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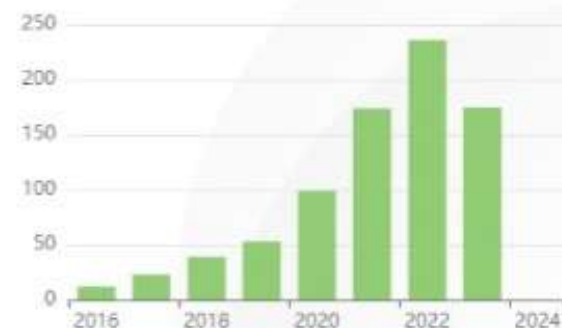
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THE INFLUENCE OF WATER TEMPERATURE ON SEAWATER EVAPORATION IN THE DESALINATION PROCESS

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ABSTRAK

Peningkatan populasi di suatu daerah berdampak pada meningkatnya permintaan akan air bersih. Meskipun air melimpah, sekitar 97% air di Bumi berupa air laut yang tidak dapat dikonsumsi secara langsung karena mengandung zat berbahaya. Proses desalinasi digunakan untuk mengubah air laut menjadi air minum dengan memanfaatkan panas dan kecepatan angin. Penelitian ini mengeksplorasi suhu 30°C, 45°C, 60°C, dan 70°C dengan kecepatan angin sebesar 2,6 m/s. Pengumpulan data dilakukan untuk setiap variasi suhu selama 2 jam dengan catatan setiap 15 menit, bertujuan untuk mengevaluasi efektivitas evaporasi dan kondensasi dalam proses desalinasi. Hasil penelitian menunjukkan bahwa suhu 70°C menghasilkan penguapan tertinggi (800 gram), meskipun dengan tingkat kondensasi sebesar 26,25%. Pengukuran kondensat menunjukkan pH sebesar 7,2 dan pembacaan meter TDS berkisar antara 125 hingga 138 ppm.

Kata kunci: desalinasi; konversi air laut; efektivitas penguapan; laju kondensasi.

ABSTRACT

Population An increase in population in an area results in an increased demand for clean water. Although water is abundant, about 97% of the water on Earth is seawater, which cannot be consumed directly because it contains harmful substances. Desalination process is used to convert seawater into drinking water by utilizing heat and wind speed. This research explores temperatures of 30°C, 45°C, 60°C, and 70°C with a wind speed of 2.6 m/s. Data was collected for each temperature variation for 2 hours with records every 15 minutes, aiming to evaluate the effectiveness of evaporation and condensation in the desalination process. The results showed that 70°C resulted in the highest evaporation (800 grams), although with a condensation rate of 26.25%. Condensate measurements showed a pH of 7.2 and TDS meter readings ranged from 125 to 138 ppm.

Keywords: desalination; seawater conversion; evaporation effectiveness; condensation rate.

1. PENDAHULUAN

Pertumbuhan penduduk di suatu daerah [1-3] menyebabkan kebutuhan air juga bertambah. Ketersediaan air bersih terbatas [4-6] tetapi sangat penting untuk menopang kehidupan yang layak bagi manusia ataupun

makhluk hidup lainnya [7] karena air merupakan sumber [8] atau elemen penting [9] bagi kehidupan makhluk di muka bumi termasuk di dalamnya manusia. Banyak sekali kegiatan manusia [10] yang menjadikan air sebagai sumber utama seperti minum, mencuci, memasak dan kegiatan-kegiatan lainnya. Air

kerap kali mudah untuk ditemukan dikarenakan sumber air tersebut dapat diperoleh dari tanah, mata air, air sungai, danau dan air laut. Namun tidak semua air dapat dikonsumsi dikarenakan ada beberapa kandungan zat-zat berbahaya [11,12] yang ada di dalam air tersebut. Selain itu 97% air di muka bumi [13,14] merupakan air laut yang tidak dapat digunakan oleh manusia secara langsung. Terbatasnya air yang tersedia mendorong banyaknya penelitian untuk mengubah air laut menjadi air tawar [15,16] dengan cara proses distilasi [17-26]. Proses distilasi ini mengubah air laut menjadi air tawar dengan menggunakan panas dengan menggunakan tenaga matahari [27-39] sehingga air menjadi layak dikonsumsi [40].

