2. Analysis of Design

97 Statistical data analysis must be carried out to predict and recognize the model. **Table 2** shows the
98 statistical analysis of a quadratic model for PDI, a 2FI model for ZP, and linear models for PS, pH, AT, and
99 spreadability. This table identifies factors with p-values less than a predefined threshold (0.01 and 0.05, with
100 a 99% and 95% confidence level, respectively) as influential factors. **Besides the significant p-value, a large F-**
101 **value minimizes error in the model and lack of fit, preferably non-significant to fit the model data (27,39).**
102 Depending on the most significant R-squared value and the least residual predictive sum of squares value,
103 the six responses demonstrated distinct models in their application. The chosen model had a non-statistically
104 significant lack of fit, and model validation was confirmed by the residual plot test of the regression model,
105 which was supported by supplemental information for all responses. Compared to the 2FI and simple linear
106 models, the quadratic model represented the impacts of numerous variables, including individual factors,
107 interactions, and the quadratic influence on the response.

108 2.3.3. Effect of Independent Variables on Dependent Variables

109 PS parameters are often used to characterize nanoparticles. Negs globule diameter means (Y_1) was
110 adjusted from 160.8 nm (STD#10) to 457.0 nm (STD#2). Based on **Table 2**, the PS response indicates a
111 significantly linear model with an F-value of 5.69 (p-value 0.0201 < 0.05). The suggested linear model
112 equation can be seen as follows:

$$113 Y_1 = 253.39 - 77.64X_1 + 3.61X_2$$

114 The equation shows that the Negs PS is significantly (p-value < 0.01) affected by emulsifier
115 concentration (X_1). The positive coefficient has a synergistic effect on the response. **In contrast, the negative**
116 **coefficient has an antagonistic effect which concludes the inverse relationship of the independent variable**
117 **with a response (38,40).** In addition, this factor has a more significant coefficient. It directly affects the PS,
118 which means that increasing emulsifier concentration causes a decrease in the globule diameter PS in the
119 TEO-loaded Negs. High emulsifier (above 5.5%) resulted in globules measuring below 200 nm, and low
120 emulsifier (below 5.5%) produced globules above 200 nm. **This fact is in line with other studies that**
121 **increasing emulsifiers could reduce droplet Negs PS (41,42).** The emulsifier reduced the interfacial surface
122 tension between the water and oil phases, which decreased the free energy required to disrupt or break the
123 globules and resulted in a smaller droplet diameter. It can also produce a protective cover over the globules,
124 preventing them from coalescence. However, the emulsifier must absorb quickly enough around the droplet
125 to form this protective layer (42,43).

126 Meanwhile, Carbopol® 980 had no significant effect on PS. **The same findings were obtained from the**
127 **globule size Negs from Carbopol® 934 and 940 as gel former (19).** A response surface plot (**Figure 2A**) may
128 therefore be used to represent the combined influence of variables X_1 and X_2 , which shows that Y_1 changes
129 linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with
130 span-80/tween-80 (X_1) – not Carbopol® 980 (X_2) – was evidenced from the comparative plot of the response
131 surface. From this explanation, it can be concluded that the PS can be changed by selecting the right X_1 level.

132 In **Table 1**, Negs PDI (Y_2) varied from 0.000 (STD#2 and #4) to 0.571 (STD#1, #5, #6, #8, #9, #11 to
133 #14). PDI measures the distribution of molecular mass in a sample. The smaller PDI (close to 0), the more
134 stable the Negs formulation caused; the large PDI indicates particles formed are not uniform, and the
135 formulation will flocculate quickly. **An index value less than 0.05 is included in monodisperse, while an**
136 **index greater than 0.7 indicates that the sample has a broad PS distribution. A-0.2 and below are considered**
137 **acceptable for nanoparticle preparations (44).** Based on **Table 2**, the PDI response shows a non-significant
138 quadratic model with an F-value of 1.02 (p-value 0.4645 > 0.05). Measurements between variables and
139 responses are not a precise cause of that. The suggested quadratic model equation can be seen as follows:

$$Y_2 = 0.57 + 0.027X_1 - 0.017X_2 - 0.034X_1X_2 - 0.16X_1^2 - 0.086X_2^2$$

140
141 The equation and p-value of each factor did not significantly affect PDI. The response surface plot
142 (**Figure 2B**) depicts the combined influence of variables X_1 and X_2 , which shows that Y_2 changes with the
143 sum of the two variables by model.

144 The Negs ZP (Y_3) was in the range of 13.90 mV (STD#2) to 46.72 mV (STD#10) (**Table 1**). **ZP**
145 **represents the electric charge between the shear plane of a final outer layer and bulk solution, which**
146 **significantly affects dispersion stability (45).** This factor is strongly influenced by the composition of the
147 Negs and its electrical phenomena. TEO-loaded Negs, **which have positive ZP, show good interaction with**
148 **negatively charged skin (46).** ZP is the scientific term for the electrokinetic potential in colloidal systems. The
149 high electric charge on the nanoparticle surface will prevent nanoparticle aggregation because of the strong
150 repulsion between particles. **The ZP requirement for stability is above ± 30 mV (19). The higher ZP and the**
151 **slower aggregation are formed to prevent separation (47).** Based on **Table 2**, the ZP response indicates a
152 significant 2FI model with an F-value model of 7.06 (p-value 0.0078 < 0.05). The linear model equation
153 suggested by the software can be seen as follows:

$$Y_3 = 31.54 + 8.27X_1 + 2.95X_2 + 10.59X_1X_2$$

154
155 The equation shows that the Negs ZP was significantly affected by the emulsifier (X_1) and gel former-
156 emulsifier interaction (X_1X_2). **ZP was usually influenced by the physicochemical properties of the drug,**
157 **polymer, carrier, electrolyte presence, and their adsorption (48).** One study stated that adding Carbopol®
158 only slightly increased the ZP Negs (19). The response surface plot (**Figure 2C**) depicts the combined
159 influence of variables X_1 and X_2 , showing that Y_3 changes with the sum of the two variables. Nevertheless,
160 the higher gradient in the response surface with span-80/tween-80 (X_1) - not Carbopol® 980 (X_2) - is the
161 evidence from the comparative plot of the response surface. This description concludes that the ZP can be
162 changed by selecting the proper X_1 level.

163 The pH test was carried out to measure Negs's acidity or alkalinity level. The pH values (Y_4) were in
164 the range of 4.57 (STD#13) to 6.39 (STD#14) (**Table 1**). The pH requirement of Negs is the same as the skin
165 pH. Too-acidic preparations can irritate the skin and cause a stinging sensation, while too-alkaline
166 preparations can cause dry and itchy skin. The pH result test on 14 formulations was eligible and compatible
167 with the skin (4.5-6.5) (49). Based on **Table 2**, a linear model was found to be significant in pH response with
168 a model F-value of 145.08 (p-value < 0.0001). The linear model equation suggested by the software can be
169 seen as follows:

$$Y_4 = 5.48 - 0.069X_1 - 0.75X_2$$

170
171 The equation shows that the pH was significantly (p-value < 0.0001) affected by gel former (X_2).
172 **Carbopol is a high molecular weight homopolymer and acrylic acid copolymer crosslinked with polyalkenyl**
173 **polyethers (50). They are anionic and acidic (2.5-4.0 in 2% dispersion) when not neutralized with bases to**
174 **achieve a specific viscosity (50,51).** Therefore, adding Carbopol to a formulation with a fixed amount of base
175 (triethanolamine) will significantly lower the pH of Negs. The response surface plot (**Figure 2D**) may then be
176 used to depict the combined influence of variables X_1 and X_2 , which shows that Y_4 changes linearly with the
177 sum of the two variables. Nevertheless, the higher gradient in the response surface with Carbopol® 980 (X_2) -
178 not span-80/tween-80 (X_1) - is the evidence from the comparative plot of the response surface. This
179 description concludes that the choice of the X_2 level affects the pH Negs.

180 **Spreadability was measured to ensure comfortable use on the skin because it spreads quickly (24).**
181 **Terms of good dispersion are 5-7 cm. If the dispersion is too small, it is relatively difficult to spread when**
182 **applied to the skin, while the dispersion tends to spread too quickly when applied, so it will cause an**
183 **uncomfortable feeling when used (52).** Based on the results, only one Neg did not meet the requirements,
184 namely F11. **Table 1** shows the range of spreadability (Y_5) from 4.90 cm (STD#11) to 6.25 cm (STD#10).
185 Based on **Table 2**, the spreadability response indicates a significant linear model with an F-value of 386.43
186 (p-value < 0.0001). The linear model equation suggested by the software can be seen as follows:

$$Y_5 = 5.58 + 0.53X_1 - 0.033X_2$$

The equation shows that the spreadability of Negs was significantly (p -value < 0.0001) positively affected by the emulsifier (X_1). That was also found in other literature studies (24). The higher the Carbopol-contained Negs, the more viscous Negs. AT and spreadability have the opposite results. The higher the Negs' viscosity, the higher the adhesive strength produced, while the smaller the dispersion power (53). The response surface plot (Figure 2E) depicts the combined influence of variables X_1 and X_2 , which shows that Y_5 changes linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with span-80/tween-80 (X_1) - not Carbopol® 980 (X_2) - is the evidence from the comparative plot of the response surface. This description concludes that the scatter can be changed by choosing the right X_1 level.

Topical dosage forms, such as Negs, adhere to the skin in two ways: they adhere directly to the rough surface to form a "mechanical interlock" and to the surface via interaction (54). Good adhesion to Negs supports a higher concentration gradient towards the skin and provides more drug penetration (18). Adhesive strength is directly related to the AT on the Negs as measured using the single-lap shear test method with slight modifications (55,56). The test was carried out by applying a shear load to the plates that flank the sample, which had been pre-loaded, and given a measured force; the time taken for the plates to separate was recorded as AT. In our study, the AT was from 4.22 s (STD#14) to 7.08 s (STD#13) in Table 1. They met the requirements based on the AT test results on a 14 formulation. An AT was carried out to see how long a Negs could be attached to the skin. The AT requirement is more than 4 seconds. The longer a Negs could be attached to the skin, it showed the better result, where it is expected that more active substances can be absorbed due to the time the Negs was in contact with the skin (57). Based on Table 2, the linear model was found to be significant in the AT response with an F-value model of 46.35 (p -value < 0.0001). The linear model equation suggested by the software can be seen as follows:

$$Y_6 = 5.74 - 0.20X_1 + 1.18X_2$$

The equation reveals that the AT was significantly (p -value < 0.0001) affected by the gel former (X_2) or, indirectly, the same adhesion strength. This finding is in agreement with the literature (58,59). A response surface plot (Figure 2F) may therefore be used to depict the combined influence of variables X_1 and X_2 , which shows that Y_6 changes linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with Carbopol® 980 (X_2) - not span-80/tween-80 (X_1) - is the evidence from the comparative plot of the response surface. From this explanation, it can be concluded that the AT can be changed by selecting the right X_2 level. Details of the ANOVA results for measured responses are also presented in Table 2. In the end, the emulsifier factor significantly affected the response of PS, ZP, and spreadability of TEO-loaded Negs. At the same time, gel former affects the AT and pH.

2.4. Optimized TEO-loaded Negs

The formulation was optimized with Design-Expert®, version 13 software. The optimized Negs were selected based on the minimum PS and PDI; maximum ZP; value in pH range, spreadability, and AT (Table 3). Variables composition for optimized Negs is span-80/tween-80 of 8.68% and Carbopol® 980 of 1.18% with a desirability value of 0.801. The formulation with the maximum desirability value is the optimal formulation generated from the optimization phase of the program (60) – the optimization value formed as indicated by the desirability value close to one.

The desirability value range is 0-1. Figure 2 describes the optimization results in the form of a 2D contour. Contour is a two-dimensional response image that was presented using a predictive model for PS, PDI, ZP, pH, spreadability, and AT response values. The contour graph shows the desirability value of 0.801, which is the closest value to 1 compared to the other points. Figure 3 shows the projection in the form of a 3D surface; the low area shows low desirability, while the high area shows high desirability and is getting closer to 1. At this stage, the software predicts the response values shown in Table 4. Three confirmation runs need to be performed to validate optimization (61). The optimization model and estimates are validated by the observed optimized Negs, which show an acceptable variation from the predicted values (Table 4). We tested the optimized Negs' physical properties for further investigation, such as organoleptic, homogeneity, freeze-thaw, viscosity, and flow properties.

2.5. Evaluation of TEO-loaded Negs

Nanoemulsion systems can cover oily drugs' bitter or unpleasant taste (62). The organoleptic results of TEO-loaded Negs have a less distinctive turmeric odor, which is white and semisolid (Table 5). That is due

239 to the drug entrainment of oil with the oil phase effectively preventing evaporation and masking its specific
240 odor (63). The homogeneity test results of the optimum TEO-loaded Negs formulation showed a
241 homogeneous preparation, as evidenced by the absence of coarse grains. This homogeneity was correlated
242 with the optimal formulation of PS and PDI of 182.3 ± 5.5 and 0.242 ± 0.003 , respectively. The low PDI
243 indicates uniformity or homogeneous dispersion of globules Negs (64). In addition, the small size of the
244 globule (± 200 nm) is not included in the coarse dispersion (41).

245 The thermodynamic stability test of the system was carried out using a freeze-thaw cycle to identify
246 the presence of metastable Negs in the optimal formulation. It aims to see the separation of the water and oil
247 phases due to the influence of extreme temperatures (65). The thermodynamic stability of any system is
248 determined by the change in free energy between the system and its surroundings (66). The test results on
249 the optimum formula for six cycles showed promising results; namely, there was no separation. This
250 stability was correlated with the ZP of the optimal formulation of 57.23 ± 2.91 mV. The surface charge's
251 magnitude was directly related to the stability of any Negs. It is evidenced by the high repulsive force
252 between the Negs globules preventing coalescence, which was characterized by the absence of phase
253 separation (67). Similar results were found in the Negs study containing thymoquinone, which had ZPs
254 between -26.7 and -30.6 mV (66).

255 The pH conditions indirectly affect the viscosity indicated by Negs because they influence the
256 swelling ability of Carbopol® 980. This excipient is a gel former and a thickener (52). It plays an essential role
257 in the viscosity of Negs. Carbopol is dispersed in water to form an acidic colloidal solution with a low
258 viscosity. Neutralizing with triethanolamine increases Negs' viscosity because a stable water-soluble gel was
259 formed (50). Viscosity was carried out with #7 spindle (Brookfield digital RV DV-E) at 50 rpm of $32240 \pm$
260 2257.7 cP, indicating significantly high viscosity on Negs with pH 4.5. The magnitude of the viscosity is
261 correlated with AT and spreadability. Viscosity is inversely proportional to spreadability (52). In contrast,
262 the AT is directly proportional to the viscosity. A high-viscosity system will form stronger interfacial
263 interactions and increase intermolecular interactions in the polymer network, increasing cohesion, adhesion
264 strength, and AT (54).

265 Determining the rheology of a semisolid preparation is essential for controlling the consistency
266 required to ensure the performance and formulation durability and to describe the mechanical (flow
267 properties) system (68). The rheological study was conducted in the shear rate range of 6.81 – 40.86 s⁻¹ at 25°C.
268 The consistency index equals the apparent viscosity at a shear rate of 1 s⁻¹. The consistency index measured
269 on TEO-loaded Negs was 155.67 cP and $n = 0.22$. The flow index measures the system's deviation from
270 Newtonian behavior ($n = 1$). A value of $n > 1$ indicates dilatation or shear thickening flow, and $n < 1$
271 indicates pseudoplastic or shear thinning. The flow index typically lowers the thicker the base. Negs
272 produce a 0.22 flow index, which implies pseudoplastic flow behavior. A colloidal network structure aligned
273 with the shear direction and decreases viscosity as the shear rate increases have led to this pseudo-plasticity.
274 The developed system will require a specific force to discharge (69). The results of the flow properties test
275 showed that the optimal formula made was a pseudoplastic thixotropic flow type (Figure 4). Thixotropic is a
276 flow property expected in pharmaceutical preparations because it has high consistency in the container but
277 can be poured and dispersed easily (70).

278 4. CONCLUSION

279 Based on the results of the CCD-RSM analysis, the optimum span-80/tween-80 as an emulsifier is
280 8.68%, and Carbopol® 980 as a gel former was 1.18%. The resulting response is a PS of 182.3 nm, PDI 0.242,
281 ZP 57.23 mV, pH 4.51, AT 6.45 seconds, and spreadability of 6 cm. Optimized formulation viscosity is 32240
282 cP with pseudoplastic thixotropic flow properties. Thus, the developed TEO-Negs can be a potential
283 delivery system and a promising suitable approach for topical preparations.

284 5. MATERIALS AND METHODS

285 5.1. Materials

286 TEO (*Curcuma longa*) was purchased from Darjeeling Sembrani Aroma (Indonesia), sorbilene O E/P from
287 Lamberti (Italy), span-80 from Croda (Singapore), propylene glycol from Dow Chemical Pacific (Singapore),
288 Carbopol® 980 NF from Lubrizol AM (Cleveland), nipagin M from Clariant Produkte (Deutschland),
289 propylparaben from Alpha Chemika (India), triethanolamine from Dow Chemical Pacific (Switzerland), and
290 1.1-diphenyl-2-picrylhydrazyl (DPPH) from Smart-Lab (Indonesia).

291 5.2. Methods

292 5.2.1. Identification of TEO

293 Organoleptic tests include observing form, color, and odor. Phenolic identification was carried out by
294 adding one drop of 5% FeCl₃ to a 1 ml sample. Dark green to black colors indicate phenolic compounds'
295 presence (71). Terpenoid testing was performed by adding Lieberman-Burchard reagent containing
296 anhydrous acetic acid and concentrated sulfuric acid (3:1) into a 1 ml sample. Brownish or violet ring form
297 indicates the presence of terpenoids (72). Spectrophotometry was used to determine antioxidant activity
298 with the DPPH method (5). The calibration curve for the DPPH concentration against absorbance was made
299 at a maximum wavelength of 516 nm. The absorbance was measured in a mixture of sample solution and
300 DPPH with a particular concentration after 30-min of incubation in a dark room. IC₅₀ was calculated from
301 the inhibition percentage and absorbance.

302 5.2.2. Preparation of TEO-loaded Negs

303 TEO-loaded Negs were produced using a high-energy method, which used a mechanical device to produce a
304 highly disruptive force to break up the water and oil phases to obtain nano-sized globules (18). The oil phase
305 (M1) was prepared by mixing span-80 with 5% turmeric oil using a magnetic stirrer (WiseStir Wisd) at 1,500
306 rpm for 20 min. A total of 0.18% methylparaben and 0.02% propylparaben dissolved in 15% propylene glycol
307 (M2). Then, the distilled water was stirred with tween-80, and M2 was added gradually until homogeneous
308 at 1,500 rpm for 20 min (M3). M1 was stirred with M3 until homogeneous at 1,500 rpm for 40 min to form a
309 clear and transparent nanoemulsion, then let left for 24 hours. The gel base was prepared by mixing
310 Carbopol® 980 NF with distilled water at 70°C and left for 24 hours. Then, gradually add 1% triethanolamine
311 to form a gel mass. Nanoemulsion was added slowly into the gel base while homogenized using a
312 homogenizer (AEG) at 2,000 rpm for 10 min.

313 5.2.3. Experimental Design

314 This study selects the CCD-RSM method to develop the TEO-loaded Negs formulation. For preliminary
315 screening on PS, PDI, ZP, pH, AT, and spreadability, a 2-factor CCD-RSM at two levels (high and low) was
316 used. Based on previous experiments and study literature, high and low variables were determined. The
317 CCD of the statistical package Design-Expert® version 13 software (Stat-Ease Inc., Minneapolis, MN) was
318 used to examine the influence of the specified independent variable on the response variable to obtain the
319 optimal formulation for TEO-loaded Negs. CCD planned 14 experiments were done under controlled
320 circumstances (Table 3). The independent variables were span-80/tween-80 (X₁) and Carbopol® 980
321 concentration (X₂). The observed response of the dependent variables was PS (Y₁), PDI (Y₂), ZP (Y₃), pH (Y₄),
322 spreadability (Y₅), and AT (Y₆).

323 5.2.4. Determination of PS, PDI, and ZP of Negs

324 PS and PDI of Negs were assessed using the Delsa Max Pro Particle Size Analyzer LS 100Q (Beckman
325 Coulter, USA) at 25°C utilizing the dynamic light scattering (DLS) method or photo correlation spectroscopy.
326 For analysis, 1 ml of samples was dispersed in 9 ml aqua pro injection. Into the cuvette, 1 ml of suspension
327 and 5 ml of aqua pro injection were added as a diluent, and the results were read on the instrument. All
328 measurements were made at a scattering angle of 90°. ZP was determined by particle size analyzer through
329 mobility and conductivity measurements. The temperature was set to 25°C, and the mean electric field was
330 set to 16 V/cm (73). The final result is the mean of each sample's three repeated measurements.

331 5.2.5. Determination of pH, Spreadability, and AT

332 pH was measured at a temperature of 25°C using a pH meter that had been previously calibrated with buffer
333 solutions of pH 4 and 7. The calibration step was completed when the pH value indicated on the screen
334 matched the correct pH value and was stable. Afterward, the electrode was dipped in Negs and recorded the
335 value shown on the screen (74). Spreadability was measured by adding 0.5 g of Negs in the center of a glass
336 covered with another glass. The preparation diameter was measured longitudinally and transversely; for

337 every minute added, 50 g was to a total weight of 150 g (75). Adhesion time determine by the single-lap
338 shear test method (54). A-0.5 g of Negs was placed on a slide, then covered with another slide, and given a
339 load of 1 kg for 3 min. The glass object was mounted on the test apparatus, and 80 g of the load was released
340 until both glass objects were released, and the time was recorded (75).

341 5.2.6. Organoleptic Observation and Homogeneity Test

342 Negs were placed in a glass object and directly observed for color, smell, and shape (76). A-0.1 g Negs were
343 spread over the slide, and homogeneity was observed. Test preparation is declared homogeneous if no
344 coarse grains exist (77).

345 5.2.7. Viscosity and Rheological Flow

346 Viscosity and rheology were determined with a Brookfield RV DV-E Viscometer with appropriate spindle
347 and speed. A-500 ml of Negs were put into a beaker glass; the spindle was installed, and the measured value
348 was recorded as viscosity Negs. In this study, the spindle used spindle no. 7. Flow properties were
349 determined by measuring the viscosity using a right spindle from low to high rotational speed and vice
350 versa (74). Flow index and consistency index are determined from the power law equation:

$$351 \tau = K r^n$$

352 where τ is the shear stress, K is the consistency index, r is the shear rate, and n is the flow index.

353 Taking logs on both sides,

$$354 \log \tau = \log K + n \log r$$

355 So, from the log, shear stress Vs. Log shear rate plot, the plot slope was used as the flow index and the
356 antilog of the Y-intercept as the consistency index (69).

357 5.2.8. Freeze-thaw Test

358 Negs were stored at 4 ± 2 °C, then transferred to 40 ± 2 °C for 48 hours (1-cycle), then repeated for 6-cycles.
359 Phase separation was observed in each cycle (74).

360 5.2.9. Statistical Analysis

361 The Design-Expert® version 13 software was used to conduct the statistical study (Stat-Ease Inc.,
362 Minneapolis, MN). Analysis was done at a sig – $p < 0.05$ and $p < 0.01$ after three times of each measurement.

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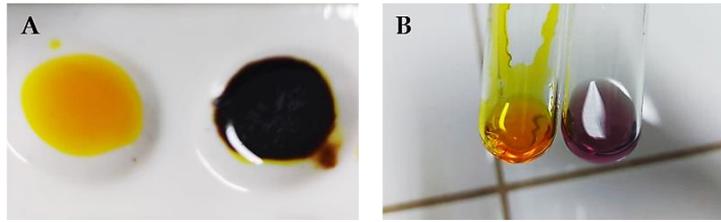
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Figure 1. The result of qualitative test observation is a phenolic test with a positive result marked in a blackish green color (a) and a terpenoid test with a positive result marked with a purple color (b).

547 **Table 1.** Evaluation results of independent variables and dependent variables with the CCD design for optimizing TEO-
548 loaded Negs.