Mengonsumsi air bersih yang berasal dari air laut sudah dilakukan di pesisir pantai [41] dan beberapa peneliti juga sudah menguji hasil destilasi air laut [42]. Karena panas yang dihasilkan matahari tidak dapat dikendalikan beberapa peneliti lain menggunakan pemanas [43-46]. Panas akan meningkatkan temperatur air yang juga meningkatkan tekanannya sampai akhirnya terjadi penguapan. Penguapan pada dasarnya terjadi karena adanya perbedaan tekanan antara permukaan air dengan ruang di atasnya [47] sehingga aliran udara yang bergerak di atas air juga akan mempengaruhi laju penguapan, meskipun dengan temperatur yang tetap [48].

Sejauh ini belum ada penelitian yang menggunakan variasi temperatur dengan kecepatan angin tertentu dan mengukur penguapannya. Oleh karena itu penelitian ini bertujuan menentukan pengaruh temperatur terhadap efektivitas pengolahan air laut menjadi air tawar dengan kecepatan angin yang tetap.

2. METODE PENELITIAN

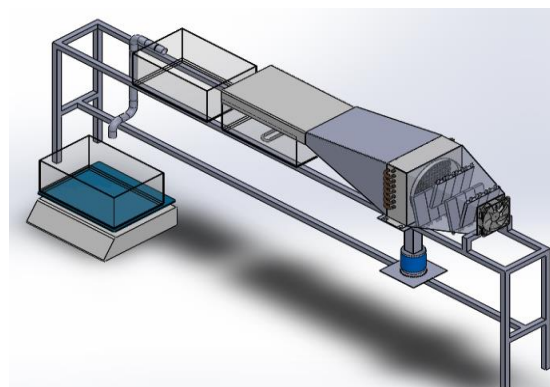
Jenis penelitian ini merupakan penelitian eksperimental menggunakan prototipe destilasi air laut [49]. Pada penelitian ini air laut dinaikkan temperaturnya menggunakan pemanas dengan kecepatan angin 2.6 m/s. Temperatur yang digunakan ialah 30°C, 45°C, 60°C, 70°C.

Data diambil setiap per-15 menit selama jangka waktu 2 jam untuk setiap variasi suhu yang digunakan. Panas yang dihasilkan berasal dari elemen heater yang dikendalikan temperaturnya dengan menggunakan pengontrol temperatur. Aliran angin dihasilkan oleh kipas yang menarik uap air hasil penguapan di ruang penguapan. Untuk mengukur tingkat kelembaban udara, temperatur dan kecepatan angin maka digunakan alat alat pada tabel 1.

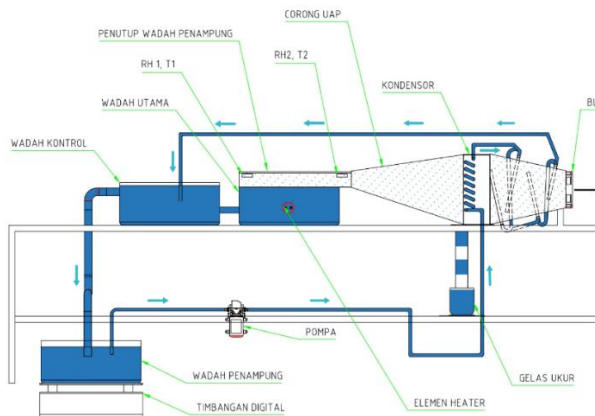
Tabel 1. Alat Penelitian

No	Alat	Spesifikasi
1	Thermostat XH-W3001	Rentang -50°C - 110°C, 0.1°C
2	Anemometer GM816	Rentang 0 – 30 m/s, 0,1 m/s,
3	Thermometer digital	Rentang -50°C - 110°C, ±0.1°C
4	Hygrometer digital	Range 10% - 99%, ±1%
5	Timbangan digital 40 kg	Beban maksimum 40 kg, resolusi 0.005 kg
6	Timbangan digital 5 kg	Beban maksimum 5 kg, resolusi 1 gr