Std	X ₁ (%)	X ₂ (%)	Y ₁ (nm)	Y ₂	Y ₃ (mV)	Y ₄	Y ₅ (cm)	Y ₆ (s)
1	5.50	1.02	206.9 ± 0.7	0.571 ± 0.000	35.39 ± 1.50	5.45	5.55	5.65
2	2.32	1.15	457.0 ± 0.9	0.000 ± 0.000	13.90 ± 1.92	4.60	5.00	6.99
3	8.68	0.90	166.7 ± 0.4	0.378 ± 0.074	31.64 ± 0.53	6.35	6.20	4.23
4	2.32	0.90	411.0 ± 5.8	0.000 ± 0.000	30.67 ± 0.89	6.29	5.00	4.30
5	5.50	1.02	302.1 ± 3.6	0.571 ± 0.000	36.45 ± 0.45	5.46	5.60	5.63
6	5.50	1.02	263.0 ± 3.7	0.571 ± 0.000	18.49 ± 1.77	5.47	5.55	5.64
7	8.68	1.15	198.1 ± 1.2	0.242 ± 0.003	57.23 ± 2.91	4.65	6.15	6.95
8	5.50	1.02	246.9 ± 7.2	0.571 ± 0.000	26.60 ± 0.17	5.46	5.60	5.67
9	5.50	1.02	206.0 ± 2.3	0.571 ± 0.000	31.77 ± 1.01	5.46	5.60	5.65
10	10.00	1.02	160.8 ± 1.3	0.285 ± 0.024	46.72 ± 1.97	5.31	6.25	5.84
11	1.00	1.02	244.2 ± 4.3	0.571 ± 0.000	31.26 ± 0.57	5.78	4.90	6.88
12	5.50	1.02	238.1 ± 3.0	0.571 ± 0.000	30.78 ± 0.53	5.45	5.55	5.64
13	5.50	1.20	206.2 ± 1.6	0.571 ± 0.000	30.56 ± 0.78	4.57	5.50	7.08
14	5.50	0.85	240.5 ± 2.2	0.571 ± 0.000	20.12 ± 0.16	6.39	5.65	4.22

Y₁: Particle size (PS), Y₂: Polydispersity index (PDI), Y₃: Zeta potensial (ZP), Y₄: pH, Y₅: Spreadability, Y₆: Adhesion time (AT)

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551 **Table 2.** Statistical analysis of PS (Y_1), PDI (Y_2), ZP (Y_3), pH (Y_4), spreadability (Y_5), and AT (Y_6) TEO-loaded Negs on
552 CCD.

Factors		Y_1	Y_2	Y_3	Y_4	Y_5	Y_6
X_1	Coefficient	-77.64	0.027	8.27	-0.069	0.53	-0.20
	p-value	0.0063**	0.7325	0.0080**	0.1430	<0.0001**	0.1406
X_2	Coefficient	3.61	-0.017	2.95	-0.75	-0.033	1.18
	p-value	0.8783	0.8289	0.2668	<0.0001**	0.1157	<0.0001**
X_1X_2	Coefficient		-0.034	10.59			
	p-value		0.7602	0.0136*			
X_1^2	Coefficient		-0.16				
	p-value		0.0818				
X_2^2	Coefficient		-0.086				
	p-value		0.3087				
Intercept	Coefficient	253.39	0.57	31.54	5.48	5.58	5.74
Degree of freedom		2	5	3	2	2	2
Sum of squares		48332.05	0.24	1065.33	4.48	2.28	11.49
Mean of squares		24166.02	0.047	355.11	2.24	1.14	5.74
F-value		5.69	1.02	7.06	145.08	386.43	46.35
p-value		0.0201	0.4645	0.0078	<0.0001	<0.0001	<0.0001
R-Squared		0.5084	0.3898	0.6794	0.9635	0.9860	0.8939

553 X_1 : span-80/tween-80; X_2 : Carbopol® 980

554 * p-value < 0.05

555 ** p-value < 0.01

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Table 3. The independent and dependent variables with levels and limits in CCD for TEO-loaded Negs development.

Variables	Code	Start point (- α)	Low level (-1)	Central level	High level (+1)	Start point (+ α)	Units
Independent variables							
Span-80/tween-80	X ₁	0.85	2.32	1.02	5.50	1.20	% w/w
Carbopol® 980	X ₂	1	0.90	5.50	1.02	10	% w/w
Dependent variables							
Limits							
Particle size (PZ)	Y ₁	Minimum					nm
Polydispersity index (PDI)	Y ₂	Minimum					
Zeta potensial (ZP)	Y ₃	Maximum					mV
pH	Y ₄	is in range					
Spreadability	Y ₅	is in range					cm
Adhesion time (AT)	Y ₆	is in range					s

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Table 4. Optimized TEO-loaded Negs' actual and predicted values for each response.

Responses	Predicted values	Actual value ^a	Error ^b (%)
Y1 (nm)	180.2	182.3 ± 5.5	1.165
Y2	0.250	0.242 ± 0.003	-3.200
Y3 (mV)	56.30	57.23 ± 2.91	1.652
Y4	4.50	4.51 ± 0.02	0.222
Y5 (cm)	6.07	6.0 ± 0.2	-1.153
Y6 (s)	6.98	6.45 ± 0.19	-7.593

a Data listed is the mean ± standart deviation, n = 3

b Error (%) = [(Actual value - Predicted value)/Predicted value] * 100%

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Table 5. Additional evaluation on TEO-loaded Negs with an optimal formulation.

Evaluation	Result
Organoleptic	Color: White; Odor: typical turmeric
Homogeneity	Homogeneous
Freeze-Thaw	Cycles 0-6, no separation occurs
Viscosity ^a	32240 ± 2257,7 cP

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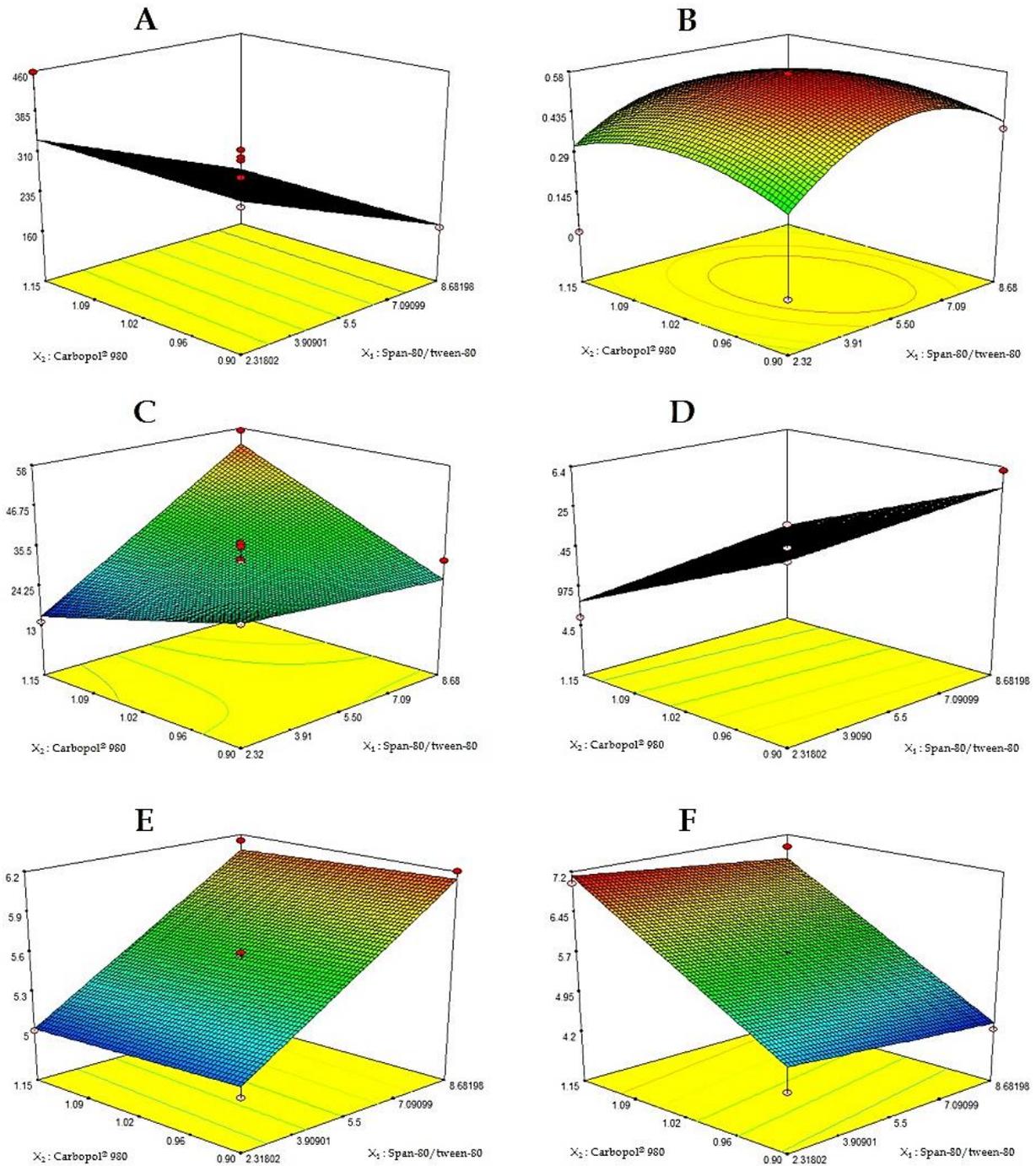
^a Data listed is the mean ± standart deviation, n = 3

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Figure 2. Effect of span-80/tween-80 and Carbopol[®] 980 concentration on PS (a) PDI (b) ZP (c) pH (d) spreadability (e) AT (f).

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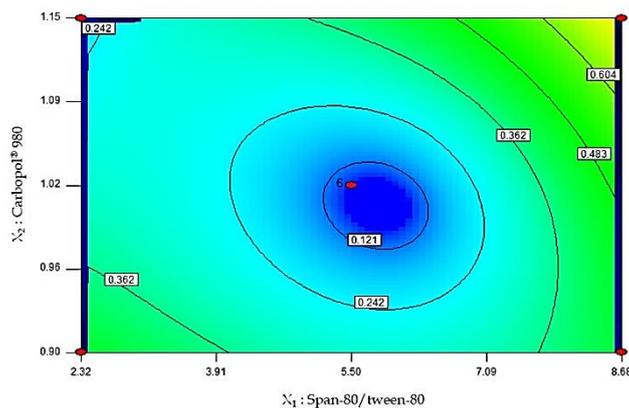


Figure 3. Contour plot desirability value of optimal formulation.

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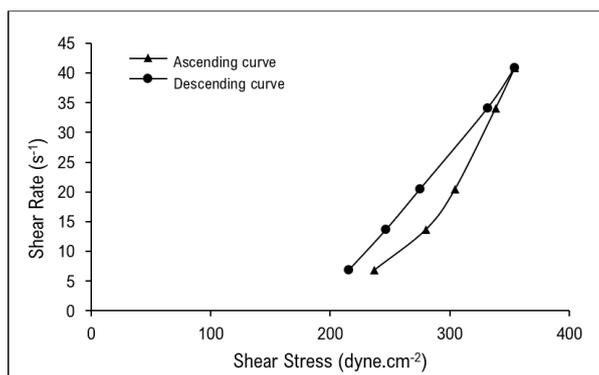


Figure 4. Optimal formulation flow properties.

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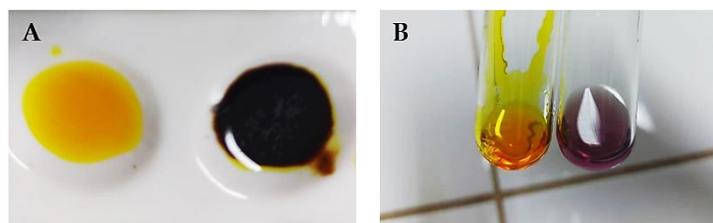
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46 turmeric aroma. Identification results in the phenolic test showed a black-green color (**Figure 1A**) due to the
47 formation of a phenolic-Fe³⁺ complex (30). In the terpenoid test, reaction results show a purple color (**Figure 1B**) as a
48 positive sign. Based on these results, it can be concluded that TEO contains phenolic and terpenoid compounds.
49 Several studies reported that TEO has powerful antioxidant potential both *in-vitro* and *in-vivo* (2,8,31). Oxygenated
50 monoterpenes and sesquiterpenes, such as ar-turmerone, α-turmerone, and β-turmerone, were included in the
51 terpenoid compounds and reported to play a role in this activity. In addition, contained phenol was also reported to
52 act as an excellent free radical scavenger because its reduction potential was lower than oxygen's (8). In our study, the
53 antioxidant activity of TEO was expressed as an IC₅₀ of 9.88 μm/ml, which was included in the powerful antioxidant
54 category because of below 50 μm/ml (32). It differed from other studies' results, which showed an IC₅₀ of 2274.02
55 μm/ml and above 1000 μm/ml (2,10). Meanwhile, similar findings stated IC₅₀ values of 10.03 μm/ml, 3.227 μm/ml,
56 and 14.5 μm/ml (33–35). The difference value was due to the use of TEO from different origin sources, regional
57 conditions, and the extraction method used.

58 5.2.1. Identification of TEO

59 Organoleptic tests include observing form, color, and odor. Phenolic identification was carried out by adding one
60 drop of 5% FeCl₃ to a 1 ml sample. Dark green to black colors indicate phenolic compounds' presence (71). Terpenoid
61 testing was performed by adding Liebermann-Burchard reagent containing anhydrous acetic acid and concentrated
62 sulfuric acid (3:1) into a 1 ml sample. Brownish or violet ring form indicates the presence of terpenoids (72).
63 Spectrophotometry was used to determine antioxidant activity with the DPPH method (5). The calibration curve for
64 the DPPH concentration against absorbance was made at a maximum wavelength of 516 nm. The absorbance was
65 measured in a mixture of sample solution and DPPH with a particular concentration after 30-min of incubation in a
66 dark room. IC₅₀ was calculated from the inhibition percentage and absorbance.



67
68 **Figure 1.** The result of qualitative test observation is a phenolic test with a positive result marked in a blackish green color (a) and a
69 terpenoid test with a positive result marked with a purple color (b).

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72 5) Why did the authors choose turmeric essential oil? How the authors analyzed the effectiveness of the
73 oil.

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Response:

76 *This study was conducted based on the adequate TEO availability in our country, and several research reports have*
77 *promising results on its pharmacology activity. As we said earlier, the qualitative identification of TEO showed that*
78 *the TEO used was positive for containing phenolic and terpenoid compounds. Both groups of these compounds were*
79 *reported to provide antioxidant activity. Accordingly, tests of antioxidant activity test on TEO were carried out and*
80 *noted in the section "Identification of TEO" (lines 69-70 and 298-301).*

81
82

82 6) Result and discussion part should be supported by the literatures.

83
84

Response:

85 We accepted your advice. Results and discussion enriched with sufficient literature. We added 32 additional
86 references that support our discussion of the results we found in this study. The distribution of the supplementary
87 literature is as follows:

- 88 1. Identification of TEO: References 2, 8, 10
- 89 2. Preparation of TEO-loaded Negs: References 36, 37
- 90 3. Experimental Design by CCD-RSM: References 18, 19, 24, 27, 30, 31, 32, 33, 34, 35, 36, 37, 39, 50, 51, 54, 55,
91 56, 58, 59
- 92 4. Optimized TEO-loaded Negs: References 60, 61
- 93 5. Evaluation of TEO-loaded Negs: References 62, 63, 64, 66, 67, 68, 69
- 94
- 95 7) In methods: Explain what TEA means.
- 96

97 **Response:**

98 *The TEA in question is triethanolamine. We have changed the term "TEA" to "triethanolamine" (line 310).*

99

- 100 8) Adhesiveness is the work required to overcome the attractive forces between the surface of the sample
101 and the probe surface. The unit of adhesiveness cannot be seconds. The authors should explain the
102 adhesion study in more detail. The results of adhesion should be supported by the literature.
- 103

104 **Response:**

105 *We accepted your suggestion. We have regenerated the abbreviations section, italics, and usage of appropriate terms
106 and units. An explanation regarding the adhesive properties of the preparation has been written on lines 196-218. The
107 use of appropriate units has also been updated.*

108

109 Topical dosage forms, such as Negs, adhere to the skin in two ways: they adhere directly to the rough
110 substrate to form a "mechanical interlock" and to the substrate via interaction (54). Good adhesion to Negs supports a
111 higher concentration gradient towards the skin and provides more drug penetration (18). Adhesive strength is
112 directly related to the AT on the Negs as measured using the single-lap shear test method with slight modifications
113 (55,56). The test was carried out by applying a shear load to the plates that flank the sample, which had been pre-
114 loaded, and given a measured force; the time taken for the plates to separate was recorded as AT. In our study, the
115 AT was from 4.22 s (STD#14) to 7.08 s (STD#13) in **Table 1**. They met the requirements based on the AT test results on
116 a 14 formulation. An AT was carried out to see how long a Negs could be attached to the skin. The AT requirement is
117 more than 4 seconds. The longer a Negs could be attached to the skin, it showed the better result, where it is expected
118 that more active substances can be absorbed due to the time the Negs was in contact with the skin (57). Based on
119 **Table 2**, the linear model was found to be significant in the AT response with an F-value model of 46.35 (p-value <
120 0.0001). The linear model equation suggested by the software can be seen as follows:

121

$$Y_6 = 5.74 - 0.20X_1 + 1.18X_2$$

122

123 The equation reveals that the AT was significantly (p-value < 0.0001) affected by the gel former (X_2) or,
124 indirectly, the same adhesion strength. This finding is in agreement with the literature (58,59). A response surface plot
125 (**Figure 2F**) may therefore be used to depict the combined influence of factors X_1 and X_2 , which shows that Y_6 changes
126 linearly with the sum of the two factors. However, the higher gradient in the response surface with Carbopol® 980
127 (X_2) - not span-80/tween-80 (X_1) - is the evidence from the comparative plot of the response surface. From this
128 explanation, it can be concluded that the AT can be changed by selecting the right X_2 level. Details of the ANOVA
129 results for measured responses are also presented in **Table 2**. In the end, the emulsifier factor significantly affected the
response of PS, ZP, and spreadability of TEO-loaded Negs. At the same time, gel former affects the AT and pH.

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132 Reviewer: 2

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134 Comments to the Author

135 The content of the manuscript and the topic seems in line with journals target audience but there are some
136 serious drawbacks need to be addressed:

137

138 1) The English of the manuscript needs to be revised by a native speaker or a professional editing service,

139

140 **Response:**

141 *We apologize for our poor writing. Improvements have been made with the help of the Language editing service.*

142

143 2) The application of CCD method is not clear nor understandable in the manuscript. In particular, how
144 many center points were used, which $\pm\alpha$ was preferred in the design etc?

145

146 **Response:**

147 *We accepted your suggestion and added this table. Several details provide to clarify the CCD design.*

148

149

Table 3. The independent and dependent variables with levels and limits in CCD for TEO-loaded Negs development.

Variables	Code	Start point (- α)	Low level (-1)	Central level	High level (+1)	Start point (+ α)	Units
Independent variables							
Span-80/tween-80	X ₁	0.85	2.32	1.02	5.50	1.20	% w/w
Carbopol® 980	X ₂	1	0.90	5.50	1.02	10	% w/w
Dependent variables							
Limits							
Particle size (PZ)	Y ₁	Minimum					nm
Polydispersity index (PDI)	Y ₂	Minimum					
Zeta potential (ZP)	Y ₃	Maximum					mV
pH	Y ₄	is in range					
Spreadability	Y ₅	is in range					cm
Adhesion time (AT)	Y ₆	is in range					s

150

151

152 3) The most significant missing factor is validation of the CCD. After application of CCD-RSM, suggested
153 optimal points are better presented in table3. Once the optimal points are clarified, three Negs
154 formulation at this suggested optimal point are needed to be prepared experimentally along with all
155 the characterization in order to show the correlation between predicted and experimentally found
156 values.

157

158 **Response:**

159 *We accepted your advice. We have added a discussion of the CCD validation process (lines 231-233).*

160

161 4) The purpose of the Negs formulation is not clear in the manuscript which is another bottleneck of the
162 manuscript. The purpose of this formulation needs to be addressed and further characterized along
163 with the purpose.

164

165 **Response:**

166 *We accepted your advice. In the introduction section, we improve the writing to clarify the choice of nanoemulgel in
167 delivering TEO.*

168
169
170
171

**RESPONSE TO REVIEWERS
OF THE JOURNAL OF RESEARCH IN PHARMACY**

172
173

MANUSCRIPT ID: MPJ-10622.REV-1

174 Reviewer: 1

175
176 Comments to the Author

177 The authors' response to my previous comments was presented as a separate file without referring to the
178 comments and It is impossible to follow the updated manuscript file since the changes made are not
179 traceable. Under these circumstances the manuscript is not acceptable.

180
181 **Response:**

182 *We apologize for the ignorance of the "response to reviewers" writing. We have improved the response*
183 *writing. Hopefully, it can be accepted by reviewers. Thank you.*

**6. Bukti Konfirmasi Review dan Hasil Review
Ketiga (6 Maret 2023)**

Journal of Research in Pharmacy : Revision request for your manuscript

1 message

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Mon, Mar 6, 2023 at 12:42 AM

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Suggestions

1. Reviewer Comments

Updated manuscript covers my earlier points but there are some minors points:

- The whole manuscript needs to be checked in terms of non-scientific and non-specific statements such as "...non-friendly solvents to the formulation....". In particular what does "non-friendly" statement refer to??? Please be specific.
- The resolution of the Figure 2 and Figure3 should be enhanced. The number and the letters are pixelized.

Manuscript Information

Manuscript ID: MPJ-10622.REV-2

Title in English: Response surface methodology for optimization of turmeric essential oil-loaded nanoemulgel

Small Title in English: No information entered

Authors: Nining Nining¹, Anisa Amalia¹, Fatimatuz Zahrok²

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Keywords in English: Central composite design; nanoemulgels; response surface methodology; turmeric essential oil; topical delivery.

Manuscript Type: Research article

Processing Status: Minor Revision

Manuscript Files

File Name	File Size	Date Created	Category	Description
MPJ-10622-1-cover-letter-nining-et-al..pdf (../pdf-files/out/12282-MPJ-10622-1-cover-letter-nining-et-al..pdf)	31 KB	Sep 03, 2022	Cover letter	None
MPJ-10622-9-isi-jrp-checklist.pdf (../pdf-files/in/12282-MPJ-10622-9-isi-jrp-checklist.pdf)	3388 KB	Sep 03, 2022	Author Checklist Form	None
MPJ-10622-8-isi-jrp-copyright-form-integrated.pdf (../pdf-files/in/12282-MPJ-10622-8-isi-jrp-copyright-form-integrated.pdf)	327 KB	Sep 03, 2022	Copyright Transfer Form	None
MPJ-10622-4-manuscript-nining-et-al..pdf (../pdf-files/out/12282-MPJ-10622-4-manuscript-nining-et-al..pdf)	818 KB	Sep 03, 2022	Main Document	None
MPJ-10622-7-figure-1..jpeg (../pdf-files/in/12282-MPJ-10622-7-figure-1..jpeg)	513 KB	Sep 03, 2022	Figure	None

MPJ-10622-9-figure-2..jpeg (../pdf-files/in/12282-MPJ-10622-9-figure-2..jpeg)	190 KB	Sep 03, 2022	Figure	None
MPJ-11427-3-manuscript-nining-et-al.-rev01-proofread.rev-1.pdf (../pdf-files/out/12282-MPJ-11427-3-manuscript-nining-et-al.-rev01-proofread.rev-1.pdf)	2042 KB	Nov 23, 2022	Main Document	None
MPJ-11427-8-figure-1..rev-1.jpg (../pdf-files/in/12282-MPJ-11427-8-figure-1..rev-1.jpg)	26 KB	Nov 23, 2022	Figure	None
MPJ-11427-7-figure-2..rev-1.jpg (../pdf-files/in/12282-MPJ-11427-7-figure-2..rev-1.jpg)	232 KB	Nov 23, 2022	Figure	None
MPJ-11427-9-figure-3..rev-1.jpg (../pdf-files/in/12282-MPJ-11427-9-figure-3..rev-1.jpg)	82 KB	Nov 23, 2022	Figure	None
MPJ-11427-3-response-to-reviewers.rev-1.pdf (../pdf-files/out/12282-MPJ-11427-3-response-to-reviewers.rev-1.pdf)	32 KB	Nov 23, 2022	Response to Reviewers	None
MPJ-12282-7-manuscript-revision-nining-et-al..rev-2.pdf (../pdf-files/out/12282-MPJ-12282-7-manuscript-revision-nining-et-al..rev-2.pdf)	0 KB	Feb 13, 2023	Main Document	None
MPJ-12282-9-manuscript-revision-nining-et-al..rev-2.pdf (../pdf-files/out/12282-MPJ-12282-9-manuscript-revision-nining-et-al..rev-2.pdf)	2042 KB	Feb 13, 2023	Main Document	None
MPJ-12282-9-response-to-reviewers-2-.rev-2.pdf (../pdf-files/out/12282-MPJ-12282-9-response-to-reviewers-2-.rev-2.pdf)	0 KB	Feb 13, 2023	Response to Reviewers	None
MPJ-12282-8-response-to-reviewers-2-.rev-2.pdf (../pdf-files/out/12282-MPJ-12282-8-response-to-reviewers-2-.rev-2.pdf)	343 KB	Feb 13, 2023	Response to Reviewers	None

Score Sheet

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Does the content and value of the work justify publication in Marmara Pharmaceutical Journal ?