Penelitian ini menggunakan alat seperti yang ditampilkan pada gambar 1 dan Skema yang ditampilkan pada gambar 2. Penelitian dilaksanakan di Laboratorium Konversi Energi, Fakultas Teknologi Industri dan Informatika Universitas Muhammadiyah Prof. Dr Hamka, Jakarta.



Gambar 1. Desain Alat Destilasi



Gambar 2. Skema Alat Destilasi

Penelitian dilakukan dengan mengalirkan air laut dari wadah penampung seperti ditampilkan pada gambar 3. Air laut dialirkan menuju wadah utama melalui kondensor dan dipanaskan menggunakan heater yang diatur dengan thermostat dengan 4 variasi suhu yaitu; 30°C, 45°C, 60°C dan 70°C dengan kecepatan angin tetap 2,6 m/s.



Gambar 3. Alat Destilasi

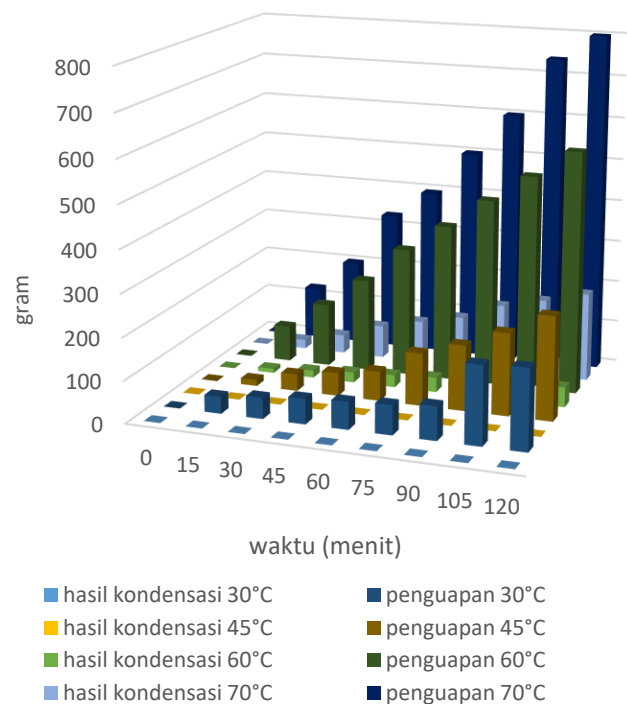
Berdasarkan hasil uji coba, peralatan yang digunakan bekerja dengan baik termasuk juga pemeriksaan kebocoran dan retak pada sambungan.

3. HASIL DAN PEMBAHASAN

Untuk mempermudah melakukan analisa maka data hasil pengujian dan perhitungan digambarkan dalam bentuk grafik [50]. Pengumpulan data dilakukan dengan variasi temperatur heater 30°C, 45°C, 60°C dan 70°C. Lalu, waktu pengambilan data ini diambil per-15 menit sekali selama dalam kurun waktu 2

jam tiap varian suhu panas yang dihasilkan oleh heater.

Sensor kelembaban dan temperatur digunakan untuk mengetahui temperatur dan kandungan uap air di udara yang terjadi karena proses penguapan [51]. Hasil pengujian ditampilkan menggunakan grafik pada Gambar 4.