After revision

Does the title of the manuscript reflect the contents of the study ?

Yes

Are the keywords sufficient and appropriate ?

Yes

Is the summary concise and informative?

Yes

Is the text divided appropriately according to the article type ?

Yes

Is the language adequate?

Yes

Are the nomenclature and scientific terminology correct?

Yes

Are the references complete and recent?

Yes

Are the figures tables and graphics necessary ?

Yes

Are the figures tables and graphics clear ?

No

Is the introduction part

sufficiently developed

Are the experimental procedures sound?

Yes

Is the results and discussion part

sufficiently developed

Is conclusion sufficient and correlated with the results ?

Yes

Is the information about the approval of ETHICAL COMMISSION presented ?

Not applicable

**7. Bukti Konfirmasi Review Submit Revisi Ketiga,
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Journal of Research in Pharmacy : Confirmation for revised manuscript

1 message

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Response surface methodology for optimization of turmeric essential oil-loaded nanoemulgel

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ABSTRACT: Turmeric Essential Oil (TEO) has an antioxidant and anti-inflammatory activity to be formulated in a topical dosage form. Nanoemulgels (Negs) development, based on varying concentrations of emulsifiers and gel formers, affects their characteristics and stability. This study focuses on optimizing TEO-loaded Negs based on physical and mechanical characterization, which have promising topical applications. Negs were created using the high-energy approach and optimized using Response Surface Methodology (RSM) and the Central-Composite Design (CCD) for the optimization of span-80/tween-80 (X_1) and Carbopol® 980. (X_2). Observed variable responses were particle size (PS) (Y_1), polydispersity index (PDI) (Y_2), zeta potential (ZP) (Y_3), pH (Y_4), spreadability (Y_5), and adhesion time (AT) (Y_6). Actual responses of Negs were compared with the CCD-RSM predictions to validate the model. In addition, other physical evaluations were observed, such as organoleptic observations, homogeneity, freeze-thaw tests, viscosity, and flow properties. Optimized TEO-loaded Negs were made with 8.68% span-80/tween-80 and 1.18% Carbopol® 980. The evaluation results showed the optimal TEO-loaded Negs on nano-metric size (182.3 ± 5.5 nm) with low PDI (0.242 ± 0.003), good ZP (-57.23 ± 2.91 mV), pH (4.51 ± 0.02), spreadability (6.0 ± 0.2 cm), and AT (6.45 ± 0.19 s). TEO-loaded Negs have an excellent appearance and did not run phase separation at extreme temperature storage with pseudoplastic thixotropy flow. Thus, the developed TEO-loaded Negs can be a potential delivery system and a promising suitable approach for topical preparations.

KEYWORDS: Central composite design; nanoemulgels; response surface methodology; turmeric essential oil; topical delivery.

1. INTRODUCTION

Turmeric is the dried rhizome of *Curcuma longa* L. (Zingiberaceae), which derives from Southeast Asia and is cultivated mainly in India, followed by Bangladesh, China, Thailand, Cambodia, Indonesia, Malaysia, and the Philippines (1). Steam distillation extracts turmeric essential oil (TEO) from the turmeric rhizome (2). Chemical constituents with the most significant proportion were oxygenated monoterpenes and sesquiterpenes, which include β -turmerone, α -turmerone, and ar-turmerone (3–5). The pharmacological activities of TEO have been reported in the form of antioxidants, anti-inflammatory, antinociceptive, antidermatophytic, antifungal, and antibacterial activities (2,6–9). These reducing power and radical scavenging abilities are associated with the high antioxidant potential of TEO (8,10). This pharmacological activity justifies its use in various applications, including cosmetics and phytomedicines (7,11,12). Like other essential oil, TEO has limited use due to its volatility, instability under certain conditions, lipophilicity, and low aqueous solubility (13,14). Many recent studies are oriented toward solving these limitations, so that efficacy of the essential oil lasts longer and increases.

Previously, TEO has been developed for cream as a conventional drug delivery system, patch, and nanoemulsions (7,11,15,16). Most water-based liquid or semisolid systems have limitations in delivering lipophilic drugs (17). Nanoemulsions are an established alternative for delivering lipophilic drugs by increasing topical absorption (18). **The main advantage of topically administered nanoemulsions is the ability to increase penetration and permeation of drugs through the skin without adding non-physical enhancers and a large number of surfactants to the formulation, which can cause skin irritation, especially with long-term usage (18,19).** However, this system has problems with low viscosity due to poor

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33 spreadability and skin retention (20). Nanoemulgels (Negs), a combination of nanoemulsion and hydrogel,
34 were made to improve the characteristics (18,19,21).

35 Negs consist of a hydrogel system and an emulsion with nano-sized globules. An emulsifier in the
36 form of surfactants and co-surfactants stabilizes the emulsion, which serves as a drug delivery platform (18).
37 Surfactants reduce the interfacial surface tension of immiscible liquids and change the entropy of the
38 dispersion, thereby stabilizing a thermodynamically unstable system; co-surfactants are combined with
39 surfactants in the emulsification process by disrupting the surface layer (22). The emulsifier plays a role in
40 the emulsification process to increase stability when the product is kept for an extended time. On the other
41 side, gels are made from polymers, a gel former, that expand after absorbing a liquid (23). Gel former
42 increase the viscosity of the formulation and can react with emulsifiers to change the thickness (24). In
43 addition, topical Negs can increase patient compliance because of their non-irritating, non-greasy
44 characteristics and improved drug release (25). However, the available methods for manufacturing Negs
45 exhibit various limitations, which directly or indirectly affect the quality of the Negs formulations.

46 Currently, the principle of quality by design is adopted to ensure the quality of drugs, their safety, and
47 efficacy (26). The quality by design (QbD) trend is used to develop, optimize, and investigate the interaction
48 between particular variables and their related responses to achieve the optimal formulation (27). Our study
49 used Central Composite Design (CCD) on Response Surface Methodology (RSM) to select optimal
50 formulations and predict models that rely on statistical analysis (ANOVA) and exact equations (28,29). In
51 this aspect, we intended to develop a TEO-loaded Negs that would be optimized using a complete 2²-
52 factorial design and determine the independent factors' precise influence on the investigated dependent
53 variables. The choice and procedure factors were span-80/tween-80 (X₁) and Carbopol® 980 (X₂). Response
54 variables investigated were particle size (PS) (Y₁), polydispersity index (PDI) (Y₂), zeta potential (ZP) (Y₃),
55 pH (Y₄), spreadability (Y₅), and adhesion time AT) (Y₆). This work is the first step in developing an
56 optimized TEO preparation suitable for transdermal drug delivery for topical application.

57 2. RESULTS AND DISCUSSION

58 2.1. Identification of TEO

59 TEO identification is carried out to ensure the quality of the active ingredients. The tests included
60 organoleptic, phytochemical identity in the form of phenolics and terpenoids, and antioxidant activity by the
61 DPPH method. The results of this study can be seen in **Figure 1**. Organoleptically, TEO is an orange liquid
62 with a distinctive turmeric aroma. Identification results in the phenolic test showed a black-green color
63 (**Figure 1A**) due to the formation of a phenolic-Fe³⁺ complex (30). In the terpenoid test, reaction results show
64 a purple color (**Figure 1B**) as a positive sign. Based on these results, it can be concluded that TEO contains
65 phenolic and terpenoid compounds. Several studies reported that TEO has powerful antioxidant potential
66 both *in-vitro* and *in-vivo* (2,8,31). Oxygenated monoterpenes and sesquiterpenes, such as α -turmerone,
67 α -turmerone, and β -turmerone, were included in the terpenoid compounds and reported to play a role in this
68 activity. In addition, contained phenol was also reported to perform as an excellent free radical scavenger
69 because its reduction potential was lower than oxygen's (8). In our study, the antioxidant activity of TEO
70 was presented as an IC₅₀ of 9.88 μ m/ml, which was included in the powerful antioxidant category because of
71 below 50 μ m/ml (32). It differed from other studies' results, which showed an IC₅₀ of 2274.02 μ m/ml and
72 above 1000 μ m/ml (2,10). Meanwhile, similar findings stated IC₅₀ values of 10.03 μ m/ml, 3.227 μ m/ml, and
73 14.5 μ m/ml (33–35). The difference value was due to the use of TEO from different origin sources, regional
74 conditions, and the extraction method used.

75 2.2. Preparation of TEO-loaded Negs

76 TEO-loaded Negs were produced using the high-energy method efficiently. A high-pressure
77 homogenizer provides sufficient energy to increase the interfacial area and generate nano-size globules (36).
78 An emulsion system with a nanodroplet size must have flow properties that allow it to pass through the
79 homogenizer (37). Flow properties are inversely correlated with the amount of emulsifier and gel former
80 added to the Negs formulation. In this study, the selected optimized formulation was the one that produces
81 Negs with the smallest size, as shown by the limitations in **Table 3**. Therefore, the range of regulated
82 emulsifiers and gel formers concentrations still provides flow properties that allow the flow to remain good
83 using a high-energy homogenizer. In general, the preparation of Negs was carried out in 3-stages, namely,

84 nanoemulsions preparation, gel preparation, and Negs preparation. Negs were successfully produced in 14-
85 formulation with variations of emulsifiers and gel former with compositions that can be seen in **Table 1**.

86 2.3. Experimental Design by CCD-RSM

87 2.3.1. Fitting the Model

88 TEO-loaded Negs were optimized based on CCD in the RSM. CCD was used to establish the optimal
89 concentrations of emulsifier (X_1) and gel former (X_2) as the key parameters influencing the dependent
90 response. Prediction of the factorial axial design and the possible curvature in the response could be
91 obtained from the optimization process with an effective second-level design (38). Predicting the significant
92 influence of the independent variable on the dependent variable is essential for generating TEO-loaded
93 Negs. Based on the literature survey, two factors were chosen as independent variables, and six responses
94 were chosen as the dependent variable with the most significant influence on Negs. The independent
95 variables with their levels and the observed response variables are presented in **Table 1**.

96 2.3.2. Analysis of Design

97 Statistical data analysis must be carried out to predict and recognize the model. **Table 2** shows the
98 statistical analysis of a quadratic model for PDI, a 2FI model for ZP, and linear models for PS, pH, AT, and
99 spreadability. This table identifies factors with p-values less than a predefined threshold (0.01 and 0.05, with
100 a 99% and 95% confidence level, respectively) as influential factors. Besides the significant p-value, a large F-
101 value minimizes error in the model and lack of fit, preferably non-significant to fit the model data (27,39).
102 Depending on the most significant R-squared value and the least residual predictive sum of squares value,
103 the six responses demonstrated distinct models in their application. The chosen model had a non-statistically
104 significant lack of fit, and model validation was confirmed by the residual plot test of the regression model,
105 which was supported by supplemental information for all responses. Compared to the 2FI and simple linear
106 models, the quadratic model represented the impacts of numerous variables, including individual factors,
107 interactions, and the quadratic influence on the response.

108 2.3.3. Effect of Independent Variables on Dependent Variables

109 PS parameters are often used to characterize nanoparticles. Negs globule diameter means (Y_1) was
110 adjusted from 160.8 nm (STD#10) to 457.0 nm (STD#2). Based on **Table 2**, the PS response indicates a
111 significantly linear model with an F-value of 5.69 (p-value 0.0201 < 0.05). The suggested linear model
112 equation can be seen as follows:

$$113 Y_1 = 253.39 - 77.64X_1 + 3.61X_2$$

114 The equation shows that the Negs PS is significantly (p-value < 0.01) affected by emulsifier
115 concentration (X_1). The positive coefficient has a synergistic effect on the response. In contrast, the negative
116 coefficient has an antagonistic effect which concludes the inverse relationship of the independent variable
117 with a response (38,40). In addition, this factor has a more significant coefficient. It directly affects the PS,
118 which means that increasing emulsifier concentration causes a decrease in the globule diameter PS in the
119 TEO-loaded Negs. High emulsifier (above 5.5%) resulted in globules measuring below 200 nm, and low
120 emulsifier (below 5.5%) produced globules above 200 nm. This fact is in line with other studies that
121 increasing emulsifiers could reduce droplet Negs PS (41,42). The emulsifier reduced the interfacial surface
122 tension between the water and oil phases, which decreased the free energy required to disrupt or break the
123 globules and resulted in a smaller droplet diameter. It can also produce a protective cover over the globules,
124 preventing them from coalescence. However, the emulsifier must absorb quickly enough around the droplet
125 to form this protective layer (42,43).

126 Meanwhile, Carbopol® 980 had no significant effect on PS. The same findings were obtained from the
127 globule size Negs from Carbopol® 934 and 940 as gel former (19). A response surface plot (**Figure 2A**) may
128 therefore be used to represent the combined influence of variables X_1 and X_2 , which shows that Y_1 changes
129 linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with
130 span-80/tween-80 (X_1) - not Carbopol® 980 (X_2) - was evidenced from the comparative plot of the response
131 surface. From this explanation, it can be concluded that the PS can be changed by selecting the right X_1 level.

In **Table 1**, Negs PDI (Y_2) varied from 0.000 (STD#2 and #4) to 0.571 (STD#1, #5, #6, #8, #9, #11 to #14). PDI measures the distribution of molecular mass in a sample. The smaller PDI (close to 0), the more stable the Negs formulation caused; the large PDI indicates particles formed are not uniform, and the formulation will flocculate quickly. An index value less than 0.05 is included in monodisperse, while an index greater than 0.7 indicates that the sample has a broad PS distribution. A-0.2 and below are considered acceptable for nanoparticle preparations (44). Based on **Table 2**, the PDI response shows a non-significant quadratic model with an F-value of 1.02 (p-value 0.4645 > 0.05). Measurements between variables and responses are not a precise cause of that. The suggested quadratic model equation can be seen as follows:

$$Y_2 = 0.57 + 0.027X_1 - 0.017X_2 - 0.034X_1X_2 - 0.16X_1^2 - 0.086X_2^2$$

The equation and p-value of each factor did not significantly affect PDI. The response surface plot (**Figure 2B**) depicts the combined influence of variables X_1 and X_2 , which shows that Y_2 changes with the sum of the two variables by model.

The Negs ZP (Y_3) was in the range of 13.90 mV (STD#2) to 46.72 mV (STD#10) (**Table 1**). ZP represents the electric charge between the shear plane of a final outer layer and bulk solution, which significantly affects dispersion stability (45). This factor is strongly influenced by the composition of the Negs and its electrical phenomena. TEO-loaded Negs, which have positive ZP, show good interaction with negatively charged skin (46). ZP is the scientific term for the electrokinetic potential in colloidal systems. The high electric charge on the nanoparticle surface will prevent nanoparticle aggregation because of the strong repulsion between particles. The ZP requirement for stability is above ± 30 mV (19). The higher ZP and the slower aggregation are formed to prevent separation (47). Based on **Table 2**, the ZP response indicates a significant 2FI model with an F-value model of 7.06 (p-value 0.0078 < 0.05). The linear model equation suggested by the software can be seen as follows:

$$Y_3 = 31.54 + 8.27X_1 + 2.95X_2 + 10.59X_1X_2$$

The equation shows that the Negs ZP was significantly affected by the emulsifier (X_1) and gel former-emulsifier interaction (X_1X_2). ZP was usually influenced by the physicochemical properties of the drug, polymer, carrier, electrolyte presence, and their adsorption (48). One study stated that adding Carbopol® only slightly increased the ZP Negs (19). The response surface plot (**Figure 2C**) depicts the combined influence of variables X_1 and X_2 , showing that Y_3 changes with the sum of the two variables. Nevertheless, the higher gradient in the response surface with span-80/tween-80 (X_1) - not Carbopol® 980 (X_2) - is the evidence from the comparative plot of the response surface. This description concludes that the ZP can be changed by selecting the proper X_1 level.

The pH test was carried out to measure Negs's acidity or alkalinity level. The pH values (Y_4) were in the range of 4.57 (STD#13) to 6.39 (STD#14) (**Table 1**). The pH requirement of Negs is the same as the skin pH. Too-acidic preparations can irritate the skin and cause a stinging sensation, while too-alkaline preparations can cause dry and itchy skin. The pH result test on 14 formulations was eligible and compatible with the skin (4.5-6.5) (49). Based on **Table 2**, a linear model was found to be significant in pH response with a model F-value of 145.08 (p-value < 0.0001). The linear model equation suggested by the software can be seen as follows:

$$Y_4 = 5.48 - 0.069X_1 - 0.75X_2$$

The equation shows that the pH was significantly (p-value < 0.0001) affected by gel former (X_2). Carbopol is a high molecular weight homopolymer and acrylic acid copolymer crosslinked with polyalkenyl polyethers (50). They are anionic and acidic (2.5-4.0 in 2% dispersion) when not neutralized with bases to achieve a specific viscosity (50,51). Therefore, adding Carbopol to a formulation with a fixed amount of base (triethanolamine) will significantly lower the pH of Negs. The response surface plot (**Figure 2D**) may then be used to depict the combined influence of variables X_1 and X_2 , which shows that Y_4 changes linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with Carbopol® 980 (X_2) - not span-80/tween-80 (X_1) - is the evidence from the comparative plot of the response surface. This description concludes that the choice of the X_2 level affects the pH Negs.

Spreadability was measured to ensure comfortable use on the skin because it spreads quickly (24). Terms of good dispersion are 5-7 cm. If the dispersion is too small, it is relatively difficult to spread when applied to the skin, while the dispersion tends to spread too quickly when applied, so it will cause an uncomfortable feeling when used (52). Based on the results, only one Neg did not meet the requirements, namely F11. **Table 1** shows the range of spreadability (Y_5) from 4.90 cm (STD#11) to 6.25 cm (STD#10). Based on **Table 2**, the spreadability response indicates a significant linear model with an F-value of 386.43 (p-value < 0.0001). The linear model equation suggested by the software can be seen as follows:

$$Y_5 = 5.58 + 0.53X_1 - 0.033X_2$$

The equation shows that the spreadability of Negs was significantly (p -value < 0.0001) positively affected by the emulsifier (X_1). That was also found in other literature studies (24). The higher the Carbopol-contained Negs, the more viscous Negs. AT and spreadability have the opposite results. The higher the Negs' viscosity, the higher the adhesive strength produced, while the smaller the dispersion power (53). The response surface plot (**Figure 2E**) depicts the combined influence of variables X_1 and X_2 , which shows that Y_5 changes linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with span-80/tween-80 (X_1) - not Carbopol® 980 (X_2) - is the evidence from the comparative plot of the response surface. This description concludes that the scatter can be changed by choosing the right X_1 level.

Topical dosage forms, such as Negs, adhere to the skin in two ways: they adhere directly to the rough surface to form a "mechanical interlock" and to the surface via interaction (54). Good adhesion to Negs supports a higher concentration gradient towards the skin and provides more drug penetration (18). Adhesive strength is directly related to the AT on the Negs as measured using the single-lap shear test method with slight modifications (55,56). The test was carried out by applying a shear load to the plates that flank the sample, which had been pre-loaded, and given a measured force; the time taken for the plates to separate was recorded as AT. In our study, the AT was from 4.22 s (STD#14) to 7.08 s (STD#13) in **Table 1**. They met the requirements based on the AT test results on a 14 formulation. An AT was carried out to see how long a Negs could be attached to the skin. The AT requirement is more than 4 seconds. The longer a Negs could be attached to the skin, it showed the better result, where it is expected that more active substances can be absorbed due to the time the Negs was in contact with the skin (57). Based on **Table 2**, the linear model was found to be significant in the AT response with an F-value model of 46.35 (p -value < 0.0001). The linear model equation suggested by the software can be seen as follows:

$$Y_6 = 5.74 - 0.20X_1 + 1.18X_2$$

The equation reveals that the AT was significantly (p -value < 0.0001) affected by the gel former (X_2) or, indirectly, the same adhesion strength. This finding is in agreement with the literature (58,59). A response surface plot (**Figure 2F**) may therefore be used to depict the combined influence of variables X_1 and X_2 , which shows that Y_6 changes linearly with the sum of the two variables. Nevertheless, the higher gradient in the response surface with Carbopol® 980 (X_2) - not span-80/tween-80 (X_1) - is the evidence from the comparative plot of the response surface. From this explanation, it can be concluded that the AT can be changed by selecting the right X_2 level. Details of the ANOVA results for measured responses are also presented in **Table 2**. In the end, the emulsifier factor significantly affected the response of PS, ZP, and spreadability of TEO-loaded Negs. At the same time, gel former affects the AT and pH.

2.4. Optimized TEO-loaded Negs

The formulation was optimized with Design-Expert®, version 13 software. The optimized Negs were selected based on the minimum PS and PDI; maximum ZP; value in pH range, spreadability, and AT (**Table 3**). Variables composition for optimized Negs is span-80/tween-80 of 8.68% and Carbopol® 980 of 1.18% with a desirability value of 0.801. The formulation with the maximum desirability value is the optimal formulation generated from the optimization phase of the program (60) – the optimization value formed as indicated by the desirability value close to one.

The desirability value range is 0-1. **Figure 2** describes the optimization results in the form of a 2D contour. *Contour* is a two-dimensional response image that was presented using a predictive model for PS, PDI, ZP, pH, spreadability, and AT response values. The contour graph shows the desirability value of 0.801, which is the closest value to 1 compared to the other points. **Figure 3** shows the projection in the form of a 3D surface; the low area shows low desirability, while the high area shows high desirability and is getting closer to 1. At this stage, the software predicts the response values shown in **Table 4**. Three confirmation runs need to be performed to validate optimization (61). The optimization model and estimates are validated by the observed optimized Negs, which show an acceptable variation from the predicted values (**Table 4**). We tested the optimized Negs' physical properties for further investigation, such as organoleptic, homogeneity, freeze-thaw, viscosity, and flow properties.

2.5. Evaluation of TEO-loaded Negs

Nanoemulsion systems can cover oily drugs' bitter or unpleasant taste (62). The organoleptic results of TEO-loaded Negs have a less distinctive turmeric odor, which is white and semisolid (**Table 5**). That is due

239 to the drug entrainment of oil with the oil phase effectively preventing evaporation and masking its specific
240 odor (63). The homogeneity test results of the optimum TEO-loaded Negs formulation showed a
241 homogeneous preparation, as evidenced by the absence of coarse grains. This homogeneity was correlated
242 with the optimal formulation of PS and PDI of 182.3 ± 5.5 and 0.242 ± 0.003 , respectively. The low PDI
243 indicates uniformity or homogeneous dispersion of globules Negs (64). In addition, the small size of the
244 globule (± 200 nm) is not included in the coarse dispersion (41).

245 The thermodynamic stability test of the system was carried out using a freeze-thaw cycle to identify
246 the presence of metastable Negs in the optimal formulation. It aims to see the separation of the water and oil
247 phases due to the influence of extreme temperatures (65). The thermodynamic stability of any system is
248 determined by the change in free energy between the system and its surroundings (66). The test results on
249 the optimum formula for six cycles showed promising results; namely, there was no separation. This
250 stability was correlated with the ZP of the optimal formulation of 57.23 ± 2.91 mV. The surface charge's
251 magnitude was directly related to the stability of any Negs. It is evidenced by the high repulsive force
252 between the Negs globules preventing coalescence, which was characterized by the absence of phase
253 separation (67). Similar results were found in the Negs study containing thymoquinone, which had ZPs
254 between -26.7 and -30.6 mV (66).

255 The pH conditions indirectly affect the viscosity indicated by Negs because they influence the
256 swelling ability of Carbopol® 980. This excipient is a gel former and a thickener (52). It plays an essential role
257 in the viscosity of Negs. Carbopol is dispersed in water to form an acidic colloidal solution with a low
258 viscosity. Neutralizing with triethanolamine increases Negs' viscosity because a stable water-soluble gel was
259 formed (50). Viscosity was carried out with #7 spindle (Brookfield digital RV DV-E) at 50 rpm of $32240 \pm$
260 2257.7 cP, indicating significantly high viscosity on Negs with pH 4.5. The magnitude of the viscosity is
261 correlated with AT and spreadability. Viscosity is inversely proportional to spreadability (52). In contrast,
262 the AT is directly proportional to the viscosity. A high-viscosity system will form stronger interfacial
263 interactions and increase intermolecular interactions in the polymer network, increasing cohesion, adhesion
264 strength, and AT (54).