Gambar 4. Grafik Penguapan dan Kondensasi pada variasi suhu yang diujikan

Dari gambar 4 terlihat terjadinya kondensasi dipengaruhi oleh energi panas [52,53] yang dihasilkan oleh heater untuk membuat air laut menguap. Berdasarkan hasil pengukuran menggunakan timbangan bahwa penguapan terjadi mulai dari temperatur 30°C hingga temperatur yang tertinggi 70°C. Meskipun semua variasi temperatur yang diujikan terjadi proses penguapan, hanya beberapa dari variasi temperatur yang diujikan menghasilkan kondensasi. Panas yang dikandung oleh uap air dilepaskan pada dinding kondensor [54]. Meskipun demikian energi panas yang dibawa oleh uap air tidak seluruhnya dapat dilepaskan di kondensor karena adanya aliran udara. Kondensasi total yang terjadi hanya 8,25% dari penguapannya

untuk temperatur 60°C dan 26,25% pada temperatur 70°C. Hasil tersebut menunjukkan bahwa sebagian besar uap air yang dihasilkan tidak berhasil dikondensasikan oleh kondensor yang digunakan karena penampang kondensor yang cukup luas memungkinkan uap air tidak bersentuhan dengan dinding sehingga temperatur uap tidak turun di bawah temperatur embunnya. Oleh karena itu desain kondenser perlu diperbaiki dan ditingkatkan performanya.

Selama 2 jam eksperimen dilakukan pada berbagai suhu (30°C, 45°C, 60°C, dan 70°C), semuanya menghasilkan penguapan. Namun, hanya suhu 60°C dan 70°C yang mengalami kondensasi, hasilnya ditunjukkan pada tabel 2.

Tabel 2. Hasil dari Tekanan Air dan Tekanan Udara di atas Air

Air		Udara	
Temperatur °C	Tekanan (p_a)	Temperatur °C	Tekanan (p_a)
30°C	22,67	30°C	4241,52
45°C	24,85	45°C	4813,44
60°C	26,83	60°C	5796,56
70°C	28,05	70°C	6544,78

Pengujian kandungan [55] pada air laut dan hasil kondensasi perlu dilakukan untuk mengetahui isi dari kandungan air laut [56] sebelum mengalami proses desalinasi dan setelahnya. Dari hasil pengukuran pH meter dan TDS meter diketahui air laut sendiri memiliki nilai pH 8 dan TDS meter 30320 PPM, sedangkan pada pengukuran sampel lainnya terkait pH dan nilai TDS meter ialah pada sampel 1 didapati pH 7.2/ TDS meter 125 PPM, sampel 2 didapati pH 7.2/ TDS meter 138 PPM. Dari hasil pengukuran pH meter dan TDS meter di atas menunjukkan bahwa sampel 1 dan sampel 2 dapat dikatakan sebagai air yang memenuhi syarat air bersih.

4. KESIMPULAN

Hasil eksperimen menyatakan selama 2 jam hasil pengujian di tiap variasi temperatur 30°C, 45°C, 60°C, dan 70°C semua menghasilkan penguapan. Tetapi hanya temperatur 60°C dan 70°C sajalah yang menghasilkan kondensasi. Berdasarkan tujuan penelitian ini yaitu mengetahui tingkat keefektifitan untuk menghasilkan penguapan dan kondensasi, dapat dipastikan temperatur 70°C menjadi temperatur yang paling efektif dalam menghasilkan penguapan dan kondensasi.

UCAPAN TERIMA KASIH

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THE INFLUENCE OF NOZZLE INJECTION PRESSURE ON SEAWATER EVAPORATION INSIDE AN EVAPORATOR TUBE

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ABSTRACT

Indonesia, the largest archipelagic country in the world, possesses a vast marine area. Despite being surrounded by the sea, many coastal communities in Indonesia lack access to clean water. Seawater distillation presents a viable solution to this scarcity. This process involves separating salt from seawater to produce fresh water. This study aims to analyze the effect of nozzle spray pressure on the evaporation process of seawater to optimize fresh water production. Experiments were conducted using a fogging nozzle with a diameter of 0.3 mm, varying the nozzle pressure at 40 psi, 70 psi, and 100 psi. The data were statistically analyzed to determine the impact of nozzle pressure on seawater evaporation. The results indicate that the highest evaporation occurred at a nozzle pressure of 40 psi, yielding 10 g of condensed seawater, whereas the lowest evaporation was observed at 100 psi, producing 4 g.