265 Determining the rheology of a semisolid preparation is essential for controlling the consistency
266 required to ensure the performance and formulation durability and to describe the mechanical (flow
267 properties) system (68). The rheological study was conducted in the shear rate range of 6.81 – 40.86 s⁻¹ at 25°C.
268 The consistency index equals the apparent viscosity at a shear rate of 1 s⁻¹. The consistency index measured
269 on TEO-loaded Negs was 155.67 cP and $n = 0.22$. The flow index measures the system's deviation from
270 Newtonian behavior ($n = 1$). A value of $n > 1$ indicates dilatation or shear thickening flow, and $n < 1$
271 indicates pseudoplastic or shear thinning. The flow index typically lowers the thicker the base. Negs
272 produce a 0.22 flow index, which implies pseudoplastic flow behavior. A colloidal network structure aligned
273 with the shear direction and decreases viscosity as the shear rate increases have led to this pseudo-plasticity.
274 The developed system will require a specific force to discharge (69). The results of the flow properties test
275 showed that the optimal formula made was a pseudoplastic thixotropic flow type (**Figure 4**). *Thixotropic* is a
276 flow property expected in pharmaceutical preparations because it has high consistency in the container but
277 can be poured and dispersed easily (70).

278 3. CONCLUSION

279 Based on the results of the CCD-RSM analysis, the optimum span-80/tween-80 as an emulsifier is
280 8.68%, and Carbopol® 980 as a gel former was 1.18%. The resulting response is a PS of 182.3 nm, PDI 0.242,
281 ZP 57.23 mV, pH 4.51, AT 6.45 seconds, and spreadability of 6 cm. Optimized formulation viscosity is 32240
282 cP with pseudoplastic thixotropic flow properties. Thus, the developed TEO-Negs can be a potential
283 delivery system and a promising suitable approach for topical preparations.

284 4. MATERIALS AND METHODS

285 4.1. Materials

286 TEO (*Curcuma longa*) was purchased from Darjeeling Sembrani Aroma (Indonesia), sorbilene O E/P from
287 Lamberti (Italy), span-80 from Croda (Singapore), propylene glycol from Dow Chemical Pacific (Singapore),
288 Carbopol® 980 NF from Lubrizol AM (Cleveland), nipagin M from Clariant Produkte (Deutschland),
289 propylparaben from Alpha Chemika (India), triethanolamine from Dow Chemical Pacific (Switzerland), and
290 1.1-diphenyl-2-picrylhydrazyl (DPPH) from Smart-Lab (Indonesia).

291 **4.2. Methods**

292 *4.2.1. Identification of TEO*

293 Organoleptic tests include observing form, color, and odor. Phenolic identification was carried out by
294 adding one drop of 5% FeCl₃ to a 1 ml sample. Dark green to black colors indicate phenolic compounds'
295 presence (71). Terpenoid testing was performed by adding Lieberman-Burchard reagent containing
296 anhydrous acetic acid and concentrated sulfuric acid (3:1) into a 1 ml sample. Brownish or violet ring form
297 indicates the presence of terpenoids (72). Spectrophotometry was used to determine antioxidant activity
298 with the DPPH method (5). The calibration curve for the DPPH concentration against absorbance was made
299 at a maximum wavelength of 516 nm. The absorbance was measured in a mixture of sample solution and
300 DPPH with a particular concentration after 30-min of incubation in a dark room. IC₅₀ was calculated from
301 the inhibition percentage and absorbance.

302 *4.2.2. Preparation of TEO-loaded Negs*

303 TEO-loaded Negs were produced using a high-energy method, which used a mechanical device to produce a
304 highly disruptive force to break up the water and oil phases to obtain nano-sized globules (18). The oil phase
305 (M1) was prepared by mixing span-80 with 5% turmeric oil using a magnetic stirrer (WiseStir Wisd) at 1,500
306 rpm for 20 min. A total of 0.18% methylparaben and 0.02% propylparaben dissolved in 15% propylene glycol
307 (M2). Then, the distilled water was stirred with tween-80, and M2 was added gradually until homogeneous
308 at 1,500 rpm for 20 min (M3). M1 was stirred with M3 until homogeneous at 1,500 rpm for 40 min to form a
309 clear and transparent nanoemulsion, then let left for 24 hours. The gel base was prepared by mixing
310 Carbopol® 980 NF with distilled water at 70°C and left for 24 hours. Then, gradually add 1% triethanolamine
311 to form a gel mass. Nanoemulsion was added slowly into the gel base while homogenized using a
312 homogenizer (AEG) at 2,000 rpm for 10 min.

313 *4.2.3. Experimental Design*

314 This study selects the CCD-RSM method to develop the TEO-loaded Negs formulation. For preliminary
315 screening on PS, PDI, ZP, pH, AT, and spreadability, a 2-factor CCD-RSM at two levels (high and low) was
316 used. Based on previous experiments and study literature, high and low variables were determined. The
317 CCD of the statistical package Design-Expert® version 13 software (Stat-Ease Inc., Minneapolis, MN) was
318 used to examine the influence of the specified independent variable on the response variable to obtain the
319 optimal formulation for TEO-loaded Negs. CCD planned 14 experiments were done under controlled
320 circumstances (Table 3). The independent variables were span-80/tween-80 (X₁) and Carbopol® 980
321 concentration (X₂). The observed response of the dependent variables was PS (Y₁), PDI (Y₂), ZP (Y₃), pH (Y₄),
322 spreadability (Y₅), and AT (Y₆).

323 *4.2.4. Determination of PS, PDI, and ZP of Negs*

324 PS and PDI of Negs were assessed using the Delsa Max Pro Particle Size Analyzer LS 100Q (Beckman
325 Coulter, USA) at 25°C utilizing the dynamic light scattering (DLS) method or photo correlation spectroscopy.
326 For analysis, 1 ml of samples was dispersed in 9 ml aqua pro injection. Into the cuvette, 1 ml of suspension
327 and 5 ml of aqua pro injection were added as a diluent, and the results were read on the instrument. All
328 measurements were made at a scattering angle of 90°. ZP was determined by particle size analyzer through
329 mobility and conductivity measurements. The temperature was set to 25°C, and the mean electric field was
330 set to 16 V/cm (73). The final result is the mean of each sample's three repeated measurements.

331 *4.2.5. Determination of pH, Spreadability, and AT*

332 pH was measured at a temperature of 25°C using a pH meter that had been previously calibrated with buffer
333 solutions of pH 4 and 7. The calibration step was completed when the pH value indicated on the screen
334 matched the correct pH value and was stable. Afterward, the electrode was dipped in Negs and recorded the
335 value shown on the screen (74). Spreadability was measured by adding 0.5 g of Negs in the center of a glass
336 covered with another glass. The preparation diameter was measured longitudinally and transversely; for

337 every minute added, 50 g was to a total weight of 150 g (75). Adhesion time determine by the single-lap
338 shear test method (54). A-0.5 g of Negs was placed on a slide, then covered with another slide, and given a
339 load of 1 kg for 3 min. The glass object was mounted on the test apparatus, and 80 g of the load was released
340 until both glass objects were released, and the time was recorded (75).

341 4.2.6. Organoleptic Observation and Homogeneity Test

342 Negs were placed in a glass object and directly observed for color, smell, and shape (76). A-0.1 g Negs were
343 spread over the slide, and homogeneity was observed. Test preparation is declared homogeneous if no
344 coarse grains exist (77).

345 4.2.7. Viscosity and Rheological Flow

346 Viscosity and rheology were determined with a Brookfield RV DV-E Viscometer with appropriate spindle
347 and speed. A-500 ml of Negs were put into a beaker glass; the spindle was installed, and the measured value
348 was recorded as viscosity Negs. In this study, the spindle used spindle no. 7. Flow properties were
349 determined by measuring the viscosity using a right spindle from low to high rotational speed and vice
350 versa (74). Flow index and consistency index are determined from the power law equation:

$$351 \tau = K r^n$$

352 where τ is the shear stress, K is the consistency index, r is the shear rate, and n is the flow index.

353 Taking logs on both sides,

$$354 \log \tau = \log K + n \log r$$

355 So, from the log, shear stress Vs. Log shear rate plot, the plot slope was used as the flow index and the
356 antilog of the Y-intercept as the consistency index (69).

357 4.2.8. Freeze-thaw Test

358 Negs were stored at 4 ± 2 °C, then transferred to 40 ± 2 °C for 48 hours (1-cycle), then repeated for 6-cycles.
359 Phase separation was observed in each cycle (74).

360 4.2.9. Statistical Analysis

361 The Design-Expert® version 13 software was used to conduct the statistical study (Stat-Ease Inc.,
362 Minneapolis, MN). Analysis was done at a sig – $p < 0.05$ and $p < 0.01$ after three times of each measurement.

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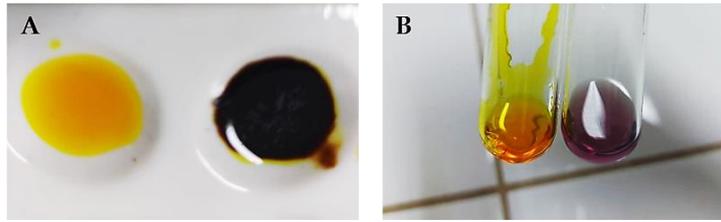
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542 **Figure 1.** The result of qualitative test observation is a phenolic test with a positive result marked in a blackish green
543 color (a) and a terpenoid test with a positive result marked with a purple color (b).

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Table 1. Evaluation results of independent variables and dependent variables with the CCD design for optimizing TEO-loaded Negs.

Std	X ₁ (%)	X ₂ (%)	Y ₁ (nm)	Y ₂	Y ₃ (mV)	Y ₄	Y ₅ (cm)	Y ₆ (s)
1	5.50	1.02	206.9 ± 0.7	0.571 ± 0.000	35.39 ± 1.50	5.45	5.55	5.65
2	2.32	1.15	457.0 ± 0.9	0.000 ± 0.000	13.90 ± 1.92	4.60	5.00	6.99
3	8.68	0.90	166.7 ± 0.4	0.378 ± 0.074	31.64 ± 0.53	6.35	6.20	4.23
4	2.32	0.90	411.0 ± 5.8	0.000 ± 0.000	30.67 ± 0.89	6.29	5.00	4.30
5	5.50	1.02	302.1 ± 3.6	0.571 ± 0.000	36.45 ± 0.45	5.46	5.60	5.63
6	5.50	1.02	263.0 ± 3.7	0.571 ± 0.000	18.49 ± 1.77	5.47	5.55	5.64
7	8.68	1.15	198.1 ± 1.2	0.242 ± 0.003	57.23 ± 2.91	4.65	6.15	6.95
8	5.50	1.02	246.9 ± 7.2	0.571 ± 0.000	26.60 ± 0.17	5.46	5.60	5.67
9	5.50	1.02	206.0 ± 2.3	0.571 ± 0.000	31.77 ± 1.01	5.46	5.60	5.65
10	10.00	1.02	160.8 ± 1.3	0.285 ± 0.024	46.72 ± 1.97	5.31	6.25	5.84
11	1.00	1.02	244.2 ± 4.3	0.571 ± 0.000	31.26 ± 0.57	5.78	4.90	6.88
12	5.50	1.02	238.1 ± 3.0	0.571 ± 0.000	30.78 ± 0.53	5.45	5.55	5.64
13	5.50	1.20	206.2 ± 1.6	0.571 ± 0.000	30.56 ± 0.78	4.57	5.50	7.08
14	5.50	0.85	240.5 ± 2.2	0.571 ± 0.000	20.12 ± 0.16	6.39	5.65	4.22

Y₁: Particle size (PS), Y₂: Polydispersity index (PDI), Y₃: Zeta potensial (ZP), Y₄: pH, Y₅: Spreadability, Y₆: Adhesion time (AT)

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551 **Table 2.** Statistical analysis of PS (Y₁), PDI (Y₂), ZP (Y₃), pH (Y₄), spreadability (Y₅), and AT (Y₆) TEO-loaded Negs on
552 CCD.