Keywords: Seawater; Distillation; Evaporation; Pressure.

ABSTRAK

Indonesia sebagai negara kepulauan terbesar di dunia mempunyai wilayah laut yang sangat luas. Meski dikelilingi laut, banyak masyarakat pesisir di Indonesia yang kekurangan akses terhadap air bersih. Distilasi air laut memberikan solusi yang tepat untuk mengatasi kelangkaan ini. Proses ini melibatkan pemisahan garam dari air laut untuk menghasilkan air tawar. Penelitian ini bertujuan untuk menganalisis pengaruh tekanan semprotan nozzle terhadap proses penguapan air laut untuk mengoptimalkan produksi air tawar. Percobaan dilakukan dengan menggunakan nozzle fogging berdiameter 0,3 mm, tekanan nozzle divariasikan pada 40 psi, 70 psi, dan 100 psi. Data dianalisis secara statistik untuk mengetahui dampak tekanan nosel terhadap penguapan air laut. Hasil penelitian menunjukkan bahwa penguapan tertinggi terjadi pada tekanan nosel 40 psi, menghasilkan 10 g air laut yang terkondensasi, sedangkan penguapan terendah diamati pada 100 psi, menghasilkan 4 g.

Kata Kunci: Air Laut; Distilasi; Evaporasi; Tekanan.

1. Introduction

Indonesia, the largest archipelagic country in the world, boasts an extensive maritime territory [1]. With 17,499 islands, 81,000 km of coastline, and 2.7 million km of inland waters, Indonesia also possesses an exclusive economic zone extending 3.1 million km, culminating in a total sea area of 5.8 million km [2]. Despite its vast maritime resources, many communities in Indonesia

still face challenges in accessing clean water. This scarcity, particularly pronounced on small islands, constitutes a significant social issue [3], as clean water is an essential need for the population [4]. Consequently, the provision of clean water remains a substantial challenge [5], especially in regions experiencing a clean water crisis, notably coastal areas. Coastal communities are especially vulnerable to saline water contamination [6], which they often use for daily activities such as bathing, washing, and consumption.

Prolonged use of saline water can lead to health issues such as hypertension, kidney disease, and skin ailments[7,8], given that clean water standards require a salt content of no more than 0.5% [9]. Seawater distillation presents a viable solution to the clean water scarcity problem [10], recognized as a promising technology[11,12].

Distillation involves heating seawater to produce steam, which is then condensed into fresh water [13]. In this study, seawater evaporation occurs within an evaporator tube [14]. Seawater is injected into the evaporator tube through a nozzle [15] and sprayed onto a heating plate [16]. The jet of seawater hitting the heating plate surface undergoes evaporation [17], which occurs when the vapor pressure of water is lower than the liquid's saturation pressure [18,19]. Additionally, evaporation is driven by the vapor pressure differential between the evaporator and the condenser due to temperature differences [20]. Water vapor moves from the evaporation chamber above the water surface in the evaporator to the low-pressure condenser, resulting in a drop in vapor pressure above the water level [21,22]. The water vapor is then condensed in the cooling pipe within the evaporator tube, forming water droplets that coalesce into fresh water [23]. Various studies have examined evaporation, including the impact of nozzle pressure on the evaporation process[24-29] and the influence of sunlight on evaporation [30-35]. However, no research has specifically investigated the effect of nozzle pressure on seawater evaporation with pressure variations of 40 psi, 70 psi, and 100 psi. Therefore, this study aims to analyze how nozzle pressure affects seawater evaporation and determine the effectiveness of each pressure variation in producing fresh water.