Factors		Y ₁	Y ₂	Y ₃	Y ₄	Y ₅	Y ₆
X ₁	Coefficient	-77.64	0.027	8.27	-0.069	0.53	-0.20
	p-value	0.0063**	0.7325	0.0080**	0.1430	<0.0001**	0.1406
X ₂	Coefficient	3.61	-0.017	2.95	-0.75	-0.033	1.18
	p-value	0.8783	0.8289	0.2668	<0.0001**	0.1157	<0.0001**
X ₁ X ₂	Coefficient		-0.034	10.59			
	p-value		0.7602	0.0136*			
X ₁ ²	Coefficient		-0.16				
	p-value		0.0818				
X ₂ ²	Coefficient		-0.086				
	p-value		0.3087				
Intercept	Coefficient	253.39	0.57	31.54	5.48	5.58	5.74
Degree of freedom		2	5	3	2	2	2
Sum of squares		48332.05	0.24	1065.33	4.48	2.28	11.49
Mean of squares		24166.02	0.047	355.11	2.24	1.14	5.74
F-value		5.69	1.02	7.06	145.08	386.43	46.35
p-value		0.0201	0.4645	0.0078	<0.0001	<0.0001	<0.0001
R-Squared		0.5084	0.3898	0.6794	0.9635	0.9860	0.8939

553 X₁: span-80/tween-80; X₂: Carbopol® 980

554 * p-value < 0.05

555 ** p-value < 0.01

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558 **Table 3.** The independent and dependent variables with levels and limits in CCD for TEO-loaded Negs development.

Variables	Code	Start point (- α)	Low level (-1)	Central level	High level (+1)	Start point (+ α)	Units	
Independent variables								
Span-80/tween-80	X ₁	0.85	2.32	1.02	5.50	1.20	% w/w	
Carbopol® 980	X ₂	1	0.90	5.50	1.02	10	% w/w	
Dependent variables								
Limits								
Particle size (PZ)	Y ₁	Minimum						nm
Polydispersity index (PDI)	Y ₂	Minimum						
Zeta potensial (ZP)	Y ₃	Maximum						mV
pH	Y ₄	is in range						
Spreadability	Y ₅	is in range						cm
Adhesion time (AT)	Y ₆	is in range						s

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Table 4. Optimized TEO-loaded Negs' actual and predicted values for each response.

Responses	Predicted values	Actual value ^a	Error ^b (%)
Y1 (nm)	180.2	182.3 ± 5.5	1.165
Y2	0.250	0.242 ± 0.003	-3.200
Y3 (mV)	56.30	57.23 ± 2.91	1.652
Y4	4.50	4.51 ± 0.02	0.222
Y5 (cm)	6.07	6.0 ± 0.2	-1.153
Y6 (s)	6.98	6.45 ± 0.19	-7.593

a Data listed is the mean ± standart deviation, n = 3

b Error (%) = [(Actual value - Predicted value)/Predicted value] * 100%

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Table 5. Additional evaluation on TEO-loaded Negs with an optimal formulation.

Evaluation	Result
Organoleptic	Color: White; Odor: typical turmeric
Homogeneity	Homogeneous
Freeze-Thaw	Cycles 0-6, no separation occurs
Viscosity ^a	32240 ± 2257.7 cP

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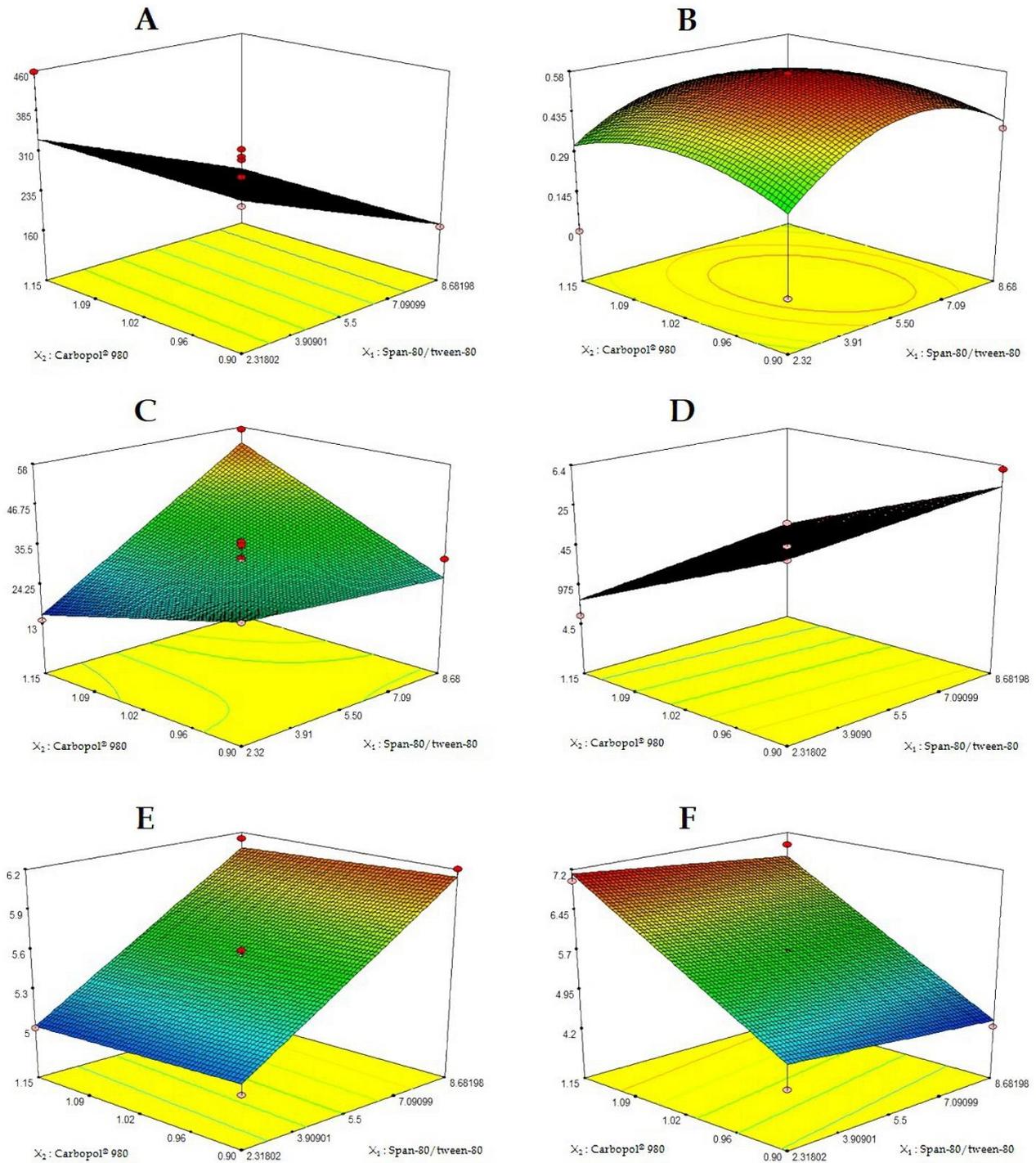
^a Data listed is the mean ± standart deviation, n = 3

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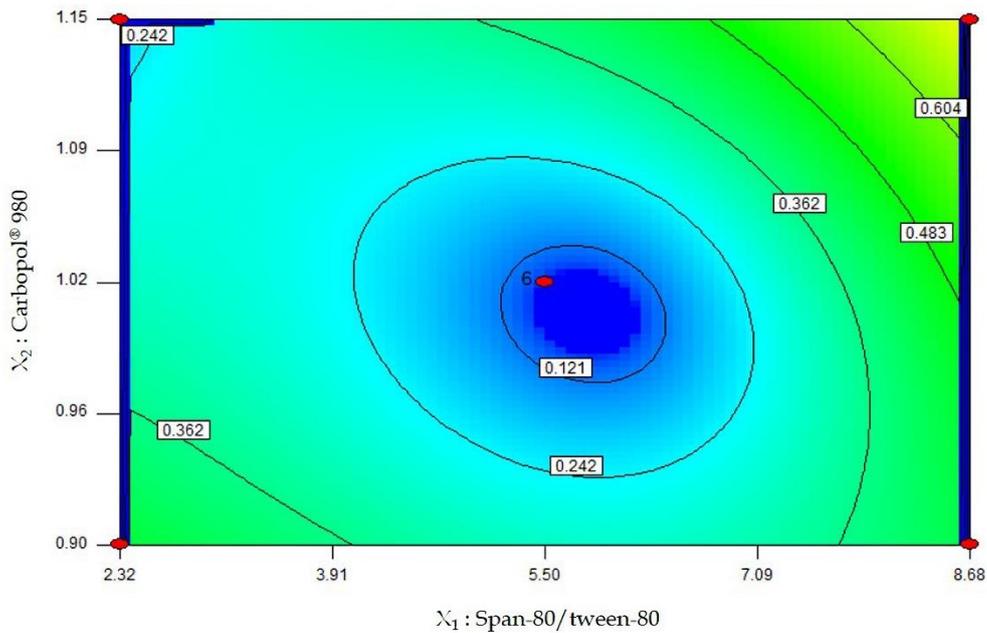
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Figure 2. Effect of span-80/tween-80 and Carbopol[®] 980 concentration on PS (a) PDI (b) ZP (c) pH (d) spreadability (e) AT (f).

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Figure 3. Contour plot desirability value of optimal formulation.

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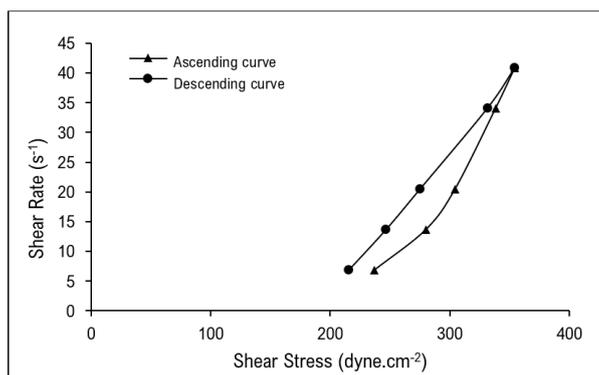


Figure 4. Optimal formulation flow properties.

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Manuscript Information**Manuscript ID:** MPJ-10622.REV-3**Title in English:** Response surface methodology for optimization of turmeric essential oil-loaded nanoemulgel**Small Title in English:** No information entered**Authors:** Nining Nining¹, Anisa Amalia¹, Fatimatuz Zahrok²**Institutions:** ¹Universitas Muhammadiyah Prof. DR. HAMKA, Pharmaceutical Technology, East Jakarta, Indonesia²Universitas Muhammadiyah Prof. DR. HAMKA, Pharmacy, East Jakarta, Indonesia**Keywords in English:** Central composite design; nanoemulgels; response surface methodology; turmeric essential oil; topical delivery.**Manuscript Type:** Research article**Processing Status:** Accepted**Abstract in English**

Turmeric Essential Oil (TEO) has an antioxidant and anti-inflammatory activity to be formulated in a topical dosage form. Nanoemulgels (Negs) development, based on varying concentrations of emulsifiers and gel formers, affects their characteristics and stability. This study focuses on optimizing TEO-loaded Negs based on physical and mechanical characterization, which have promising topical applications. Negs were created using the high-energy approach and optimized using Response Surface Methodology (RSM) and the Central-Composite Design (CCD) for the optimization of span-80/tween-80 (X1) and Carbopol® 980. (X2). Observed variable responses were particle size (PS) (Y1), polydispersity index (PDI) (Y2), zeta potential (ZP) (Y3), pH (Y4), spreadability (Y5), and adhesion time (AT) (Y6). Actual responses of Negs were compared with the CCD-RSM predictions to validate the model. In addition, other physical evaluations were observed, such as organoleptic observations, homogeneity, freeze-thaw tests, viscosity, and flow properties. Optimized TEO-loaded Negs were made with 8.68% span-80/tween-80 and 1.18% Carbopol® 980. The evaluation results showed the optimal TEO-loaded Negs on nano-metric size (182.3 ± 5.5 nm) with low PDI (0.242 ± 0.003), good ZP (-57.23 ± 2.91 mV), pH (4.51 ± 0.02), spreadability (6.0 ± 0.2 cm), and AT (6.45 ± 0.19 s). TEO-loaded Negs have an excellent appearance and did not run phase separation at extreme temperature storage with pseudoplastic thixotropy flow. Thus, the developed TEO-loaded Negs can be a potential delivery system and a promising suitable approach for topical preparations.

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