2. Methods

This research is field-based and conducted experimentally using a seawater distillation prototype [36]. The study employs a fogging nozzle with a diameter of 0.3 mm, which is injected into the evaporator tube. Seawater is distributed into the evaporator tube through the nozzle by spraying, and the nozzle pressure is controlled by an adjustable pump. Nozzle pressures of 40 psi, 70 psi, and 100 psi were used to compare the efficiency of the distillation process at different pressures. It is necessary to vary the pressures to observe their influence on the seawater evaporation process. Higher nozzle pressure can increase the rate of evaporation; however, selecting the correct nozzle pressure is crucial to avoid imperfect evaporation rates [37]. Therefore, varying nozzle pressures can help identify optimal conditions to maximize the rate of seawater evaporation. The following tools were used to measure the data in this study:

Table 1. Research Tools

No	Tool	Specifications
1	Thermostat XH-W3001	Range -50 °C - 110 °C, ± 0.1 °C
2	Heater Plate	Range -50 °C - 300 °C
3	Thermometer digital	Range -50 °C - 110 °C, ± 0.1 °C
4	Pompa Sprayer	DC 12V 100 psi
5	40 kg digital scale	Maximum load 40 kg, resolution 0.005 kg
6	5 kg digital scale	Maximum load 5 kg, resolution 1 gr

This study utilized the tools illustrated in Figure 2 and the schematic diagram shown in Figure 1. The research was conducted at the Mechanical Engineering Laboratory, Faculty of Industrial and Information Technology, Universitas Muhammadiyah Prof. Dr. HAMKA.

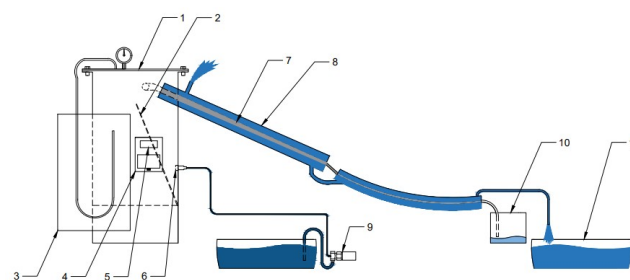


Figure 1. Distillation Device Scheme 1. Evaporator tube, 2. Heating plate, 3. PG, 4. Thermostat, 5. SH1, 6. Nozzle, 7. Steam flow, 8. Cooling water flow, 9. Pumps, 10. Freshwater reservoirs, 11. Reservoir of the outgoing cooling water flow.



Figure 2. Distillation Tools

Figure 2 illustrates the distillation device used in this study. Seawater, amounting to 5000 g, is distributed from the main container into the evaporator tube through a nozzle with a diameter of 0.3 mm, utilizing nozzle pressure variations of 40 psi, 70 psi, and 100 psi. The seawater is sprayed onto a heating plate maintained at 100°C by a thermostat. Data collection was conducted over 30 minutes at 5-minute intervals, recording the mass of water on a digital scale to determine the amount of seawater evaporated.

In Figure 2 data retrieval is performed on PG, SH1, AT1, SAL1 and SAL2.

Information:

PG : Evaporation pressure in the tube

SH 1 : Heating plate temperature

AT1 : Fresh water produced

SAL1 : Remaining seawater

SAL2 : Non-evaporating seawater

Based on the results of the trial, the equipment used worked well including checking for leaks in the joints.

3. Results and Discussion

To make it easier to do the analysis, the test result data is presented in the form of tables and graphs. In this study using nozzle pressure variations, namely 40 psi, 70 psi, and 100 psi. Sampling is carried out for 30 minutes with a time interval of 5 minutes. The data obtained include the evaporation pressure on the tube, the temperature of the heating plate, the container for holding fresh water, the remaining seawater, and the remaining seawater that does not evaporate in the tube. The following are the results of data collection which can be seen in the form of tables and graphs below.

Table 2. Nozzle Pressure 40 psi

t (Minut es)	p (Pa)	T (°C)	m (g)	SAL1 (g)	SAL2 (g)
5	20	81.2	5	4900	855
10	120	82	6	4760	
15	220	82.2	7	4625	
20	270	82.7	8	4480	
25	300	84.6	8	4370	
30	320	86.1	10	4135	

Table 3. Nozzle Pressure 70 psi

t (Minut es)	p (Pa)	T (°C)	m (g)	SAL1 (g)	SAL2 (g)
5	60	57.3	1	4750	1050
10	130	58.1	2	4525	
15	140	59.4	4	4470	
20	155	60	6	4300	
25	170	60.9	8	4150	
30	185	60.9	9	3935	

Table 4. Nozzle Pressure 100 psi

t (Minut es)	p (Pa)	T (°C)	m (g)	SAL1 (g)	SAL2 (g)
5	10	55.4	1	4650	1345
10	20	57.1	2	4360	
15	40	57.9	3	3980	
20	70	58.2	3	3670	
25	110	58.2	4	3370	
30	160	59.7	4	3100	

Based on the results of the study, data were obtained on the evaporation of seawater at various variations in nozzle pressure for 30 minutes. The results showed that the higher the nozzle spray pressure against the heating plate, the lower the evaporation pressure in the tube. This is because if the nozzle spray against the heating plate is higher, the seawater droplets that are in direct contact with the heating plate become few and only condense so that the seawater does not experience evaporation. At a burst pressure of 40 psi, the highest evaporation pressure in the tube is 320 Pa, resulting in 10 g of fresh water. At a burst pressure of 70 psi, the highest evaporation pressure in the tube is 185 Pa, resulting in 9 g of fresh water. At a burst pressure of 100 psi, the highest evaporation pressure in the tube is 160 Pa, producing 4 g of fresh water.

Figure 3 shows the difference in evaporation pressure and the amount of fresh water produced at each variation in burst pressure. The factor of mass flow rate also affects the difference in evaporation pressure at each variation in burst pressure. The mass flow rate can be obtained using the following equation:

$$\dot{m} = \rho \cdot A \cdot v \quad (1)$$

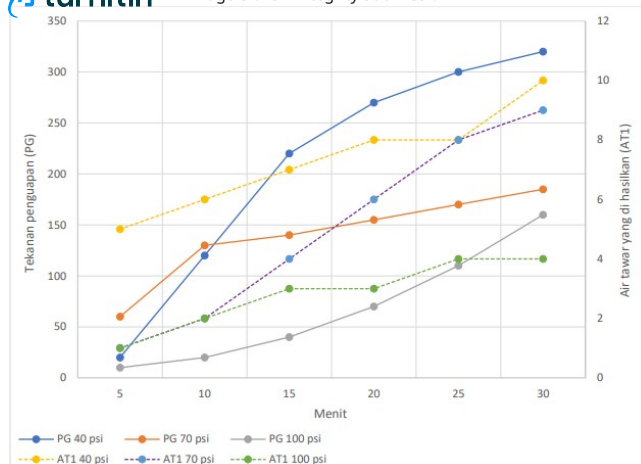


Figure 3. Evaporation Pressure Graphics (p), Nozzle Spray Pressure (Psi), Produced Fresh Water (m)

The results of the calculation show that at a burst pressure of 40 psi, a mass flow rate of 0.375 kg / s, an evaporation pressure in the tube of 320 Pa and produces fresh water as much as 10 g. At a burst pressure of 70 psi, a mass flow rate of 0.488 kg/s, an evaporation pressure in the tube of 185 Pa and produces 9 g of fresh water. At a burst pressure of 100 psi, a mass flow rate of 0.579 kg/s, an evaporation pressure in the tube of 160 Pa and produces 4 g of fresh water.

4. Conclusion

The results of testing, observation, and analysis in this study revealed a comparison of each pressure variation 40 psi, 70 psi, and 100 psi. Over a 30-minute period, the highest evaporation occurred at a nozzle pressure of 40 psi, producing 10 g of fresh water. Conversely, the lowest evaporation was observed at a nozzle pressure of 100 psi, yielding 4 g of fresh water. Based on the study's objective to analyze the effect of nozzle pressure on seawater evaporation and determine the effectiveness of each pressure variation in producing fresh water, it can be concluded that a pressure of 40 psi is the most effective for fresh water production.

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